Mechanical Properties of Graphene 1 Y.W. Sun,^{1, a)} D.G. Papageorgiou,^{1, b)} C.J. Humphreys,^{1, c)} D.J. Dunstan,^{2, d)} P. 2 Puech,^{3, e)} J.E. Proctor,^{4, f)} C. Bousige,^{5, g)} D. Machon,^{6, h)} and A. San-Miguel^{6, i)} 3 ¹⁾School of Engineering and Materials Science, Queen Mary University of London, 4 London E1 4NS, United Kingdom 5 ²⁾School of Physics and Astronomy, Queen Mary University of London, 6 London E1 4NS, United Kingdom 7 ³⁾ CEMES/CNRS UPR8011, University of Toulouse, F-31055 Toulouse, 8 France 9 ⁴⁾School of Science, Engineering and Environment, University of Salford, 10 Manchester M5 4WT, United Kingdom 11 ⁵⁾Laboratoire des Multimatériaux et Interfaces, UMR CNRS 5615, 12 Univ. Lyon, Université Claude Bernard Lyon 1, F-69622 Villeurbanne, 13 France 14 ⁶⁾ Université de Lyon, F-69000 Lyon, France and Institut Lumière Matière, 15 CNRS, UMR 5306, Université Lyon 1, F-69622 Villeurbanne, 16 France 17 (Dated: 9 April 2021) 18

The mechanical properties of graphene are reviewed with particular attention to what 19 is established and what is still uncertain. The thickness and the elastic constants are 20 clarified, and by considering also phonon frequencies it is argued that "best values" 21 come from graphite, when available. Properties not available from graphite include 22 the bending stiffness; this can be determined from studies of carbon nanotubes as 23 well as graphene. In many ways nanotubes provide access to fundamental properties 24 of graphene, not least as they are the only form of graphene that can be unsupported 25 (unstrained) in vacuum. Environmental effects are considered, both interactions with 26 substrates and with other solid and liquid media which may affect the geometrical 27 parameters defining graphene and associated elastic constant. Major uncertainties 28 persist whether slipping or sticking dominates experimental observation, both be-29 tween graphene and solid media, and between the layers of bilayer and multilayer 30 graphene. The paper concludes with a short discussion of continuum and atomistic 31 models of graphene. 32

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81 I. INTRODUCTION

Graphene has attracted enormous attention (e.g. the 2010 Nobel Prize) and research 82 effort, because of its extraordinary properties, not the least of which is its two-dimensional 83 (2D) nature. While many layered materials such as graphite and MoS_2 were already known, 84 graphene was the first material in which all the atoms are in a single plane – so for theoreti-85 cians, at least, who may ignore the electrons and consider the carbon nuclei as point masses, 86 it is a genuinely 2D material. Yet it is also a very familiar material. It is the material that 87 in stacks of millions or billions of layers, constitutes graphite, much as many sheets of paper 88 make a book. 89

There are many excellent review articles that cover the mechanical properties of graphene, 90 both experimental and theoretical, and which to be comprehensive have to have about 500 91 references.^{1–3} Why another? Our purpose is different. We aim to clarify points that are often 92 confused in the literature, and where we deem appropriate to identify problems that are as 93 yet unsolved. Some properties of graphene are just what one might expect, given what we 94 know of graphite. Here we review primarily the accuracy to which this is known. Second, 95 some properties are expected to be different, for reasons that are understood. Third, and 96 perhaps most interesting, are the anomalies. By this, we mean the behaviours of graphene 97 that are well-established experimentally, yet which lack adequate explanation according to 98 our current understanding. 99

An interesting question is "to which extent is the continuum mechanics view applicable to graphene?" Of course, this leads to the definition of a thickness for graphene, which is comparable to trying to define the thickness of an atom. The quantum nature of matter, predominant at this scale, will obviously only lead to an approximate or to a probabilistic answer.

105 II. BASIC PROPERTIES

The most basic properties determining the mechanical behaviour of a piece of a material are its physical dimensions and its elastic moduli. But how can we define or measure these mechanical properties of graphene? This apparently naive question arises from the fact that the one-atom-thick character of graphene challenges the science of the mechanics of materials, a discipline based on continuum mechanics and which has developed its successful non-atomistic view since the publication of the first book addressing the strength of materials, "Two New Sciences", written by Galileo Galilei in 1638.⁴ Of course, the discipline has subsequently integrated the atomistic nature of materials into its thinking, as in the concepts of theoretical strength and of dislocations, and in the use of molecular dynamics (MD) and density-functional (DFT) modelling.

From the material mechanics point of view, a large part of the challenge of graphene 116 mechanics arises from the ambiguities of defining the thickness of a structure – here a one-117 atom-thick surface. From this point of view, it is important to recognise that thickness is 118 not a material property, but a property of a structure, such as a plate. Moreover, even a 119 simple structure such as a corrugated-iron roofing sheet has more than one property equally 120 deserving of being called the thickness - maybe the 0.5 mm thickness of the sheet, maybe 121 the 30 mm depth of the corrugations. So the issue is not to define what the thickness of 122 graphene is, but to be clear what the context is and how the concept of thickness enters in, in 123 each context. If one wants to stack n corrugated-iron sheets at random angles, the height of 124 a stack of n sheets will be $30 \times n$ mm, while if they are aligned, the stack will be $0.5 \times n$ mm 125 high. So graphene in AB stacking as in graphite has a thickness of 3.35 Å, unambiguously. 126 We should not be surprised if this value changes for other stackings⁵ (e.g. around 3.6 Å for 127 random-angle stacking called turbostratic, and see Table I for some graphene thicknesses 128 as described by the distances from various substrates). Nevertheless, while the stacking of 129 corrugated-iron sheets may have no bearing on the physical characteristics of the individual 130 sheets, this can be a totally different issue for graphene. Depending on the geometrical 131 arrangements, modifications of the electronic structure of the individual graphene sheets 132 could be significant. 133

¹³⁴ A. Thickness of graphene

The thickness of graphene is a vexed question. The common description of graphene as a 2D material implies extension in two dimensions but not in the third (*i.e.* zero thickness). Indeed, the very definition of the thickness of graphene is complicated from a quantum mechanics point of view, as it pertains to defining the diameter of atoms. The Yakobson paradox⁶ arose through the attribution of values as low as 0.6 Å to the thickness – and hence Young's moduli as high as 5 TPa. It is interesting to see how the similar issue of the size of
atoms was addressed a century ago.

Sir William Bragg proposed the hypothesis that atoms of a given element could be con-142 sidered to be spheres with a fixed radius.⁷ However, the crystal structures of some metals 143 and compounds led to proposals that some atoms have lower symmetry than spherical.⁸ 144 Wyckoff re-examined the question and concluded that the hypothesis of constant radii (or 145 other shapes with fixed sizes) must be rejected but that the evidence supported atomic radii 146 that vary more or less according to their environment.⁹ This is the basis of the modern view. 147 Modern data compilations give for example the covalent radius of the carbon atom as 0.70 148 A. More precisely, the covalent radius of carbon is largest for single-bonded carbon with the 149 C-C bonds in ethane and diamond both at 1.54 Å, smaller for sp^2 -bonded (graphite at 1.42) 150 Å and ethene at 1.33 Å bond length) and smallest for triple-bonded carbon (acetylene, 1.20 151 Å). On the other hand, the van der Waals (vdW) radius of a carbon atom is given as 1.70 152 Å. An early measurement of the thickness of the benzene molecule gave 4.70 Å,¹⁰ while the 153 thickness of the larger pyrene molecule is 3.53 Å,¹¹ very close to the spacing of the graphene 154 sheets in graphite at 3.35 Å. 155

These considerations appear to give a clear meaning to the concept of the vdW thickness 156 of graphene, as much as of simpler molecules such as benzene and the higher polycyclic 157 aromatic compounds such as pyrene. It expresses the distance of closest approach of other 158 physisorbed atoms – whether carbon or anything else, because the repulsive interatomic 159 potential deriving from Pauli exclusion is largely independent of the nature of the interacting 160 atoms, and in the absence of a chemical bond, so is the vdW attractive potential. Following 161 Wyckoff,⁹ the thickness should be expected to vary with the environment, whether it is a 162 surrounding gas or a substrate, as can be seen in Table I – even quite considerably as the 163 vdW forces, while always weak, can vary by an order of magnitude. 164

¹⁶⁵ B. Graphene elastic stiffness tensor

Only in graphite is graphene found in a symmetrical environment (sandwiched between graphene sheets with only a vdW potential binding them) and with a known thickness. We may then define the graphene elastic stiffness constants c_{ij} in this situation as a reference system. To deal with possible variations in thickness in other environments, it makes sense

Substrate	Distance [Å]
$Ir(111)^{12}$	3.38
Graphite	3.35
$Pt(111)^{13}$	3.30
$SiC(0001)^{14}$	3.24
$\mathrm{Ru}(0001)^{15}$	2.2
$\mathrm{Co}(0001)^{16}$	2.2
Ni(111) ^{17,18}	2.1

TABLE I. Experimental graphene-substrate distance for various substrates.

to define the in-plane 2D elastic stiffness tensor $c_{ij}^{2D} = c_{ij}d_0$ with i, j = 1, 2, and d_0 is the 170 graphite interlayer spacing at ambient pressure. This tensor comes simply from the sp^2 bond 171 bending and stretching stiffnesses, and so is independent of the graphene thickness d per se. 172 To see this, consider making the graphene layers in graphite thinner, spaced more closely 173 and reducing a_{33} , as happens under pressure. Then there are more graphene layers per 174 unit volume, a_{33}^{-1} , and the 3D constants c_{11} and c_{12} are increased proportionately – leaving 175 c_{ij}^{2D} unchanged. The out-of-plane elastic constants, particularly c_{33} but also c_{13} , have to be 176 considered separately. This is done in Sec. III C. 177

A crucial aspect is then to know if the graphene sp^2 bonds – which largely determine c_{ij}^{2D} – are significantly influenced either by the environment of the graphene (what it is in contact with) or by its geometry (for example, bending) leading to a modification of the reference elastic stiffness constants. The zeroth-order approximation would be that there are no such influences. But if there are, we will need also to consider what are the limits in such modifications which can be admitted before saying that we are dealing with a new system. These issues are addressed in Sec. VI.

185 III. GRAPHENE REFERENCE MECHANICAL PROPERTIES

¹⁸⁶ A. Graphite 3D mechanical properties

Graphite consists of a macroscopic multilayer stack of graphene layers, held apart by the π -orbitals and Pauli exclusion, and held together by the vdW interaction, with a spacing of ¹⁸⁹ 3.35 Å.¹⁹ It is, consequently, highly anisotropic, while being isotropic in-plane. The elastic ¹⁹⁰ moduli reported in Table II were obtained from inelastic X-ray scattering,²⁰ but similar ¹⁹¹ values were reported from ultrasonic and static mechanical testing.²¹

	c_{ij} [GPa]	$c_{ij}^{2D}~[{\rm Nm}^{-1}]$
c_{11}	1109 ± 16	372 ± 5
C33	38.7 ± 0.7	
c_{12}	139 ± 36	47 ± 12
c_{13}	0 ± 3	
c_{44}	5 ± 0.3	
c_{66}	485 ± 10	
Y	1092 ± 18	362 ± 27
ν	0.125 ± 0.033	(no units)

TABLE II. Elastic moduli values obtained from inelastic X-ray scattering²⁰ in Voigt notation

Within experimental error, the c_{13} of graphite may be negative – and DFT calculations 192 support this.²² While DFT may not evaluate the interlayer spacing accurately without vdW 193 correction,²³ the related elastic constant c_{33} calculated at the experimental equilibrium spac-194 ing agrees well with the experimental c_{33} . That improves confidence in the reliability of the 195 negative calculated value of c_{13} .²² If c_{13} is truly negative, it might be due to ripples in 196 the measured graphite sample, like what happens with auxetics, prominent structures that 197 have negative Poisson's ratio.²⁴ This could be further linked to the negative thermal ex-198 pansion of graphene, which will be discussed later. In a general 3D anisotropic medium, 199 the Young's moduli for a stress σ_{11} are given by more complicated expressions than the 200 usual $Y = c_{11} - 2c_{12}^2/(c_{11} + c_{12})$. The small or vanishing value of c_{13} simplifies this to 201 $Y = c_{11} - c_{12}^2/c_{11}$ for the in-plane Young's modulus of graphene. Similarly the in-plane 202 Poisson's ratio simplifies to $\nu = c_{12}/c_{11}$. 203

²⁰⁴ B. In-plane graphene mechanical properties

Under the hypothesis that graphene can be treated as a continuous elastic medium, it is thus expected to have the 3D values given in Table II and hence the c_{ij}^{2D} values also given there. Indeed, the in-plane Young's modulus $Y^{2D} = 362 \text{ Nm}^{-1}$ is consistent with the $Y^{2D} = 340 \pm 50 \text{ Nm}^{-1}$ of monolayer freestanding graphene measured by atomic force microscopy (AFM)²⁵, which will be discussed in Sec. IV A.

For graphene in vacuum, the greatest difference from graphite is likely to be an increase 210 in the thickness, as the π -orbitals are no longer compressed by the vdW attractive force – 211 indeed, as discussed above, the thickness becomes defined by whatever convention is used 212 to specify where the π -orbitals end. The 3D elastic constants will vary inversely with the 213 thickness. The 2D elastic constants are unaffected. Neglecting any small effect on the sp^2 214 bond strength caused by the relaxation of the π -orbitals, and any small contribution of 215 the π -bonds to c_{11} and c_{12} , the 2D elastic constants will be unchanged; they are mostly 216 determined by the carbon sp²-network.²⁶ 217

Direct experimental determinations of the in-plane elastic constants of graphene are much less accurate than those of graphite, but are generally consistent with them within experimental uncertainty (see Sec. IV). The results of *ab initio* theoretical calculations also show very similar in-plane elastic constants of graphene to graphite, with the caveat that it is much harder to assess the uncertainty of these calculations than to assess experimental uncertainties.²²

Given the large uncertainties on experimental determinations of c_{ij}^{2D} in graphene, the 224 best information about possible perturbations of the in-plane sp^2 bonding comes from a 225 comparison of the phonon frequencies, and in particular the zone-centre E_{2q} optical phonon. 226 It is a Raman active mode, named the G-mode (G for graphite), at about 1600 $\rm cm^{-1}$ in all 227 sp^2 -bonded forms of carbon.²⁷ In graphite, the frequency of the G-mode is at 1575 - 1580228 cm^{-1} ,^{27–30} and it is reported at 1581 – 1592 cm^{-1} in graphene.^{31–34} This difference of about 229 0.5% puts an upper limit of about 1% on any change in the stiffness of the sp² bonds or c_{ij}^{2D} 230 in going from graphite to graphene. 231

²³² Chemical perturbations such as substitutional doping and doping by charge transfer from ²³³ a substrate or nitric acid, etc, mechanical perturbations such as bending, stretching or high ²³⁴ pressure, or structural perturbations such as stacking order, again scarcely perturb the GM ²³⁵ frequency relative to graphite, a few tens of cm⁻¹ at most. The sp² covalent bonding is ²³⁶ clearly robust. This is actually true of small molecules generally, which like graphene are ²³⁷ "all surface", yet whose vibrational frequencies are little affected from the vapour phase, ²³⁸ through solvation or liquefaction, to crystalline solid forms. It is not surprising, then, that this is so for graphene. Of course, the concept of a material that is all surface is not novel; chemistry deals largely with such things. Some readers might be alerted here that shift of tens of cm⁻¹ in the G-mode could be considered to be huge in specific studies, but when considering the corresponding change in the in-plane stiffness, such shifts are still very little compared to the large experimental uncertainties from the direct measurements on graphene by AFM (Sec. IV A).

²⁴⁵ We conclude that graphene and graphite are the same regarding in-plane stiffness within ²⁴⁶ experimental uncertainty, from the G-mode frequency. More details about the G-mode ²⁴⁷ frequency, such as the contributions from the deformed π -orbitals and up to the fifth nearest ²⁴⁸ neighbour C atoms, will be discussed in Sec. VIC and VIII, respectively.

²⁴⁹ C. Out-of-plane stiffness of graphene

On the assumption that graphene has the vdW thickness equal to the interlayer spacing 250 of graphite (Sec. II A), we may suppose that it has a stiffness in that direction, equal to that 251 of graphite, 38.7 ± 0.7 GPa.²⁰ This basic assumption is challenged by various objections. 252 For example, in graphite, c_{33} probes the interlayer potentials defined through the π -orbital 253 interaction in an AB stacking. What sense does this make when considering graphene in 254 vacuum or in a solid or liquid molecular environment? Deformation of solids is usually 255 described in terms of the changes in the distances between atoms, measured from nuclei to 256 nuclei, or in crystals, by the spacing of planes of atomic nuclei. It could be argued that with 257 only one plane of atoms, one plane of nuclei, out-of-plane deformation or strain of graphene 258 is meaningless, and with that, the c_{33} of graphene is a meaningless concept. 259

Against these challenges, thickness and stiffness certainly exist in monolayer graphene, 260 just as atomic radii exist (Sec. II A), and vary with pressure, which defines an atomic ra-261 dial stiffness. Electronic orbitals extend out of the plane of graphene nuclei and they resist 262 compression. Such resistance (more accurately, compliance) consists of two parts. One is 263 the Pauli exclusion compliance from the overlap of the π -orbitals of graphene and the outer 264 orbitals of the medium that applies compression. It naturally depends on what that medium 265 is and how it is stacked onto graphene. Moreover, the compliance has to be apportioned 266 between the compliance of the graphene and the compliance of the medium. Surprisingly, 267 even if the compressing medium is another graphene monolayer AB-stacked to it, the out-268

of-plane stiffness is calculated to be, not that of graphite, but only about half of it.³⁵ This softness is attributed to the squeezing of π -orbitals through the graphene plane in a bilayer system, whereas such squeezing-through is prohibited in graphite (infinite number of layers) by symmetry.³⁵ In addition to the compliance of the Pauli exclusion for undeformed π -orbitals, the other contribution from the graphene to the total compliance is from the deformation of the π -orbitals of the graphene. This could be estimated by calculating the energy difference between relaxed and deformed π -electron distributions.

In the absence of a conventionally-defined elastic constant c_{33} based on internuclear dis-276 tances, one approach to define the out-of-plane stiffness of graphene is to use a related 277 quantity that is itself unambiguously defined and measurable. The in-plane bonds stiffen 278 under compressive in-plane strain, which can be expressed as a 2D strain and converted to 279 a 2D stress by c_{ii}^{2D} . That has been measured by the increase in G-mode phonon frequency 280 under pressure.^{31,34,36–38} In graphene as in graphite, the 2D in-plane stress can be applied by 281 hydrostatic pressure, and the 2D stress is then directly proportional to the thickness. Since 282 graphite and graphene are very soft out-of-plane, under hydrostatic pressure the thickness 283 decreases significantly (the π -orbitals being considerably compressed). That gives a large 284 reduction of in-plane force below the linear proportionality with pressure, and therefore a 285 substantially sublinear shift of the G-mode frequency with pressure.³⁹ An experiment adopt-286 ing this approach had large experimental uncertainties, but within experimental uncertainty 287 first confirmed from the shift-rate of the G-mode that the thickness of graphene is not sig-288 nificantly different from its thickness in graphite. Then the sublinearity of the shift-rate 289 could not distinguish the graphene $c_{33} = 0 \pm 300 \text{ GPa}^{40}$ from graphite $(38.7 \pm 0.7 \text{ GPa})^{20}$. 290 Of course, there are also possible effects of the pressure medium on the graphene response; 291 these are discussed further in Sec. VI below. 292

²⁹³ D. Properties not related to graphite

In contrast to the foregoing, there are some properties of graphene that are quite distinct from, or unrelated to, any properties of graphite. It is probably accurate to say that these are all properties related to the freedom graphene has to displace in the z-direction, out of plane, in ways that are unavailable to the layers in graphite. Briefly, these include the theoretical instability of a 2D sheet, the negative thermal expansion in-plane of graphene, ²⁹⁹ and the stiffness in bending of monolayer graphene and of multilayer graphene.

Acoustic phonons in graphene; their effect on the thermal expansion and stability of graphene samples

In this section, and the following sections III D 2 to III D 4, we will review the nature 302 of the acoustic phonon modes in graphene. They are responsible for some key properties 303 of graphene: The observation of a negative thermal expansion coefficient under certain 304 conditions, lack of mechanical stability and - in consequence of this - static ripples in 305 the graphene monolayer. These properties reflect, in a fundamental way, the 2D nature 306 of graphene: The lack of restoring forces from adjacent atomic layers in the out-of-plane 307 direction, and the density of states for a 2D material varying in proportion to k rather than 308 k^2 . 309

To begin, we must briefly review the low-energy part of the phonon dispersion relation 310 of graphene. The dispersion relation has been studied for decades prior to the discovery of 311 graphene, as a simplified model for the phonons in graphite.^{41–43} Initially these calculations 312 were performed using traditional semi-classical "ball and spring" force constant models, and 313 the results obtained nowadays using DFT calculations²² are in reasonably good agreement 314 with those obtained previously using "pencil-and-paper" methods. Agreement is also good 315 with the experimental data obtained on graphite using electron energy loss spectroscopy.⁴⁴ 316 inelastic neutron⁴² and X-ray scattering.²⁶ Fig. 1 shows the low-energy part of the dispersion 317 relation, calculated in Ref. 22 using DFT. In this section we shall regard low energy as up to 318 50 meV since we are interested in the phonons which can be excited in significant number 319 by thermal energy at $T \leq 300$ K. (At 300 K, $k_B T \approx 25$ meV). 320

322 2. Mechanical stability of graphene

As shown in Fig. 1, there are three acoustic phonon modes which can be excited at 300 K. They are longitudinal (LA) and transverse (TA) polarization in-plane modes (both with linear dispersion relations in the low k limit) and a transverse polarization out-of-plane mode (ZA) with a quadratic dispersion relation in the low k limit ($E \propto k^2$).^{22,43} The softness of this mode relative to the LA and TA modes is believed to be partially due to the lack of



FIG. 1. Low-energy graphene phonon modes along the Γ -M direction in the first Brillouin zone, obtained in Ref. 22 using DFT. LA and TA modes are the longitudinal and transverse polarization in-plane modes and the ZA mode is the transverse polarization out-of-plane mode.

bending stiffness of graphene. However, the low bending stiffness is not necessary to explain 328 the quadratic dispersion relation. A transverse wave on a string (or sheet) with zero tension 329 but a bending stiffness D has $\omega(k) = k^2 (D/\rho)^{1/2}$, so the quadratic dispersion relation is fully 330 consistent with the recently measured 1.7 eV bending stiffness of graphene.^{39,45} The number 331 of phonons present at temperature T from branch i of the dispersion relation (hence with 332 energy $E_i(k)$ is obtained by multiplying the Bose-Einstein distribution by the density of 333 states D(k) and integrating over all available k (Eq. 1). The lower limit k_{min} corresponds to 334 the longest wavelength mode that can exist on the graphene sheet. So $k_{min} \approx 2\pi/L$ where L 335 is the diameter of the graphene sheet. For macroscopic graphene samples therefore $k_{min} \rightarrow 0$ 336 is a reasonable approximation. Referring to Fig. 1 and the known graphene reciprocal lattice, 337 the path $\Gamma \to M$ in reciprocal space covers a range of $2.3 \times 10^9 \text{ m}^{-1}$, whilst $k_{min} \approx 10^5 \text{ m}^{-1}$ 338 for a 10 μ m diameter graphene sheet. 339

The upper limit k_{max} should be set so as to integrate over all possible phonon modes, in reality the upper limit of the integral is set by the Bose-Einstein distribution term dropping to zero upon increasing k, E.

$$N = \int_{k_{min}}^{k_{max}} \frac{1}{e^{\beta E(k)} - 1} D(k) dk$$
 (1)

Here, we write $\beta = 1/k_B T$ for convenience. The densities of states in 1D, 2D and 3D systems, per unit area/volume, are given by:

$$D_{1D} = \frac{1}{\pi}$$

$$D_{2D} = \frac{k}{2\pi}$$

$$D_{3D} = \frac{k^2}{2\pi^2}$$
(2)

It has been shown that the value of the integral in Eq. 1 diverges to $+\infty$ when a 2D 345 density of states function is utilized, combined with a quadratic dispersion relation.⁴⁶ This 346 is due to the value of the integrand diverging in the low-k limit. The experimental obser-347 vations that genuinely free-standing graphene sheets do not exist, and that all graphene 348 sheets are covered in static ripples (so, in at least one respect, not genuinely 2D) are at-349 tributed to this divergence.^{46,47} In addition, this divergence prevents the categorization of 350 free-standing graphene sheets as a metastable form of carbon in an equivalent manner to 351 diamond, nanotubes and fullerenes at ambient conditions. 352

In this section, we are going to explore the cause of this divergence in a little more 353 detail – can it be observed in systems with other dimensionalities, and with other dispersion 354 relations? In Table III we give the integrands that would be utilized in Eq. 1 for phonon 355 modes with quadratic and linear dispersion relations for 1D, 2D and 3D systems. Since we 356 seek only to evaluate which of these integrands diverge we have assumed a finite non-zero 357 temperature and omitted all constant terms, where we assume β is a finite constant since 358 T > 0 K. The integrands are evaluated in the low-k limit (equivalent to assuming that 359 the sample is large enough to support phonons across essentially the entire Brillouin zone) 360 using first order expansions of the exponential term ($e^x \approx 1 + x$), appropriate since we are 361 considering exclusively acoustic phonons for which $E \rightarrow 0$ also in the low-k limit. 363

Summarising the findings from Table III, we observe that in a 3D system divergence is not observed for a linear or quadratic dispersion relation, for a 2D system divergence is observed only with a quadratic dispersion relation, and for a 1D system divergence is observed with either a linear or quadratic dispersion relation. These differences are due to the density of states function being different in each case.

TABLE III. Integrands $I_k^{(n)}$ to evaluate the number of phonons present at finite temperature for different dispersion relations, in systems with different dimensionality n. Constant terms and temperature dependence have been omitted.

	1D	2D	3D
$E \propto k$	$I_k^{(1)} = \frac{1}{e^k - 1}$	$I_k^{(2)} = \frac{k}{e^k - 1}$	$I_k^{(3)} = \frac{k^2}{e^k - 1}$
	$\lim_{k \to 0} I_k^{(1)} = \frac{1}{k}$	$\lim_{k\to 0} I_k^{(2)} = 1$	$\lim_{k \to 0} I_k^{(3)} = k$
$E\propto k^2$	$I_{k^2}^{(1)} = \frac{1}{e^{k^2} - 1}$	$I_{k^2}^{(2)} = \frac{k}{e^{k^2} - 1}$	$I_{k^2}^{(3)} = \frac{k^2}{e^{k^2} - 1}$
	$\lim_{k \to 0} I_{k^2}^{(1)} = \frac{1}{k^2}$	$\lim_{k \to 0} I_{k^2}^{(2)} = \frac{1}{k}$	$\lim_{k \to 0} I_{k^2}^{(3)} = 1$

To our knowledge, the consequences of this divergence have not been studied experimen-369 tally in 1D systems. Certainly its observation would require the existence of an extremely 370 long system to ensure $k_{min} \rightarrow 0$, and for the system to be free-standing to allow these 371 phonons to propagate. The nearest humankind has got to a genuinely 1D system is carbyne 372 - in recent years carbyne chains up to 600 nm in length have been synthesized, satisfying 373 the first condition.⁴⁸ But the requirement that they are enclosed inside a carbon nanotube 374 probably prevents the second condition from being satisfied. Single-walled carbon nanotubes 375 (SWCNTs) themselves are not strictly one-dimensional as far as the density of states func-376 tion is concerned. The quantization condition for the direction along the tube axis is the 377 requirement for a standing wave with allowed wavelengths determined by the tube length, 378 whilst the quantization condition for the direction tangential to the tube axis is for travel-379 ling waves with allowed wavelengths determined by the tube circumference. Thus, whilst 380 the allowed quantum states are far more widely spaced in the tangential direction, the elec-381 tron and phonon wavevectors do still have 2 degrees of freedom, and the density of states 382 function should follow a 2D form rather than the 1D form. 383

As far as graphene is concerned, the divergence in the number of out-of-plane phonons at finite temperature is believed to be responsible for the following experimental observations. Firstly, real monolayer graphene samples can only exist when provided with some mechanical support. Usually this is provided by a substrate. Samples may be described in the literature as "freestanding" when there is an aperture in the substrate or some similar arrangement, but there is always some mechanical support. Monolayer graphene samples may be suspended in a liquid such as water, in which the viscosity of the liquid provides adequate support.^{47,49} Secondly, real monolayer graphene samples are always rippled. These ripples are expected from the findings of atomistic simulations.⁵⁰ They have been observed using electron diffraction⁵¹ and also scanning tunnelling microscopy,⁵² which has confirmed that they are static ripples, with $\lambda \approx 5$ nm. The ripples become weaker for progressively thicker graphene samples as the thickness of the graphene sample itself provides the required rigidity. These ripples are a completely separate effect to the Brownian motion observed in graphene.^{53,54}

The role of ripples in ensuring stability can be understood in terms of the restoring forces. The softness of the ZA mode shown in Fig. 1, compared to the in-plane modes, is because of the lack of restoring forces due to bond-stretching in the low amplitude limit, and of those due to bending in the large-wavelength limit. The curvature induced by the ripples ensures that there is some restoring force due to bond-stretching even in the low amplitude limit, making the mode — partially – analogous to the radial breathing mode in SWCNTs.⁵⁵

403 3. Thermal expansion coefficient of graphene

The existence of thermal expansion is perhaps the most intuitive example of the interplay 404 between static and dynamic material properties. The lattice constant of a solid is considered 405 the archetypal static property, yet at T > 0 K it is altered (usually increased) by the presence 406 of phonons; the archetypal dynamic property. The observation that the vast majority of 407 materials expand upon heating is a consequence of the nature of the function V(r) giving 408 the potential energy between two of the atoms comprising the solid as a function of their 409 separation r. The Lennard-Jones potential is frequently utilized as a good approximation for 410 covalently bonded solids even though it is only strictly correct for solids where the cohesion 411 is due solely to vdW forces. In any case, the potential will always have three key features in 412 common with the Lennard-Jones potential: (1) It will be attractive for moderate values of 413 r, with a minimum at $r = r_0$, the inter-atomic separation in the absence of phonon effects. 414 (2) In the limit $r \to \infty$, $V \to 0$. (3) In the limit $r \to 0$, $V \to +\infty$ to prevent atomic overlap. 415 As a result of these features, V(r) is not symmetric about $r = r_0$ and this asymmetry will, 416 in the absence of other effects, favour thermal expansion rather than contraction. 417

This argument applies directly to any reasonably isotropic and dense 3D solid and, for that matter, a 2D solid existing in a 2D world (in which case out-of-plane phonon modes would not exist). However, graphene's position is as that of a 2D solid in a 3D world. In this case, the excitation of an out-of-plane vibration does not cause any thermal expansion
in the out-of-plane direction. However it can cause contraction in the in-plane direction as
atoms are pulled inwards by the out-of-plane movement.

Thus, for graphene to exhibit a negative coefficient of thermal expansion coefficient 424 (CTE), all that is necessary is for the contribution from the out-of-plane phonons to dom-425 inate over that from the in-plane phonons. We can see how this can be the case at low 426 temperature from Fig. 1. The quantum states available for all phonons are equally spaced 427 in k-space so, when phonons of all kinds (LA, TA, ZA) can be excited up to a certain energy, 428 the out-of-plane ZA phonons dominate as they cover a wider area of k-space. The qualitative 429 arguments proposed here are borne out by the findings of detailed theoretical calculations;²² 430 Graphene should indeed exhibit a negative CTE. 431

However, as shown earlier a graphene sheet which is genuinely freestanding, and therefore 432 free to expand and contract, cannot exist. Experimental measurements of the CTE of 433 graphene are therefore indirect and prone to large experimental, theoretical and conceptual 434 uncertainties.⁵⁶ Most commonly, it has been measured by varying temperature whilst the 435 graphene is adhered to a substrate. The graphene is assumed to adhere perfectly to the 436 substrate due to its extremely high surface area to volume ratio so, upon temperature 437 increase the graphene would be forced to expand rather than contract and is therefore 438 under significant tensile strain. The extent of this strain is usually calculated using Raman 439 measurements on the G peak,⁵⁷ though grazing incidence X-ray diffraction has also been 440 utilized.⁵⁸ Potential sources of uncertainty in such experiments include, but are not limited 441 to: 442

• The logic of such experiments is often not clearly presented. When a thin film fully 443 adheres to a substrate, the difference in thermal expansions causes a stress in the film, 444 not a strain. Stress cannot be measured by Raman. The CTE of the film should be 445 directly measured by Raman of the thermal strain in a *free-standing* specimen of the 446 film. Then to correct for the temperature effect on the phonon frequency, the phonon 447 shift at known strain and ambient temperature is compared with the data from the 448 film adhering to a substrate, at a known strain and elevated temperature. Given 449 the difficulty of studying free-standing graphene, data from theoretical simulations is 450 commonly used instead. 451

• Commonly used substrates for graphene consist of layers of different materials with different TEC. It is not always clear which layer dominates, and to what extent the layers remain bonded.

• The out-of-plane phonons causing the negative TEC should be suppressed to some extent by the presence of the substrate. It is thus not clear how applicable findings regarding graphene on a substrate are to free-standing graphene.

• We assume a value for the Grüneisen parameter (for the phonon responsible for the G peak) when the strain is calculated from the G peak Raman measurement.

• Graphene does not always adhere perfectly to the substrate.

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Notwithstanding these problems, the experimental evidence is consistent with graphene 461 having a negative TEC at temperatures up to at least 500 K.⁵⁶ (and references therein) This is 462 also the case for the in-plane measurements on graphite. Indeed, comparing reported values 463 and uncertainties of the TEC for graphene and for graphite in-plane,^{22,59-61} it is difficult to 464 find any justification for considering them to be different. The small or negative CTE is 465 reflected in the contrasting characteristics of graphene grown by chemical vapour deposition 466 (CVD) and epitaxial growth (EG). Both growth processes take place at high temperature, 467 followed by the substrate contracting upon cooling. In the case of the CVD graphene, Raman 468 measurements at ambient conditions indicate that the graphene has "relaxed" *i.e.* slipped 469 over the substrate upon cooling to stay in equilibrium. However, Raman measurements 470 made on the EG graphene at ambient conditions indicate that it has remained adhered to the 471 substrate upon cooling after growth.⁶² As a consequence it is under significant compressive 472 strain; equivalent to several GPa pressure.⁵⁶ 473

As briefly mentioned above, 3D materials with a layered structure can also exhibit a negative thermal expansion coefficient along one axis or in one plane,^{63–65} although the phonon modes that cause this will cause positive thermal expansion along some other axes. In graphite at 300 K for example, the in-plane thermal expansion coefficient is $ca. -1.4 \times 10^{-6}$ K^{-1} , but the out-of-plane thermal expansion coefficient is $ca. +25 \times 10^{-6}$ K⁻¹.²²

479 4. Grüneisen parameters and elastic bands

The negative thermal expansion coefficient resulting from the ZA mode — by definition — results in this mode having a negative Grüneisen parameter (Grüneisen parameters for all phonon modes in graphene are calculated throughout the first Brillouin zone in Ref. 22). In the Grüneisen approximation, the pressure and temperature dependence of the phonon mode can both be incorporated into the volume or lattice parameter dependence on pressure and temperature, linked by the Grüneisen parameter. For a 2D material such as graphene we can write, following the approach in Ref. 34:

$$\frac{\omega(P,T)}{\omega_0} = \left(\frac{\alpha(P,T)}{\alpha_0}\right)^{-2\gamma_{ZA}} \tag{3}$$

where $\alpha(P,T)$ is the pressure and temperature-dependent lattice parameter, as projected into the basal plane of the graphene lattice. In this case if $\gamma_{ZA} < 0$, $\omega(P,T)$ will decrease under isothermal compressive strain and increase under isothermal tensile strain. Whilst highly unusual on a microscopic level, it is the behaviour we are used to observing in common macroscopic 1D systems in everyday life: When you stretch elastic bands, or guitar strings, or the shrouds and stays of a yacht, they twang at a higher frequency!

493 5. Bending stiffness

In 3D systems, while the elastic constants discussed above are properties of a material, a bending stiffness is a property, not of a material, but of a structure, i.e., related to geometry. It is however defined in a similar way. The elastic constants are the second derivatives of the potential energy with respect to deformation (strain). Graphene has the additional degree of freedom, of bending to a curvature κ (= 1/R where R is the radius of curvature). This additional degree of freedom, bending, and hence a bending stiffness, has no analogue in graphite. To account for the bending stiffness, D, defined by the energy of curvature κ ,

$$U = \frac{1}{2}D\kappa^2 \tag{4}$$

there is a large literature in which an effective Young's modulus Y_{eff} and an effective thickness h_{eff} are introduced such that both the in-plane elastic moduli and the bending stiffness 503 can be expressed:

$$Yh = Y_{eff}h_{eff} D = \frac{Y_{eff}h_{eff}^{3}}{12} \text{ or } \frac{Y_{eff}}{1 - \nu^{2}} \frac{h_{eff}^{3}}{12}$$
(5)

where in the second expression for D it is the plane-strain modulus that is used, as is correct for a plate made of an isotropic material.

The model of Eq. 5 has had remarkable success in capturing the behaviour of graphene 506 and nanotubes, particularly in contexts where beam, plate and shell theory are used to un-507 derstand buckling behaviour under load.⁶ Due to the uncertainties of a definition of graphene 508 thickness, on the other hand, it has led to claims that nanotubes have "an extremely large 500 Young's modulus",⁶⁶ for example the value of 5.5 TPa given by Yakobson *et al.* for Y_{eff} .⁶ 510 Similarly, astonishment has been expressed at the small values of h_{eff} , as low as 0.066 nm, 511 "ultrathin compared with the C-C bond length 0.142 nm".⁶⁶ The very wide range of values 512 reported for these parameters⁶⁷ has been described as a paradox (Yakobson's paradox).^{68,69} 513 As some authors have recognised, there is no paradox,^{70–72} but much of the literature fails 514 to distinguish h and Y from h_{eff} and Y_{eff} . The wide ranges of values express only the im-515 precision in the determination of D by different methods. The unphysical values found for 516 Y_{eff} and h_{eff} simply reflected the unphysical nature of these parameters, which correspond 517 to nothing in the real world but are simply convenient ways of representing D in structural 518 engineering computational packages that do not permit D to be entered independently of Y519 and h (if any such packages exist, which is doubtful, as engineers routinely analyse structures 520 of this sort, such as honeycomb-filled or rib-reinforced plates). 521

Reported experimental methods of measuring D range from the collapse pressure P_C of 522 nanotubes under high pressure $(1.7 \pm 0.2 \text{ eV})$,⁴⁵ to the taper angle of a strip torn from an ad-523 hesive substrate $(2.1 \pm 0.1 \text{ eV})$.⁷³ A value of 2 eV was estimated from the phonon dispersion 524 measured by high resolution electron energy loss spectroscopy (HREELS), on Pt (111) sup-525 ported graphene.⁷⁴ Other measurements reported values as high as $10^3 - 10^4$ eV for rippled 526 monolayer⁷⁵ and $35.5^{+20.0}_{-15.0}$ eV for bilayer.⁷⁶ Torres-Dias *et al.* found that for nanotubes of 527 small diameters, the normalised collapse pressure $P_C R^3 = 3D$ dropped substantially below 528 the theoretical value, which could be due to the softened bending potentials at large bending 529 angles, or an effect of atomicity.⁴⁵ Carter *et al.*⁷⁷ study the Euler buckling load for a straight 530 pillar. When the compliance of a continuous pillar is concentrated at a few points (atoms, 531

or angular springs) between rigid portions, the buckling load is substantially reduced. The effect is hard to explain, but is readily derived from the TA phonon dispersion curve of the infinite linear chain, which predicts a decrease of the phase velocity of the TA mode as the wavelength decreases. In any case, this observation requires that the bending stiffness of flat graphene is obtained, as in Ref. 45, from the extrapolation of normalised experimental collapse pressures to $R^{-3} = 0.^{45,77}$

538 6. Folding

When a sheet of graphene is folded over onto itself, it adheres due to the vdW interaction. 539 The radius of the fold is determined by the strength of the vdW attraction and by the value of 540 D, and is of the order of the radius of C_{60} . An example is found in large-diameter SWCNTs, 541 which are collapsed already at ambient pressure into the shape called "dogbone" or "peanut". 542 This presents a cavity of ≈ 0.6 nm in diameter at the edge.^{78,79} This small diameter reflects 543 the very small value of the bending stiffness of monolayer graphene, in the range of 1 eV, 544 but an accurate evaluation depends on knowing the strength of the vdW adhesion holding 545 the fold folded. The atomic nature of the sheet plays a role, and folding is angle dependent 546 as reported in the study of Zhang $et al.^{80}$ They reported that graphene sheet tends to fold 547 along armchair (0° folding angle) and zigzag directions (30° folding angle). Consequently 548 the spontaneous collapse of large nanotubes must depend on the chirality, which imposes 549 the folding angle. The most unfavorable direction has the folding angle of 10° . Under very 550 high temperature annealing (2000 °C) the unfavourable angle of folded graphene generates 551 defects such as pentagons, relaxing the stress and leading to an irregular edge composed of 552 zigzag and armchair directions.⁸¹ 553

For bilayer and multilayer graphene with n layers (n > 1), the angle dependance disappears and this cavity size is governed by the number of layers, n. The length of the folding (shape of a pear) is well described by:⁸²

$$L(n) = \pi \sqrt{\frac{D}{\gamma}} = \left[\pi \sqrt{\frac{Y}{\gamma} \frac{d_{vdW}^3}{12}}\right] n^{\frac{3}{2}}$$
(6)

where D is the bending stiffness of the multilayer per unit length, d_{vdW} is the equilibrium distance between two graphene layers, Y is the Young's modulus, γ surface adhesion energy per unit length around 260 mJ·m⁻² (42.2 meV/atom).⁸² In Chen *et al*,⁸³ the bending stiffness

 $D_n(eV)$ is well fitted by $6.7 \times (nh)^2$ with n between 2 to 6. These results are completely 560 different even if for both relations, the bending stiffness of multilayer graphene is higher 561 than a simple summation of the bending stiffness of each individual layer. The mechanical 562 properties of bilayer and multilayer graphene depend critically on the issue of slipping or 563 binding between layers. If layers slip freely, then the bending stiffness of n-layer graphene 564 will be nD_1 which is not the case from simulation. On the other hand, if slipping does 565 not occur, the bending stiffness of a bilayer, D_2 , will be largely unrelated to D_1 as it then 566 derives directly from the in-plane stiffness (plus nD_1).⁸⁴ Neglecting D_1 for multilayers, we 567 have $D_2 = 2(h/2)^2 c_{11}^{2D}$, $D_3 = 2h^2 c_{11}^{2D}$ and $D_4 = 2[(3h/2)^2 + (h/2)^2]c_{11}^{2D}$ and for large n, we 568 recover the usual relation: $D_n = \frac{c_{33}}{1-\nu^2} \frac{(nh)^3}{12}$. The numerical values $D_1 = 0.7$, $D_2 = 131$ and 569 $D_4 = 1308$ eV are to be compared to the value of $D_1 = 2.1$, $D_2 = 130$ and $D_4 = 1199$ eV 570 reported by Sen *et al.*⁷³ 571

572 7. Shearing, sliding and friction between graphene layers

In multilayer graphene, as mentioned in the previous Section, whether shearing or sliding 573 occurs between layers determines the bending stiffness of the multilayer. In addition, a 574 Raman shear mode is observed at low wavenumbers, between 31 cm^{-1} (bilaver) to 43 cm^{-1} 575 (bulk).⁸⁵ Considering a linear chain model, it is possible to define an interlayer coupling 576 stiffness, $\alpha = 12.8 \times 10^{18} \text{ N} \cdot \text{m}^{-3}$. The same value fits the Raman from bilayer through to 577 graphite. It is a microscopic measure of the shear modulus, $\alpha d_0 = c_{44} = 4.3$ GPa, close to 578 the value of Table I. It corresponds to a spring constant $k = 0.419 \text{ N} \cdot \text{m}^{-1}$ for an unit cell. 579 The existence of this mode shows the corrugation of the graphene surface at the atomic 580 scale. It is also possible by friction to characterize atomically the surface. For a monolayer, 581 the presence of ripples can increase the friction by 40% compared to bulk graphite where 582 the layers are flat.⁸⁶ The friction results are well reproduced by calculation.⁸⁷ Finally, the 583 sliding, corresponding to electrostatic interactions and dispersive forces, and its dependence 584 on atomic direction has been calculated.⁸⁷ During the sliding, the interlayer distance changes 585 by 0.04 Å and the force is found to be in the range of 1.92 pN/atom considering no relaxation 586 of the atoms, which is close to the experimental value of 2.11 pN/atom. These values are 587 typical of flat 2D systems (h-BN).⁸⁷ 588

589 IV. MEASURING GRAPHENE MECHANICAL PROPERTIES

The nanoscale thickness of monolayer graphene makes the accurate measurement of its mechanical properties a challenging task. A number of techniques have been proposed in the literature^{88–90} to measure the mechanical properties of graphene membranes. Some of the most important ones are described here.

⁵⁹⁴ A. Atomic force microscopy

The use of AFM in the study of the mechanical properties of graphene usually involves 595 suspension of monolayer graphene over a substrate that has been previously patterned with 596 holes and then applying a local force to the surface of graphene with high precision. In 597 this way, the in-plane mechanical properties can be obtained. The work of Lee $et \ al.^{25}$ 598 was the first to measure the elastic modulus and fracture strength of graphene by using an 599 AFM tip to indent graphene that was suspended over circular wells. Force-displacement 600 (load-indentation) curves were obtained by indenting the membranes under constant speed. 601 The applied force can then be calculated. However, the exact theory is far from $simple^{91}$ 602 and many authors have used equations that appear to be over-simplified. The following 603 equation, for example, has been frequently used,²⁵ 604

$$F = \sigma_0^{2D}(\pi R)(\frac{\delta}{R}) + E^{2D}(q^3 R)(\frac{\delta}{R})^3$$

$$\tag{7}$$

where R is the radius of the circular well, δ is the indentation depth, σ_0^{2D} is the pre-tension 605 and E^{2D} is the 2D Young's modulus, and q can be expressed as $1/(1.05 - 0.15\nu - 0.16\nu^2)$, 606 where ν is the Poisson's ratio of graphene.²⁵ Given the uncertainties in the other parameters, 607 one may question the spurious precision implied by the inclusion of the factor q, which ranges 608 only from 0.95 to 1.05 over the whole range of possible values, $0 < \nu < 0.5$, and has the value 609 of 0.98 for the graphene value of $\nu = 0.16$. It is implausible that the contribution of the 610 pre-tension to the force, for a given depth, is independent of the radius of the well. Fitting 611 the data, values of the modulus and fracture strength of graphene are obtained. Similarly, 612 the breaking strength can be calculated from: 613

$$\sigma_{max} = \left[\frac{F_{max}E^{2D}}{4\pi R_{tip}^2}\right]^2 \tag{8}$$

where R_{tip} is the radius of the AFM tip and F_{max} is the force at which the membrane breaks.

A variation of the AFM nanoindentation method where graphene suspended over a circular hole, is the so-called beam bending method, where the 2D membrane is now in the form of a beam (or a stripe) and is suspended over a trough in the substrate. In this case the load-deformation relationship is:⁹²

$$F = \frac{Ew\pi^4}{6} \left(\frac{t}{L}\right)^3 \delta + \frac{\sigma_0 w \pi^2}{2} \left(\frac{t}{L}\right) \delta + \frac{Ew\pi^4}{8} \left(\frac{t}{L^3}\right) \delta^3 \tag{9}$$

where w, t, and L are the width, thickness and length of the beam, σ_0 is the intrinsic stress, $\delta_{20} \delta$ is the beam deflection, F is the load applied in the beam centre and E is the Young's modulus.

Although the majority of the AFM-based results in the literature agree (within large 622 experimental uncertainties) with values estimated from the bulk materials and with theo-623 retical calculations, it has been debated whether AFM nanoindentation can measure the 624 mechanical properties of macroscopic 2D membranes in a meaningful way.⁹³ The AFM tip 625 focuses on very small areas where the probability of defects is low. Intrinsic defects, rip-626 ples and crumples that are known to reduce the inherent properties of a "perfect" material 627 are very common in 2D materials, yet are sometimes ignored or overlooked. Parameters 628 that introduce uncertainties into the interpretation of data include the initial stress of the 629 2D membranes, the position of the indenter (which needs to be in the exact centre of the 630 membrane) and the indenter radius. Nevertheless, AFM nanoindentation is one of the most 631 popular experimental methods to measure the mechanical properties of 2D membranes and 632 has been used for a number of 2D materials. 633

⁶³⁴ Lee *et al.*²⁵ found the Young's modulus of monolayer graphene to be $E_{2D} = 340 \pm 50$ N ⁶³⁵ m⁻¹, corresponding to $E = 1.0 \pm 0.1$ TPa for a thickness of 3.35 Å. However, not all graphene ⁶³⁶ samples are flat. Nicholl *et al.*⁹⁴ and Ruiz-Vargas *et al.*⁹⁵ focused on crumpled and wrinkled ⁶³⁷ graphene prepared by CVD and found that the stiffness obtained by AFM nanoindentation ⁶³⁸ was reduced, compared to monocrystalline graphene.

A different, very useful application of AFM nanoindentation was reported by Cui *et al.*⁹⁶ They performed a fatigue study using a combination of static and cyclic mechanical loading of a suspended 2D film (Fig. 2 (a)). Monolayer and few-layer graphene survived more than 109 loading cycles at a mean stress of 71 GPa and a stress range of 5.6 GPa; that is higher than any material reported so far.



FIG. 2. (a) Schematic of the AFM fatigue testing set-up. Reproduced with permission from Nature Materials **19**, 405 (2020).⁹⁶ Copyright (2020) Springer Nature. (b) Strain contour map of a monolayer graphene flaked under 1.02% strain, where the strain distribution can be identified. The presence of a crack running through the upper part of the flake can be realised by the blue 0% strain line. Reproduced from Ref. 97 with permission from The Royal Society of Chemistry. (c) A deposited graphene sample in the centre of a push-to-pull micromechanical device, actuated by an external pico-indenter. The yellow arrow indicates the indentation direction during a tensile testing process. (d) Illustration of the graphene sample suspended between the device gap. (c) and (d) are reproduced from K. Cao, S. Feng, Y. Han, L. Gao, T. Hue Ly, Z. Xu, and Y. Lu, Nature Communications **11**, 284, 2020;⁹⁸ licensed under a Creative Commons Attribution (CC BY) license. (e) Schematic diagram of the bilayer graphene balloon. The left inset shows the increasing pressure on a graphene bubble, while the right inset shows the Raman G band of graphene across the balloon (line scan). Reproduced with permission from Phys. Rev. Lett. **119**, 036101 (2017).⁹⁹ Copyright (2017) The American Physical Society.

644 B. Raman spectroscopy

Raman spectroscopy has been used extensively for the study of the mechanical properties
of carbon-based materials as a result of their strong resonant Raman scattering. The welldefined characteristic Raman peaks enable the observation of a number of very interesting
phenomena and properties under uniaxial or biaxial strain.

For uniaxial strain, these experiments involve the deposition of monolayer or multilayer 649 graphene flexible onto a polymeric substrate, which is subsequently strained by 2-, 3- or 4-650 point bending under a Raman spectrometer. The application of strain leads to an elongation 651 of the C-C bonds and the lattice deformation is clearly and accurately evidenced by down-652 shifts of the Raman peaks. Mohiuddin $et \ al.^{33}$ first studied the deformation of monolayer 653 graphene deposited onto a flexible substrate under uniaxial tensile strain and found that the 654 shift of the 2D-mode is about $-60 \pm 5 \text{ cm}^{-1}/\%$ strain. This downshift corresponds to the 655 material having an elastic modulus in the order of 1 TPa and was subsequently confirmed 656 by a number of research groups.^{1,97,100,101} The G band is split by the uniaxial strain into 657 G^+ and G^- peaks. The shift of the G^+ band was $-10.8 \text{ cm}^{-1}/\%$ strain while the shift of 658 the G^- band was $-31.7 \text{ cm}^{-1}/\%$ strain. The frequency of the G-mode is related to the C-C 659 bond stiffness, though it contains non-negligible contribution from up to the fifth nearest 660 neighbour (more details in Sec. VIII).²⁶ It is nevertheless reasonable to consider that its shift 661 with strain has contribution only from the nearest neighbour, and is therefore determined 662 by the anharmonicity of the C-C bond. The physical meaning of the shift of the 2D-mode is 663 less clear as it is related to the evolution of the LO/TO phonon dispersion under strain.¹⁰² 664

The strength of monolayer graphene can be also studied by *in situ* Raman mapping. Zhao 665 et al.⁹⁷ prepared monolayer graphene by mechanical exfoliation, deposited the samples onto 666 polymer substrates and performed *in situ* Raman mapping at different strain levels to obtain 667 the strain distributions over the graphene flakes. Strain contour maps showed significant 668 events such as strain build up, edge effects and cracks which developed with increasing strain 669 (Fig. 2 (b)). Two main mechanisms of failure were observed: flake fracture, and failure 670 of the graphene/polymer interface. Low strengths were observed for these macroscopic 671 monolayer samples, only 10 - 15 GPa, an order of magnitude lower than the value of 130 672 GPa that was reported by Lee *et al.*²⁵ Simulations suggest that this was due to the presence 673 of defects.⁹⁷ Under large deformation, Raman observations also suggest a manifestation of 674

⁶⁷⁵ large nanometer-scale strain inhomogeneity within the laser spot size.^{103,104}

In multilayer graphene, Gong $et \ al.^{105}$ studied the effect of the layer number on the 676 downshift of the 2D band and found that monolayer and bilayer graphene displayed almost 677 the same redshift rate. On the other hand, with further increasing layer number, the shift 678 rate decreases significantly. The 2D Raman profile was fitted by a single Lorentzian, although 679 a broadened 2D profile is expected with increasing number of layers as it contains more 680 components,¹⁰² and the decrease of the 'average' shift rate was interpreted as showing some 681 slippage between the layers, *i.e.* reduced internal stress transfer. However, these spectra 682 were recorded at the centres of the flakes. Similarly to shear lag, slippage between layers or 683 between the lowest layer and the substrate should reduce the strain at the edge of the flake 684 first and propagate inwards as the strain is increased (Fig. 2 (b)). More detailed studies 685 would be desirable. 686

For biaxial strain, blister testing as described in Sec. IV D below is appropriate. G-mode shifts as large as -80 cm^{-1} are obtained and the deduced Gruneisen parameter of $1.8 \pm 10\%$ is compatible with biaxial strain.¹⁰⁶ The calculated slope (with some approximation on Y, D and d_{vdW}) $\Delta\omega_G/\epsilon = -57 \text{ cm}^{-1}/\%$ can be compared with later similar studies. Shin *et al.*¹⁰³ achieved biaxial reversible strain up to $\approx 2\%$. Using the same approximations, they found $\Delta\omega_G/\epsilon = -62 \text{ cm}^{-1}/\%$. With large biaxial strain, the linewidth of the G-mode increases.

Interlayer modes, the layer breathing mode and shear mode, are convenient measures of 693 adhesion and shear strength between graphene layers. A linear chain model describes the 694 frequency of the shear mode, and its change with number of graphene layers very well, as 695 discussed in Sec. III D 7. The model applies to the layer breathing mode too.¹⁰⁷ The layer 696 breathing mode becomes Raman-active, when graphene layers are twisted from AB stack-697 ing. Its intensity is usually very weak, and requires resonance condition to be observable. 698 Resonance Raman spectroscopy is particularly useful to study graphene and CNTs samples, 699 where the resonance condition is that the energy of the in-coming or out-going laser matches 700 the gap between van Hove singularities in these low-dimensional samples.^{108,109} 701

702 C. In situ tensile tests

For *in-situ* tensile tests, nanomechanical testing devices are usually introduced within a scanning electron microscope (SEM) or transmission electron microscope (TEM) and the

deformation of the graphene is followed in the images. To study its fracture toughness, Zhang 705 $et \ al.^{110}$ suspended nanocrystalline graphene over the jaws (gap) of a micromechanical device 706 driven by a nanoindenter, within an SEM chamber. This imposed uniaxial tension on the 707 graphene. Brittle fracture was observed when a central crack had been machined by FIB 708 in the graphene samples prior to testing. The fracture toughness of graphene (important 709 for engineering applications) was found to be $K_c = 4.0 \pm 0.6$ MPa, while the critical strain 710 energy release rate was $G_c = 16 \pm 5 \text{ J} \cdot \text{m}^{-2}$ (where we estimate the uncertainty on G_c from 711 their data). Cao et al.⁹⁸ reported in situ tensile tests within an SEM chamber and measured 712 the elastic properties and stretchability of monolayer CVD graphene (Fig. 2 (c,d)). The 713 Young's modulus was ~ 1 TPa, while the tensile strength was around 50 - 60 GPa, when 714 the elongation of the sample was $\sim 6\%$. Once again the actual strength of macroscopic 715 graphene samples is significantly lower than the value of 130 GPa reported by Lee *et al.*²⁵ in 716 AFM nanoindentation. Brittle fracture initiated from the edges of the samples, suggesting 717 that control of the edge states and edge effects could lead to greater strength. 718

In situ TEM can offer useful information on cracks and defects in graphene. For example, 719 Fujihara et al.¹¹¹ observed that crack propagation takes place along a specific crystallographic 720 direction, in order to create zigzag edges. Kim $et \ al.^{112}$ showed that the presence of grain 721 boundaries influences crack growth. When the stress is normal to the grain boundaries, 722 the crack can follow the boundary; however, if the crack is initiated away from the grain 723 boundaries and at some random orientation with respect to it (with the strain direction not 724 normal to the grain boundary), the stiffness the tear experiences is more or less unaffected 725 by the grain boundary and the crack will pass through the grain boundary, switching to the 726 most favourable direction in the next grain. 727

728 D. Pressurized blister method

The use of the pressurized blister (bulge, bubble, or balloon) method can provide information on the mechanical properties of 2D membranes and the interfacial adhesion between the substrate and the membrane. For these experiments, graphene is again suspended on top of a hole, or microcavity, in the substrate. The vdW forces between the substrate and the membrane hold the sample in place. Gas is fed in the hole thus pressurising the membrane. This leads to a spherical blister with a radius R. AFM is used to measure the blister (compare Raman methods in Sec. IV B). The relation between its height, δ , and the pressure difference inside and out, Δp , is,¹¹³

$$\Delta p = K(\nu) (Ed\delta^3) / R^4 \tag{10}$$

where E is the Young's modulus, and d is the graphene thickness. $K(\nu)$ is a coefficient depending on Poisson's ratio only and is very close to 3. Thus the elastic modulus can be calculated from the measured AFM deflections.

The group of Bunch¹¹⁴ first measured the adhesion of graphene on a silicon oxide substrate 740 in this way. The adhesion strength is revealed by the pressure at which the blister diameter 741 begins to exceed the hole diameter, i.e. when the graphene begins to peel away from the 742 substrate. The relationship is given by Bodetti et al as Eq. 14 of their paper.¹¹⁵ They obtain 743 adhesion energies of 0.45 \pm 0.02 J/m² for monolayer graphene and 0.31 \pm 0.03 J/m² for 744 few-layer graphene (2-5 sheets). With its high flexibility, graphene, especially monolayer 745 graphene, is able to conform to the topography of very smooth surfaces, thus leading to high 746 values of adhesion energy. However, much lower values may be observed as adhesion between 747 a graphene sheet and a substrate (or surface) is highly dependent on the surface conditions 748 such as moisture, roughness, chemical reactivity and others and a considerable spread on 749 the adhesion energies has been reported in the literature (see Ref. 2 and references therein). 750 Similarly, weakening of the adhesion in turbostratic graphite compared with Bernal graphite 751 is expected.^{5,116} In a more recent report, Wang *et al.*⁹⁹ measured the interlayer shear stress 752 of bilayer graphene by monitoring the strains in the graphene next to the blister but not 753 lifted off the substrate (Figure 2 (e)). Here strain develops as a consequence of sliding. Their 754 data provided evidence of both the lower monolayer sliding on the substrate, and the upper 755 layer sliding on the lower layer. Analysis of the data gave the interfacial shear stress of 756 monolayer graphene on SiO_2 as 1.64 MPa. This was much higher than the interlayer shear 757 stress of bilayer graphene, 40 kPa (1.04 fN/atom). The implication is that the graphene-758 SiO_2 vdW bonding is much stronger that the weak dispersion vdW bonding that holds the 759 two graphene sheets together. 760

⁷⁶¹ E. Inelastic X-ray scattering

As discussed in Section III, some key elastic moduli of graphene come directly from the experimental values for graphite, measured by inelastic x-ray scattering (IXS). This measures

the acoustic phonon branches of graphite.²⁰ The initial slope of these branches along high-764 symmetry directions gives the sound velocity, and therefore the moduli by Christoffel's 765 equation.¹¹⁷ IXS is less sensitive to structural defects than ultrasonic methods and it does 766 not have the difficulties in sample size and energy transfer limitations that inelastic neutron 767 scattering suffers.²⁰ Experimentally IXS employed to measure graphite is not suitable for 768 measurements on graphene due to the small sample volume. The alternative approaches 769 to measure the in-plane elastic moduli of graphene are introduced above, and the results 770 largely depend on interactions of graphene with its surroundings (substrates and/or pressure 771 transmitting media, PTM). Details will be discussed in Section V. 772

773 F. Density functional theory

This section particularly addresses properties that are anisotropic in-plane, which have 774 been calculated but not confirmed by experiments. Whereas it is extremely difficult to 775 do any mechanical testing on free-standing graphene, the in-plane graphene mechanical 776 properties calculated by DFT are obtained naturally freestanding, and in vacuum. It is 777 commonly considered that the calculated in-plane elastic constants are accurate (and in-778 deed they are close to the well-established experimental results on graphite)²² and they 779 are often used to parameterise various empirical models and evaluate the validity of classic 780 simulations.^{22,118–121} Strain is the input in a computational model and the resulting energy 781 and stress are calculated. The ease with which the positions of the atoms can be specified 782 enables the investigation of a number of anisotropic in-plane properties that result from 783 strain along specific directions (armchair or zigzag). 784

The undeformed hexagonal lattice is isotropic. While in-plane uniaxial strain breaks 785 the symmetry, the isotropy remains in the sense that most properties vary with the 786 strain along most directions very similarly, except two specific directions.^{122,123} We present 787 clear anisotropic response of four properties to uniaxial strain along zigzag or armchair 788 directions.^{121,124,125} Fig. 3 (a) presents that the in-plane Poisson's ratio (corresponding to 789 the elastic modulus c_{12}) shifts differently with uniaxial strain along armchair or zigzag 790 direction, and the stiffness (corresponding to c_{11}) becomes different after 15% strain along 791 armchair or zigzag direction.¹²⁴ Fig. 3 (b) shows that stress-strain curves of the shear de-792 formation (relevant to c_{66}) along these two directions are different.¹²¹ The ultimate stress of 793

graphene along these two directions is also calculated to be different by about 20%.¹²⁰ Zhou 794 et al. plotted the electron density of graphene at zero, biaxial and uniaxial stress, as shown 795 in Fig. 4.¹²⁵ In Fig. 4 (c), the high electron density area (red area) at uniaxial stress along 796 zigzag direction attracts the nuclei at corners more than the central areas of those stretch 797 bonds attracting those corresponding nuclei. That extra attraction stiffens the three-atom 798 bending force constant labeled by k_{angle} , which supports the uniaxial stress. Differently, 799 along uniaxial stress along the armchair direction, the supporting bending force constant 800 (labeled as k_{angle} in Fig. 4 (d)) is stiffened by two high electron density areas (two red areas 801 on the right). This was proposed as a possible interpretation for the anisotropy in ultimate 802 strength¹²⁵ and could be relevant to other anisotropic properties. 803



FIG. 3. (a) "The curves connected to the origin are the equivalent tensile stress (d_0 =3.34 Å) versus uniaxial strain in the x and y directions, respectively. The lines with initially negative slopes (scale labels to the right) are the finite-deformation Poisson's ratios as functions of the uniaxial strain in the x and y directions, respectively. The red circles and triangles indicate the condition where peak stress could be attained for zigzag and armchair nanotubes, respectively." Reproduced with permission from Phys. Rev. B **76**, 064120 (2007).¹²⁴ Copyright (2007) The American Physical Society. (b) "Stress-strain curves for shear deformations of graphene monolayers, obtained through MD simulations. Filled circles (open squares) show results corresponding to shear forces acting on the armchair edge (zigzag edge) of graphene. Dashed lines are guides to the eye." Reproduced from G. Kalosakas, N. N. Lathiotakis, C. Galiotis, and K. Papagelis, Journal of Applied Physics **113**, 134307 (2013),¹²¹ with the permission of AIP Publishing.



FIG. 4. "The electron density distribution contour of graphene under different tensile loads, (a) ϵ =0; (b) biaxial tension at ϵ =0.28 (the bond stretching ratio ξ = 0.25); (c) uniaxial tension along the zigzag direction at ϵ =0.28 (the bond stretching ratio ξ (Bond I)=0.2, ξ (Bond II)=-0.03); (d) uniaxial tension along the armchair direction at ϵ =0.28 (the bond stretching ratio ξ (Bond I)=0.2, ξ (Bond II)=-0.03); (d) uniaxial tension along the armchair direction at ϵ =0.28 (the bond stretching ratio ξ (Bond I)=0.36, ξ (Bond II)=0.003). Reproduced with permission from Phys. Chem. Chem. Phys. **18**, 1657 (2016).¹²⁵ Copyright 2016 The PCCP Owner Societies.

⁸⁰⁴ V. GRAPHENE IN INTERACTION WITH ITS ENVIRONMENT AT ⁸⁰⁵ HIGH PRESSURE

The mechanical properties of graphene can be probed by high pressure. In bulk materials, 806 high-pressure experiments can characterize the bulk modulus and its pressure derivative 807 through the equation of state, which links the volume variation to pressure. Many bulk 808 systems are anisotropic, and have an anisotropic elastic stiffness tensor c_{ij} . With more than 809 one atom per primitive cell, as in graphene, internal strains may occur. Particularly for 2D 810 systems, bond or directional compressibilities can be defined to fully describe the changes in 811 atomic positions due to the deformation under hydrostatic pressure.¹²⁶ Graphene is however 812 not only highly anisotropic but also is only one atom thick. While its in-plane deformation 813 modifies the C-C separation or sp^2 bond length, its out-of-plane deformation can only be 814

⁸¹⁵ described as deformation of the π -orbitals.

In high-pressure experiments, both the C-C sp^2 bonds and the spatial extension of the 816 π -orbitals will be modified by interactions with the surrounding PTM (anything from he-817 lium, argon, through to water, ethanol or various oils), which itself increases in density 818 with pressure. Interactions between graphene and the surrounding medium can range from 819 weak vdW to strong covalent interactions. In particular, the vdW interactions will become 820 stronger under pressure. The π -orbitals may become highly modified under pressure, per-821 haps creating strong dipolar interactions with neighboring molecules. At what point could 822 we consider the system to be something other than graphene (as, *e.g.* graphene oxide)? The 823 effects of pressure may be simply to change the graphene thickness, or may include doping 824 or hybridization. As in any material, the c_{ij} are expected to increase with pressure, but 825 since all the carbon atoms in graphene are in contact with the environment, we may expect 826 that c_{ij} will depend on pressure and temperature, and also on the environment. It is likely 827 that c_{33} will be most influenced by the pressure and the environment. But are the pressure 828 dependencies of other parameters dominated by the pressure or by the environment? How 829 best to compare the pressure responses of graphene and graphite? These issues are addressed 830 here. 831

Graphene can be studied in suspension, with PTM on both sides, and when the PTM is liquid, the graphene is free of non-hydrostatic strains. When graphene is supported by a substrate, the situation is quite different. We must consider what differences are significant, and what can be learned from these different environments.

This section has three parts treating the pressure response of graphene in the different environments depicted in Fig. 5. First, we will discuss the case of graphene in suspension in a fluid as shown in Fig. 5 (a). Then, the case of supported graphene immersed in a fluid PTM (b). Finally, the case of graphene sandwiched between two different (Fig. 5 (c.1)) or identical (Fig. 5 (c.2)) solids. The Fig. 5 (c) cases are also relevant to help understand the mechanical response of graphene in composite materials.

⁸⁴² A. Suspended graphene in a fluid PTM

This may be considered as the paradigmatic pressure experiment, with the two variants of Fig. 5(a). A few Raman scattering experiments have been reported corresponding to



FIG. 5. The various cases encountered in high-pressure experiments and discussed in this section are: (a) suspended graphene in a fluid PTM, (b) supported graphene in a fluid PTM, and (c) graphene sandwiched between two solids. In cases (a) and (c), one needs to distinguish between (a.1) free-standing graphene and (a.2) graphene across a hole ; and (c.1) graphene sandwiched between two different solids and (c.2) graphene in a single solid. The presence of ripples is shown in (a.1) and wrinkles in (a.2).

these configurations: an ensemble of graphene layers in suspension and considered separated each from the others (Fig. 5(a.1)) or supporting an individual graphene layer on a hollowed substrate (Fig. 5(a.2)). In both cases, we may note that the graphene may not be flat due to the spontaneous formation of ripples (Fig. 5(a.1)) and wrinkles (Fig. 5(a.2)).^{127,128} Wrinkles and ripples differ by their aspect ratio.¹²⁷ Ripples are isotropic, with an amplitude ~ nm, and an aspect ratio ~ 1. Wrinkles are more aligned and larger, having an aspect ratio > 10, due to the partial decoupling of bending and stretching modes.^{50,127,128}

In Fig. 5 (a.2), while the graphene sheet may be under tension at ambient pressure (resulting in wrinkles with an axis perpendicular to the trench),¹²⁷ we might expect that differential contraction under pressure of the support and the graphene (unless the support is made of diamond) would result in a loss of tension and eventually buckling in a different direction or even crumpling.¹²⁷ However, this may be prevented by the evolution of the adhesion of graphene to the internal walls of the trench. Bunch *et al.* observe a minimum

Туре	PTM^{1}	$P_{hydrostatic} (P_{Max})$	$\frac{\partial \omega_G}{\partial P}$ hydro (P_{Max})	Ref
		[GPa]	$[\mathrm{cm}^{-1}/\mathrm{GPa}]$	
$suspension^{(2)}$	N_2	2 (8)	4.7	34
supported ^{(3)}	Fluorinert	1.0(4.2)	- (5.6)	37
suspension	DMF	1.6(7)	5.4	40
supported ^{(4)}	H_2O	1.0(35)	- (3.4)	131

TABLE IV. Summary of high-pressure Raman experiments on suspended graphene.

(1) see text

(2) The sample were films with a mixture of monolayer, bilayer and few-layer graphene and having 2D characteristic signature of few-layer graphene rather than single-layer graphene.

(3) From the detachment from graphene supported on Si/SiO_2 using as PTM a 1:1 mixture of Fluorinert FC70 and FC77 (nonpolar).

(4) A gold microscopy grid was used to suspend graphene.

⁸⁵⁸ non-zero tension,¹²⁹ due to this, and analysed by Lu and Dunn.¹³⁰

All high-pressure experiments on suspended graphene have used Raman spectroscopy as a probe. Raman spectroscopy does not give direct access to the pressure evolution of the unit cell parameters which would determine the elastic constants. Raman spectroscopy provides nevertheless a signature of the response of the C-C interatomic potential through the Gmode pressure dependence. Table IV summarizes the most relevant results from suspended graphene studies.

All the studies in Table IV were done using different PTM (FC-70 is perfluorotripenty-865 lamine and FC-77 is a perfluorocycloether; DMF is N,N-dimethylformamide). Only two 866 studies (using nitrogen and DMF as the PTM) were performed in hydrostatic conditions 867 throughout the pressure range (*i.e.* with a liquid PTM). On the other hand, the nitrogen 868 PTM study was of a mixture of monolayer, bilayer and few-layer graphene, which makes its 869 results difficult to interpret. The DMF PTM study is the only one in which a G-band pres-870 sure dependence has been obtained for a monolayer graphene sample suspended in liquid. 871 This study was able to obtain a $\frac{\partial \omega_G}{\partial P}$ in hydrostatic conditions from a careful analysis from 872 a total of 4 pressure points in the DMF hydrostatic domain up to 1.6 GPa. The value of 873 5.4 $\rm cm^{-1}/GPa$ was obtained with a linear fit and 5.6 $\rm cm^{-1}/GPa$ with a quadratic fit using 874
the quadratic coefficient of the graphite fit. 28 These are 15 to 25 % higher than the graphite 875 G-mode slope which is $4.3 - 4.7 \text{ cm}^{-1}/\text{GPa}$ using the same quadratic coefficient.²⁸ This 876 difference is partly due to the non-negligible impact of interlayer-coupling on the in-plane 877 vibrations in graphite (see Sec. VIC). There are reports pointing to PTM-induced charge 878 transfer to graphene in polar media^{132,133} even from the first stages of compression,¹³³ and of 879 pressure-induced reactivity with water.¹³⁴ Doping, p or n, leads to an enhanced value of the 880 G-mode frequency¹³⁵ which may explain an increase of the G-mode pressure coefficient in 881 graphene. Experiments using an inert PTM would clarify whether there is a difference be-882 tween graphene in-plane elastic constants and those of graphite, or if the observed differences 883 are related to polarization-induced modifications of the electronic structure of graphene. 884

Another interesting aspect is that, as pressure is increased, the PTM fluid viscosity 885 increases. The vdW graphene-fluid interaction may overcome the thermal energy of the 886 fluid molecules. An organized fluid layer may then appear in contact with graphene. This 887 constitutes a loss of 2D character which may also affect the mechanical properties of graphene 888 and may explain differences in the Raman response with different PTM. DFT and MD 880 modelling - DFT is performed at zero temperature - have in fact shown in carbon nanotubes 890 under high pressure the formation of a structurally coherent contact layer of CO_2 or water 891 around carbon nanotubes.^{136,137} In fact, DFT modelling shows that in carbon aromatic 892 systems such as benzene, vdW interaction plays an important role in phase stability at high 893 pressures.¹³⁸ 894

We may conclude this part by underlining that dipole or other enhanced VdW interactions with the PTM molecules and the possible loss of 2D character through the formation of a PTM contact layer need to be explored as possible mechanisms modifying the graphene mechanical properties at high pressure.

⁸⁹⁹ B. Supported Graphene in a fluid PTM

This is the case depicted in Fig. 5 (b), *i.e.* graphene on a substrate and immersed in a fluid PTM. In this case, the PTM applies hydrostatic pressure on the graphene-substrate system. With increasing pressure, the substrate (unless it is diamond) contracts much faster than the graphene, which puts the graphene under a large biaxial compression. How much graphene contracts and is strained in response to the substrate shrinking is governed by the graphene-substrate adhesion and friction, graphene/PTM adhesion, graphene stiffness, and
graphene bending modulus. We will now discuss the role of each parameter.

907 1. Role of the substrate

Graphene stiffness is one of the largest-ever measured, with a Young's modulus of ~ 1 TPa 908 (Table II),^{25,139} while its bending stiffness is often considered as negligible;^{140,141} indeed so 909 small that it is hard to measure in a direct manner. Many reported values (depending on 910 the temperature and flake size) are often about 1 eV,¹⁴² and it was measured in carbon nan-911 otubes as 1.7 eV (see Sec. III D 5).⁴⁵ Thus, it is expected that graphene will tend to bend or 912 wrinkle rather easily in order to relieve in-plane compression.^{143–145} The interaction between 913 the graphene layer and its supporting surface, namely the graphene-substrate adhesion, is 914 therefore an extremely important parameter governing the graphene response to biaxial 915 compression: if the graphene-substrate adhesion is poor, the graphene will not fully follow 916 the substrate's deformation and instead will slide or wrinkle to reduce its stress. However, 917 the adhesion of graphene to its substrate is an intricate mixture of (a) the interaction energy 918 between carbon and surface atoms, (b) substrate surface roughness, (c) graphene number 919 of layers, (d) commensurability of the graphene and substrate lattices, and (e) the normal 920 force from the PTM, which must modify the effects of (a) to (d). As a consequence, each 921 graphene-substrate system is unique and the amount of strain transferred from the substrate 922 to the graphene layer can only be assessed in a phenomenological way, for example with a 923 "strain transfer efficiency" parameter α .^{38,146} Now to discuss each of these parameters: 924

(a) The strength of the graphene-substrate interaction energy can be assessed from the 925 graphene-substrate distance, which varies greatly from one substrate type to another - as 926 shown in Table I which gives some experimentally measured distances. Graphene deposited 927 on metals tends to form covalent bonds that greatly decrease the graphene-substrate dis-928 tance. This would correspond to a much increased adhesion energy and forces for peeling 929 off and for sliding. One thus expect the strain transfer efficiency α to be close to 1 in such 930 systems – and $\alpha = 1$ for a copper substrate was indeed observed up to a critical stress of 931 2 GPa.^{34,38} Moreover, the history of the graphene sample plays a huge role in the interaction 932 with its substrate: whether the graphene was transferred or grown on the substrate has 933 a large impact on its adhesion and residual stress. As discussed in Sec. IIID3, epitaxial 934

graphene is under large strain after cooling (hence showing $\alpha = 1$), whereas CVD (usually on Cu) graphene shows the occurrence of ripples to release the stress (resulting in a $\alpha < 1$). An even lower α on Cu is expected if graphene is transferred onto it.

(b) When the surface roughness is too high, total unbinding from the substrate surface 938 can occur, at ambient pressure^{145,147} or under high pressure.³⁷ Some substrates show the oc-939 currence of a critical stress beyond which the strain transfer efficiency is greatly diminished,³⁸ 940 which is probably due to a partial unbinding of the graphene layer from its substrate. While 941 not measured experimentally, this critical stress is supposed to be roughness-dependent for 942 a given substrate. The substrate surface roughness also greatly influences the *friction* be-943 tween the graphene layer and its substrate, which in turns plays a role on the mechanism of 944 the mechanical response of graphene to biaxial compression. When the strain transfer effi-945 ciency $\alpha < 1$, it means that either (a) graphene slips all over the substrate; or (b) graphene 946 wrinkles, ripples or crumples. Or a combination of (a) and (b) may occur. In the case of a 947 rough substrate with a conforming graphene (as is the case most of the time), case (a) has 948 an extended energy cost because it involves the whole graphene surface, while case (b) only 949 involves the local elastic energy cost of buckling. Depending on the substrate roughness, a 950 varying proportion of slipping and buckling may thus occur. However, the application of a 951 normal force by the hydrostatic pressure to the surface will certainly hinder the formation of 952 wrinkles or ripples. Wrinkles were however observed under hydrostatic pressure (at 4 GPa) 953 even in nano-graphite (~ 30 -layer graphene).¹⁴⁸ 954

(c) In the same manner, adhesion to a rough substrate is decreased when the number of graphene layers increases, *i.e.* when the graphene bending rigidity increases.³⁶ With an increasing number of layers, the conformation of graphene to its rough substrate may decrease, resulting in a decreased friction. We may note that in MoS_2 , which has a much higher bending stiffness than graphene, a bimodal behavior has been seen when supported on Si/SiO_2 . A mixture of regions showing strong adhesion and other regions showing weak adhesion was seen.¹⁴⁹

(d) Finally, it was for example shown that for a given substrate such as Co(0001), the graphene-substrate distance decreases when the graphene lattice is matched with the substrate lattice,¹⁵⁰ allowing for covalent C-Co bonding. Such an effect therefore plays a large role on the starting *stress+doping* state of graphene. In the following however, we will not consider substrates in which this type of interaction may happen. Pressure-induced doping effects will then be due essentially to interaction with the PTM and will be discussed in
the following subsection. One should also note that the (in)commensurability of the lattices
should also play a role in the friction between graphene and its substrate.

(e) All these parameters may be impacted by the modification of the graphene-substrate distance due to the application of pressure. However, in pressure ranges for which the PTM is liquid (hydrostatic pressure), the graphene Raman response is always linear,^{34,37,38,40,131,133} which tends to show that this effect is limited at pressures below ~ 10 GPa.

The mechanical response of graphene to high pressures is usually followed through in situ 974 Raman spectroscopy, as stress induces a deformation of the carbon bonds and thus shifts 975 the Raman features such as the G-mode (ω_G) or the 2D-mode (ω_{2D}). But similar shifts can 976 also arise from doping. Graphene is a π -conjugated 2D material, thus its electronic struc-977 ture is highly sensitive to its environment. This charge sensitivity allows doping through 978 gating or intercalating ("substrate doping", see e.q. Ref. 151 for a review), but it can also 979 occur due to the interaction with polar molecules in the PTM which can be enhanced at 980 high pressure¹³³ ("PTM doping"). Overall, the doping and strain contribution to the ω_G 981 shift can be unravelled by following the slope of $\partial \omega_{2D} / \partial \omega_G^{151,152}$ (see Fig. 6). Finally, the 982 substrate type and the graphene preparation method (as-prepared, transferred, synthesised, 983 exfoliated...) play a large role in the ω_G frequency^{150,151} and the graphene-substrate dis-984 tance (Table I). Large variations of the ω_G peak position can be observed for graphene on 985 a substrate according to the crystallographic orientation, matching or random. For exam-986 ple, while $\omega_G = 1581 \text{ cm}^{-1}$ for free graphene, it redshifts down to 1452 cm⁻¹ for oriented 987 graphene on Co(0001), and it can vary between ~ 1550 and 1600 cm⁻¹ across a single sam-988 ple of misaligned graphene on Co(0001).¹⁵⁰ In the case of Co(0001), matching the graphene 989 and substrate lattice orientations allows chemical bonding between the two, resulting in a 990 shortening of the graphene-substrate distance and a significant stretching of the graphene 991 lattice, thus decreasing the ω_G frequency. However, this is peculiar to Co(0001) as, usually, 992 the CVD synthesis of graphene on a metallic substrate results in a compressive stress of the 993 graphene layer, and thus an increased ω_G frequency.¹⁵¹ On the other hand, the incommen-994 surability of both lattices can result in large local variations of the ω_G position, that can be 995 due to both strain and/or doping. 996

⁹⁹⁷ In conclusion, the substrate plays a very important role both at ambient pressure and ⁹⁹⁸ under high pressure. Before the application of pressure, and together with the history of



FIG. 6. Lee diagram¹⁵² allowing retrieving the strain and doping (ε_X and n_X) at a point X in the (ω_G, ω_{2D}) space. The origin $(\omega_G, \omega_{2D})_0$ is the reference state at ambient conditions. Reproduced with permission from J. Raman Spectrosc. **49**, 130 (2018).¹⁵¹ Copyright 2018 John Wiley & Sons, Ltd.

the sample, the substrate determines the reference state of the graphene, *i.e.* the graphene-999 substrate distance as well as the initial stress and doping states. During pressure application, 1000 it determines the strain of the graphene layer through the phenomenological parameter α , 1001 this strain being relative to the initial state. It is inappropriate to consider the evolution 1002 of the Raman G-mode frequency as a function of pressure;³⁸ rather, it should be considered 1003 as a function of the strain, the actual strain depending on the reduction of the substrate 1004 dimensions. Overall, in the simple case where $\alpha = 1$ and in the absence of doping, a Raman 1005 G-mode frequency dependence on substrate strain ε of $\frac{\partial \Delta \omega_G}{\partial \varepsilon} = -60 \pm 3 \text{ cm}^{-1}/\%$ is expected.³⁸ 1006 Since $\frac{\omega}{\omega_0} = (1+\varepsilon)^{-2\gamma_{E_{2g}}}$ for in-plane biaxial compression,³⁴ this corresponds to a Grüneisen 1007 parameter $1.8 < \gamma_{E_{2g}} < 2.0$ – in excellent agreement with uniaxial strain experiments³³ and 1008 ab initio modeling.¹⁵³ This further confirms the importance of substrate-induced biaxial 1009 strain on the properties of graphene. 1010

1011 2. Role of the PTM

Mechanical response When the PTM is fluid, it has a much larger compressibility а. 1012 than the substrate. This, of course, is of no importance while it exerts purely hydrostatic 1013 pressure. However, fluid PTM display large viscosities at pressures in the GPa range, and 1014 local crystallisation of the fluids occurs at the graphene surface.¹⁵⁴ Whatever the state of 1015 the PTM, this configuration results in a biaxial stress field in the system: the top surface 1016 of the graphene layer is in contact with the fluid that applies hydrostatic stress, while 1017 its bottom surface is in contact with the solid substrate which imposes a bi-axial stress 1018 that is determined by the equation of state of the substrate. By symmetry, there are no 1019 shear stresses σ_{xz} and σ_{yz} applied to the graphene, except at the edges and other lateral 1020 inhomogenities. 1021

In the case of monolayer graphene, the mechanical response of the graphene layer is mainly 1022 governed by the substrate.^{34,36–38,146} In the case of bilayer graphene, that remains so while the 1023 PTM remains liquid¹³³ (however, after solidification of the PTM, a stress gradient between 1024 layers could be measured, as will be discussed in the next section). One might expect, indeed, 1025 that the application of pressure would result in an increased graphene-substrate adhesion by 1026 reducing the graphene-substrate distance; and also that it would prevent the graphene layer 1027 from buckling and hence forming wrinkles or ripples. As mentioned in (b) above, substrate 1028 surface roughness can cause a critical stress above which the graphene layer is not able to 1029 follow the deformation of its substrate.³⁸ So the reality is more complex: despite gigapascals 1030 of pressure applied on the graphene layer, it seems that it is still able to buckle to reduce 1031 $\rm stress.^{148}$ 1032

To sum up, while the PTM remains liquid, the evidence is that the *mechanical response* of graphene is fully governed by the adhesion to the substrate. This goes for mono- and bilayer graphene only; for thicker samples the graphene bending rigidity increases and the adhesion to the (usually rough) substrate is reduced, leading to a mechanical response closer to graphite.^{36,146}

¹⁰³⁸ b. Electronic response Comparing inert PTM (such as Ar, N₂) with polar PTM (such ¹⁰³⁹ as 4:1 methanol:ethanol), a significant increase in the pressure coefficient of the Raman G-¹⁰⁴⁰ band was reported, from ~ 7 to ~ 10 cm⁻¹ GPa⁻¹ for SiO₂/Si substrate), for both mono-¹⁰⁴¹ and bilayer graphene.³⁶ This change was however later refuted.^{37,38} Nevertheless, a pressure-

induced decrease of the G-band FWHM was observed when using 4:1 methanol:ethanol 1042 PTM, which was attributed to a doping contribution.³⁶ However, it was not clear whether 1043 this doping remained constant over the whole pressure range.³⁷ A recent study of twisted 1044 isotopically-labelled bilayer graphene in 4:1 methanol:ethanol PTM by Forestier $et \ al.$ ¹³³ 1045 clarified the doping effect of the PTM. Here, the observation of a difference in the response 1046 of the two layers made it possible to conclude that there was a pressure-induced doping due 1047 to the alcohol PTM: the G-mode of the upper layer in contact with the PTM showed a 1048 larger pressure shift than the lower layer in contact with the substrate, demonstrating the 1049 occurrence of a doping due to the polar PTM which increases with pressure. It is worth 1050 mentioning here that the isotopically-labelled bilayer graphene of this study is made of two 1051 CVD graphene layers transferred on top of each other, resulting in a sample behaving like 1052 two independent layers on top of each other. The difference in isotopic masses decouples the 1053 G-modes of the two layers, so that they can be separately resolved. 1054

¹⁰⁵⁵ C. Graphene sandwiched between two solids

The investigation of the mechanical behavior of graphene monolayers and bilayers inter-1056 acting with solids under pressure on both sides, as in Fig. 5 (c), constitutes an interesting 1057 route to better understand the mechanical properties of graphene-based nanocomposites. 1058 Under sufficient pressure, any fluid transforms to a solid at ambient temperature. There-1059 fore, increasing the pressure sufficiently in the two situations discussed in the previous sub-1060 sections, the systems will evolve to i) graphene between two different solids (Fig. 5 (c.2), for 1061 supported graphene on a substrate) or ii) graphene between two identical solids (Fig. 5 (c.1), 1062 for graphene in suspension). This leads to an asymmetrical or to a symmetrical environment. 1063 Crystallization of the PTM has at least two effects. First, macroscopically, it modifies the 1064 stress field with the appearance of differential strain at the graphene-PTM interface. Second, 1065 microscopically, it creates a periodic potential in interaction with the graphene sheet. 1066

1067 1. Different solids

¹⁰⁶⁸ Crystallization of the PTM when compressing a supported graphene leads to an asymme-¹⁰⁶⁹ try of the environments and of the associated stress fields. Each side of the graphene plane ¹⁰⁷⁰ is in contact with a different solid medium. However, the change of state (fluid to solid) of ¹⁰⁷¹ the PTM is usually unnoticed in high-pressure Raman spectroscopy on supported monolayer ¹⁰⁷² graphene.¹⁴⁶ The preferred PTM are soft vdW solids (such as argon or nitrogen) and are ¹⁰⁷³ considered to provide quasi-hydrostatic conditions even after solidification.¹⁵⁵ The bonding ¹⁰⁷⁴ of such solids only leads to weak vdW type interactions with the graphene. The interactions ¹⁰⁷⁵ between these solids and the graphene can only marginally modify the pressure-induced ¹⁰⁷⁶ behavior which remains dominated by the effect of the substrate.¹⁴⁶

In contrast, crystallization of the PTM is clearly observed by Raman spectroscopy experiments on bilayer graphene where each layer experiences different conditions. The signature of the solidification is a change in the pressure-dependence of the G-peak position and/or a change of the width of this peak. The change of slope may be attributed to additional differential strain components and the broadening is related to inhomogeneity of the stress field. As the spectroscopic signatures are mainly affected by these external effects, it is difficult to assess any intrinsic effect, *i.e.* any modification of the elastic properties of graphene.

The difference of the applied stress on each side of a twisted bilayer graphene have been evidenced and quantified by the high-pressure Raman experiment on isotopically labelled bilayer graphene mentioned in Sec. V B 2 b. Strain differences up to ~ 0.1 % between the two graphene layers were observed when applying pressures of up to 10 GPa with nonpolar solid environments.¹³³

1089 2. Identical solids

There are only a few reports of high-pressure experiments on graphene in suspension 1090 above the crystallization pressure of the PTM. It represents a major experimental challenge 1091 to characterize and manipulate a mono- or bi-layer, to load it in a high-pressure cell as 1092 suspended and to follow experimentally the high-pressure behavior across and above the 1093 solidification of the PTM. A particularly important issue is the form of suspension. If the 1094 graphene is freely floating in the PTM, then after solidification it will be subject to the 1095 strain of the PTM as pressure is further increased and the sample volume decreases and 1096 changes shape. However, if the graphene is supported over a trench in a substrate, then 1097 after solidification it will be pressed down into the trench, as the PTM will typically be 1098 much more compressible than the substrate, and this will put the graphene under high 1099

1100 tensile strain.

Table IV includes data from studies before and after solidification of the PTM. Sun et al. 1101 dissolved PMMA on which CVD-grown graphene had been transferred.⁴⁰ The solvent (DMF) 1102 was used as the PTM. So this graphene was assumed to be freely-floating. The G-peak pres-1103 sure coefficient changed at around 2 GPa, from 5.4 $\rm cm^{-1}/GPa$ to 7.5 $\rm cm^{-1}/GPa$. This 1104 was interpreted as resulting from the adhesion to the more compressible solid PTM, with 1105 the strain transmission effect as for supported graphene. In Tao *et al.*¹³¹ bilayer graphene 1106 suspended on Au grid was sandwiched between ice in different phases. It was not possible 1107 to fit the G-band data evolution using the quadratic term of graphite determined by Han-1108 fland et al.²⁸ Tao et al. report a linear G-band evolution with a low pressure coefficient of 1100 $3.4 \text{ cm}^{-1}/\text{GPa}$ up to pressures of 40 GPa^{131} without any noticeable variation at the PTM 1110 freezing point. Filintoglou et al.³⁷ observed graphene sandwiched in solid fluorinert and 1111 obtained a G-band pressure slope of 5.4 $\rm cm^{-1}/GPa$. These results, with G-band pressure 1112 coefficients between 3.4 and 7.5 $\rm cm^{-1}/GPa$, show the extreme sensitivity of graphene to the 1113 nature of the PTM, and the details of the experiment influencing the transmitted strain. 1114 Solidification has a drastic impact for suspended graphene, contrary to the previous case, 1115 supported graphene, for which the solidification has almost no noticeable effect. The high-1116 pressure mechanical response of graphene can be related to graphene composites. In partic-1117 ular, and not surprisingly, it has been shown that the level of adhesion between graphene 1118 and polymer matrix is a key factor in the mechanical response of nanocomposites.^{156,157} 1119

After solidification (crystallization or vitrification), the main issue the nature of the ap-1120 plied stress that includes shear components. However, one may consider the effect of the 1121 increasing pressure on the intrinsic mechanical properties of graphene, especially the c_{33} elas-1122 tic constant. Increasing interaction between π -orbitals and the surrounding solid media can 1123 strongly modify the π -orbitals and hence perhaps the sp² bonds. A detailed quantification 1124 will be introduced in Sec. VIC. This seems to be especially true of polar PTM that ex-1125 hibit piezo-doping and ultimately the formation of covalent bonds.¹³⁴ In this case, graphene 1126 cannot be considered still as an isolated system under external perturbation. 1127

1128 VI. DRESSED GRAPHENE

1129 A. Functionalisation

Graphene can react with, for instance, oxygen, hydrogen or fluorine. When covalent bonds 1130 develop with other molecules in low proportions we may speak about defective graphene. 1131 When those bonds develop extensively we will prefer to speak about other materials, some 1132 aperiodic, such as graphene oxide, some periodic, such as graphane (hydrogenated graphene 1133 monolayer) or diamane (hydrogenated bilayer of graphene)¹⁵⁸ (see Fig. 7) or fluorinated 1134 single-layer diamond.¹⁵⁹ In these cases there will be either local or extended modifications of 1135 the covalent graphene bonding scheme, with a modification of the associated elastic stiffness 1136 constants. For single-layer graphene oxide, the experimental Young's modulus derived from 1137 AFM measurements is substantially reduced to 207.6 ± 23.4 GPa.¹⁶⁰ As most of these other 1138 materials are new, their properties are derived from *ab initio* calculations.¹⁶¹ For instance 1139 it has been shown by DFT calculations that graphene oxide with increasing proportions 1140 of (-O) or (-OH) displays a progressive elongation of the C-C bond and a softening of the 1141 mechanical properties.¹⁶² 1142



FIG. 7. Diamane structure with sp^3 bonding for C atoms.

¹¹⁴³ B. Derivative geometry

We may also consider making a nanotube by rolling up a graphene sheet until we connect the edges through covalent bonds. This change in geometry leads to a topologically different graphene-based system, the carbon nanotube. Of course, the synthesis of carbon nanotubes does not correspond with this *gedanken* experiment, which, rather, raises the issue whether carbon nanotubes should be considered to be a material, derived from graphene by the change in geometry, or to be a structure made of graphene. Many DFT calculations find that the in-plane Young's modulus of small-diameter carbon nanotubes walls is reduced, due to the curvature-induced modification of the C-C hybridization. We may then expect that extensively corrugated, ripple or wrinkled graphene could exhibit an average bonding scheme differing from flat graphene and hence with different local 2D elastic stiffness constants as well as the large-scale reduced stiffness due to the corrugation. See Sec. VII for further discussion of nanotubes.

1156 C. Effect of vdW interactions

Geometry and covalent bonding have an effect on the 2D elastic stiffness constants of 1157 graphene. What about vdW interactions? Consider the difference between graphene in 1158 graphite and in epitaxially grown bilayer graphene, two different cases of vdW graphene 1159 stacking. In graphite the vdW distance between graphene layers is 3.35 Å which may be 1160 considered as the graphene thickness in that particular case (see Sec. II A and Table I). In 1161 bilayer graphene grown on a SiC(0001) surface, the measured graphene-graphene distance is 1162 3.24 Å.¹⁶³ In multilayer graphene grown on an SiC substrate, the distance between graphene 1163 layers is found to be 3.9 Å after the first graphene layer in contact with the SiC substrate.¹⁶⁴ 1164 X-ray diffraction measurements have revealed a certain degree of rotational disorder in the 1165 stacking of these graphene layers.¹¹⁶ Do these differences in thickness imply changes in the 1166 C-C sp2 bonding? 1167

The thickness of graphene in these different cases is to be related to the extension of its π -1168 orbitals. This is certainly a point of view in rupture with the Galilean continuum mechanics 1169 approach, but wholly consistent with the modern approach to the radius of atoms. In 1170 vacuum, of course, quantum mechanics tells us that the spatial extension of the π -orbitals 1171 electron cloud is to infinity, like the hydrogen 1s state. Any definition of a finite extension 1172 of an electron orbital in vacuum (such as the Bohr radius) is thus entirely arbitrary. Indeed 1173 such definitions are better described as characteristic lengths (such as the Bohr radius) which 1174 are not arbitrary, but also not obviously the extension or size of the atom. Where the pi-1175 orbitals of a graphene monolayer are delimited by meeting the electronic orbitals of adjacent 1176 materials, their extension is to be defined in just the same way as the size of atoms is defined 1177 (see Sec. II A and V).) In the case of graphite, it is very simple - the distance to the point 1178

between the graphene layers about which the π -electron density is symmetrical. In the case 1179 of a graphene monolayer with other materials either side, or multilayer graphene in other 1180 than Bernal stacking, we need to seek criteria (as with the size of atoms in multi-element 1181 mixtures and compounds) which allow the consistent attribution of a thickness to, e.g. the 1182 side of a graphene monolayer in contact with another layer in AA or in turbostratic contact, 1183 and also to the other side in contact with perhaps sapphire. We should certainly define the 1184 graphene thickness in the asymmetric context as the addition of two different contributions 1185 on each side of the carbon nuclei. 1186

Practically, the shift of the G-mode frequency under out-of-plane compression is a suitable 1187 quantity to quantify the weak modification of the in-plane elastic constants by deformation 1188 of π -orbitals. The G-mode, as mentioned previously, is an in-plane anti-phase vibration 1189 of C-C atom pairs and is therefore closely related to the in-plane stiffness of graphene and 1190 graphite. Its eigenvectors (E_{2g}) in graphene and graphite are shown in Fig. 8. The dynamical 1192 equation of a 1D spring can be written as $Ku = \omega^2 u$, where K is the force constant, u is 1193 the displacement and ω is the frequency. And it can be extended to 2D for the G-mode of 1194 graphene:¹⁶⁶ 1195

$$\begin{pmatrix} \omega_0^2 & 0\\ 0 & \omega_0^2 \end{pmatrix} \begin{pmatrix} u1\\ u2 \end{pmatrix} = \omega^2 \begin{pmatrix} u1\\ u2 \end{pmatrix}, \tag{11}$$

where the u_1 and u_2 are the relative displacement of the two carbon atoms along the two equivalent in-plane directions, as the hexagonal lattice of graphene is isotropic in-plane. When an additional graphene layer is added, Eq. 11 becomes¹⁶⁷

$$\begin{pmatrix} \omega_0^2 & 0 & C & 0 \\ 0 & \omega_0^2 & 0 & C \\ C & 0 & \omega_0^2 & 0 \\ 0 & C & 0 & \omega_0^2 \end{pmatrix} \begin{pmatrix} u1 \\ u2 \\ u3 \\ u4 \end{pmatrix} = \omega^2 \begin{pmatrix} u1 \\ u2 \\ u3 \\ u4 \end{pmatrix},$$
(12)

where u_3 and u_4 are the displacement of the two carbon atoms in the added layer, and C_{1200} accounts for the interlayer coupling. The longitudinal and transverse modes are not coupled due to the hexagonal lattice, hence all the zero elements in the force constant tensor. The solutions to the secular equation of Eq. 12 are



FIG. 8. "Phonon eigenvectors of graphene and graphite. Every phonon eigenvector of graphene gives rise to two vibrations of graphite. For example, the in-phase combination of the two layers for the E_{2g} optical mode of graphene yields $E_{2g} \otimes A_{1g} = E_{2g}$ and the out-of-phase combination $E_{2g} \otimes B_{1u} = E_{1u}$. Next to the graphite modes it is indicated whether they are Raman (R) or infrared (IR) active and the experimentally observed phonon frequencies. The translations of graphite are omitted from the figure." Reproduced with permission from Phil. Trans. Roy. Soc. A **362**, 2271 (2004).¹⁶⁵ Copyright 2004 The Royal Society.

$$\omega_{(1)}^{2} = \omega_{0}^{2} + C$$
$$\omega_{(2)}^{2} = \omega_{0}^{2} + C$$
$$\omega_{(3)}^{2} = \omega_{0}^{2} - C$$
$$\omega_{(4)}^{2} = \omega_{0}^{2} - C$$

The two different solutions correspond to the E_{1u} and E_{2g} G-modes of graphite, where carbon atoms in both layers in a unit cell vibrate in-plane and in anti-phase, but the vibrations of the two layers are in-phase and out-of-phase, respectively. Typical experimentally measured values of the graphite E_{1u} and E_{2g} frequencies are 1587 and 1580 cm⁻¹, respectively.^{27,168} From the measured frequencies of E_{1u} and E_{2g} of graphite, we can calculate the $\omega_0=1583.5$ cm^{-1} for the G-mode of a graphene plane in graphite – the G-mode frequency of graphene should be slightly higher than graphite, even their in-plane stiffnesses are the same.

To quantify the effect of deformation of the π -orbitals on the G-mode frequency, we can introduce out-of-plane strain and calculate the shift of G-mode frequency. The off-diagonal term C in Eq. 12 for interlayer coupling can be expanded in terms of out-of-plane strain ε_{zz} . The diagonal terms can be expanded too, to account for the possible modification of the in-plane sp² bond stiffness by the compression of the π -orbitals,

$$\begin{pmatrix} \omega_0^2 + A\varepsilon_{zz} & 0 & C + B\varepsilon_{zz} & 0 \\ 0 & \omega_0^2 + A\varepsilon_{zz} & 0 & C + B\varepsilon_{zz} \\ C + B\varepsilon_{zz} & 0 & \omega_0^2 + A\varepsilon_{zz} & 0 \\ 0 & C + B\varepsilon_{zz} & 0 & \omega_0^2 + A\varepsilon_{zz} \end{pmatrix} \begin{pmatrix} u1 \\ u2 \\ u3 \\ u4 \end{pmatrix} = \omega^2 \begin{pmatrix} u1 \\ u2 \\ u3 \\ u4 \end{pmatrix}$$
(13)

¹²¹⁵ and the solution to the secular equation is:

$$\omega_{(1)}^{2} = \omega_{0}^{2}(E_{2g}^{(2)}) + (A+B) \times \varepsilon_{zz}$$
$$\omega_{(2)}^{2} = \omega_{0}^{2}(E_{2g}^{(2)}) + (A+B) \times \varepsilon_{zz}$$
$$\omega_{(3)}^{2} = \omega_{0}^{2}(E_{1u}) + (A-B) \times \varepsilon_{zz}$$
$$\omega_{(4)}^{2} = \omega_{0}^{2}(E_{1u}) + (A-B) \times \varepsilon_{zz}$$

which indicates an increasing E_{1u} and E_{2g} splitting, and therefore different shift rates with pressure of these two modes. Without the contribution from the coupling to adjacent layers, the G-mode frequency of graphene under out-of-plane compression is $\omega = \omega_0 + A \times \varepsilon_{zz}$. Sun *et al.* calculated the G-mode frequencies of graphite under out-of-plane compression by DFT, and introduced a new parameter γ' to compare the contribution from out-of-plane strain on with that from in-plane strain:¹⁶⁷

$$\frac{\Delta\omega}{\omega_0} = -\gamma(\varepsilon_{xx} + \varepsilon_{yy}) \mp \frac{1}{2}SDP(\varepsilon_{xx} - \varepsilon_{yy}) - \gamma'\varepsilon_{zz}$$
(14)

where γ is the Grüuneisen parameter and SDP is the shear deformation potential. The values of γ' are -0.0131 and 0.0585 for E_{2g} (GM) and E_{1u} of graphite, respectively, and the value for the GM of graphene is in the middle. These values, compared to a typical value 1.90 of the Grüuneisen parameter for the in-plane strain contribution,^{33,62,166} are indeed very small. However, both graphene and graphite are about 30 times more compressible out-ofplane than in-plane.²⁰ So, under hydrostatic compression, the contribution to the G-mode frequency from out-of-plane compression, while smaller than the in-plane contribution, is not negligible.

Three useful points can be summarised from the above discussion: first, deformation 1230 of π -orbitals can modify the in-plane bond stiffness by a non-negligible amount, especially 1231 when the vdW interaction between graphene and the surrounding medium increases under 1232 compression; second, although the G-mode is a good measure of in-plane stiffness, its fre-1233 quencies in graphene and graphite are slightly different even if the in-plane stiffnesses are 1234 the same; and third, the shift rate of the monolayer graphene G-mode with pressure should 1235 be higher than that graphite, because for the graphene G-mode, there is no impact from the 1236 coupled vibration in adjacent layers. 1237

Returning to the issue of different vdW interaactions with different media at ambient pressure, some authors have proposed that the graphene layers in multilayer graphene on 4H-SiC(0001) behave as monolayer graphene¹¹⁶ due to the predominant rotational stacking faults weakening the graphene-graphene interaction. Hence, following the same principle of thickness definition, the flat turbostratic (free-standing) graphene thickness will be 3.9 Å, *i.e.* ~6.5 % greater than the vdW graphite distance.

We may then conclude that different schemes of vdW stacking lead to changes in the 1244 π -orbitals. If we need to consider effects on the 2D elastic stiffness constants of the π -1245 orbitals, then they should be considered as affected by graphene stacking schemes. The 1246 effect of such changes on the sp² bonds are from deformed π -orbitals modifying the in-plane 1247 C-C sp² bonds. The weak modifications of the C-C sp² bonds by the π -orbitals can be 1248 quantified in detail as above. In particular, as graphene is little affected by weak van der 1249 Waals interactions or geometry variations with curvatures less than about $\sim 1 \text{ nm}^{-1}$, the 1250 electronic structure bonding scheme determining the 2D elastic constants of graphene is 1251 preserved. Flat or weakly bent graphene in vacuum or in graphite or in single-wall and 1252 multi-wall nanotubes may be then considered as having the same 2D elastic constants as 1253 graphite. 1254

1255 VII. CARBON NANOTUBES: PROPERTIES OF GRAPHENE

¹²⁵⁶ Carbon nanotubes, particularly single-walled and double-walled carbon nanotubes, are ¹²⁵⁷ interesting structures in their own right (but that is outside the scope of this review). They

provide perhaps the only way in which graphene can be studied free-standing and in vacuum 1258 - graphene with nothing touching either side and nothing to constrain it in-plane either. 1259 Closed nanotubes provide the opportunity to study graphene with vacuum inside and other 1260 media outside, while open-ended nanotubes provide other possibilities. Most important, and 1261 our focus here, they can reveal aspects of the mechanical properties of graphene, such as its 1262 bending stiffness, that are difficult or impossible to study in other forms of graphene. Many 1263 of the opportunities to learn about graphene by studying nanotubes have not yet been fully 1264 exploited, as we shall see below. 1265

Most of the work we discuss depends on Raman spectroscopy, observing the G-mode and the other graphene/graphite phonons, but crucially the radial breathing mode (RBM) which has no equivalent in graphite.

1269 A. G-mode in nanotubes

The G-mode has the inconvenience that it is hard to resolve the contributions of nanotubes of different diameters and chiralities. For that reason, much more attention has been paid to the RBM (Sec. VII B).

The G-mode frequency is sensitive to confinement effects, dynamical effects, and curva-1273 ture, which lift the degeneracy to give G^+ and G^- peaks.¹⁶⁹ In Piscanec *et al*,¹⁶⁹ the effect 1274 of curvature is deduced from the difference between experimental phonon wavenumbers and 1275 calculations and obeys $\Delta \omega = -\zeta \times d^{-2}$. Values are given for the TO mode (circumferential 1276 motion) with $\zeta = 25.16 \text{ cm}^{-1} \cdot \text{nm}^2$ and LO modes (axial motion) with $\zeta = 12.0 \text{ cm}^{-1} \cdot \text{nm}^2$. 1277 Up to now, there is no direct calculation of the curvature effect. However, it can be esti-1278 mated from a simple continuum model. We consider the thickness of the graphene h_G and 1279 treat the electrons on both sides (*i* for inner and o for outer) in a first approximation as a 1280 continuous medium. The tension (o) and compression (i) are opposite but not equal, giving 1281 a strain at the centre. Energy equilibrium gives $\epsilon_i^2 \times (d/2 - h_G/4) = \epsilon_o^2 \times (d/2 + h_G/4)$. 1282 For a curved plate, we have: $\epsilon_o = h_G/2d$. So we deduce the average strain for the center 1283 to be $\langle \epsilon \rangle = (\epsilon_i + \epsilon_o)/2 = h_G^2/(8d^2)$. Using the strain coefficient $-57 \text{ cm}^{-1}/\%$, ¹⁷⁰ we find 1284 $\Delta \omega_{TO} = -80.0 \times d^{-2}$. This gives the correct dependence on d but too large a value for ζ . 1285 However, a homogenous medium is not a satisfactory model for graphene (c.f. the Yacobson 1286 paradox, Sec. II A). Refining the model as a structure, (σ bonds at the centre, and π for 1287

the inner and outer material), the relation is the same but the coefficient is different. From energy equilibrium, we have $\epsilon_{\sigma} = \sqrt{\frac{c_{11}^{\sigma}(h_G - h_{\sigma})}{c_{11}^{\pi}h_{\sigma}}} \times \epsilon_{\pi} = \sqrt{\frac{c_{11}^{\sigma}(h_G - h_{\sigma})}{c_{11}^{\pi}h_{\sigma}}} \times \epsilon_{\pi} = 0.170 \times \epsilon_{\pi}$ leading to $\Delta \omega_{TO} = -13.6 \times d^{-2}$, which is in a more reasonable range and $\epsilon_{\sigma} = \frac{2.38 \times 10^{-3}}{d^2}$, to be compared with $\frac{1.6 \times 10^{-3}}{d^2}$ from calculation.¹⁷¹

1292 B. RBM in nanotubes

In the RBM fundamental, all atoms move radially together. The restoring force is straightforwardly due to c_{11} , and the frequency depends inversely on the diameter, and to a much lesser extent on the chirality, in the 100-300 cm⁻¹ region of the spectrum. Moreover, tunable excitation picks out those tubes that are resonant with the excitation wavelength. Consequently, very detailed studies of the RBM have been reported.¹⁷²

Simple models can account for this mode using continuous mechanics or atomic descrip-1298 tions. We start by supposing the nanotube wall to consist of a 2D sheet of continuum 1299 material with the 2D graphene elastic constants $c_{11}^{2D} = c_{11}d = 372 \text{ Nm}^{-1}$ and $c_{12}^{2D} = c_{12}d =$ 1300 47 Nm^{-1} . In the RBM motion, the wall has tangential strain but no axial strain (the RBM 130 frequency is too high to induce any axial motion), so the relevant elastic stiffness constant 1302 is c_{11}^{2D} . In contrast, the approach of Mahan,¹⁷³ modelling with a three-dimensional isotropic 1303 plate, invokes not only c_{11} but also c_{12} , which is incorrect. The potential energy per unit 1304 length of tube at the extreme of a sinusoidal motion $r = A \cos \omega t$ is 1305

$$U_{max} = \frac{1}{2}c_{11}^{2D}\epsilon^2 \times 2\pi R_C = \pi R_C c_{11}^{2D} \frac{A^2}{R_C^2}$$
(15)

¹³⁰⁶ while the kinetic energy at the center of the motion is

$$E_{max} = \frac{1}{2}mA^{2}\omega^{2} = \frac{1}{2}A^{2}\omega^{2} \times 2\pi R_{C}Nm_{0}$$
(16)

where $N = 3.8 \times 10^{19}$ is the number of carbon atoms of mass m_0 in a unit area of graphene. Equating U_{max} and E_{max} , and rearranging, we have

$$\omega_{RBM} = \frac{1}{R_C} \sqrt{\frac{k_{11}}{Nm_0}} \equiv \frac{235}{d(nm)} \mathrm{cm}^{-1}$$
(17)

where the diameter $d = 2R_C$, in excellent agreement with experiments.¹⁷⁴ At much higher frequency, 1590 cm⁻¹, the G-mode phonon corresponds to atomic motion in antiphase. To relate the RBM to the G-mode considering atomic motion, a 1D model (atoms equispaced around a circle) has been proposed by Gerber *et al.*¹⁷⁵ leading to :

$$\omega_{RBM} = \frac{a_0}{d} \omega_G = \frac{0.142 \text{nm} \times 1590 \text{cm}^{-1}}{d(\text{nm})} = \frac{226}{d(\text{nm})} \text{cm}^{-1}$$
(18)

¹³¹³ in excellent agreement with Eq. 17. Considering a 2D system, the equations are the same ¹³¹⁴ because the E_{2g} G-mode is doubly degenerate in the plane allowing the basis to be aligned ¹³¹⁵ with the chiral vector, which is the circumference. The approximation here is that the ¹³¹⁶ G-mode degeneracy is not lifted by the curvature.

For nanotubes in a medium (e.g. a liquid or nanotube bundles) the vdW interaction leads to an upshift of the RBM frequencies above the values of Eq. 17 by some 10 - 20 cm⁻¹. The stiffening of the vdW interaction under pressure is largely responsible for the further increase of RBM frequencies under pressure.¹³⁶

The phonon spectrum of nanotubes includes also the soft modes, in the $10 - 100 \text{ cm}^{-1}$ 1321 spectral region, which are the higher-order modes of a series in which the RBM is the zeroth 1322 member. Unlike the RBM, the soft modes depend on the bending stiffness D. The n^{th} soft 1323 mode has 2n + 2 nodes around the circumference of the tube. They should soften under 1324 pressure, and would go to zero frequency at the collapse pressure. However, like the RBM, 1325 they are raised in frequency by the vdW interaction with the PTM, and the increase in this 1326 interaction with pressure actually results in the soft modes stiffening instead of softening 1327 under pressure.¹⁷⁶ 1328

1329 C. SWCNTs under pressure

The pressure dependence of single-walled nanotubes provides two opportunities at least 1330 to learn about graphene. The diameters of nanotubes are usually given as defined by the 1331 nuclear positions, in contrast to taking the outside diameter over the electron orbitals. That 1332 is what the general formula $d = a_0 \sqrt{n^2 + mn + m^2}$ gives, and that is the diameter usually 1333 considered when analysing the response of nanotubes to high pressure, for example, the rate 1334 of shift of the G-mode and RBM phonon frequencies under pressure. If the pressure P were 1335 applied at the radius r = d/2, pressure coefficients of, for example, the G-mode phonon would 1336 be expected to be approximately r/h times the graphite or graphene pressure coefficients, 1337 where h is the relevant graphene thickness (see Sec. II A). (A full analysis would take into 1338

account the differing tangential and axial stresses in a tube under pressure, respectively Pr1339 and Pr/2.) However, if the graphene has a thickness h and the pressure is applied at a 1340 radius r + h/2, the stresses on the sp² bonds will be greater and the pressure coefficients 1341 correspondingly increased. How much they are increased, however, depends critically on the 1342 mechanical properties of graphene and their response to pressure and bending. We are not 1343 aware of a full analysis along these lines of the nanotube pressure coefficients. The situation 1344 is further confused by the stiffening of the RBM mode which is largely due to the increasing 1345 vdW interaction between the PTM and the nanotube,¹³⁶ and also by any effects of the PTM 1346 on the graphene as discussed in the previous section, which may account for the different 1347 pressure coefficients reported for nanotubes in different PTM.^{177–180} 1348

¹³⁴⁹ D. SWCNT Collapse

At sufficiently high pressures or large diameters, nanotubes collapse. Collapse of nan-1350 otubes under high pressure has also been challenging, not least because, apparently as ob-1351 vious a case of Euler buckling as the collapse of pillars, the collapse of tubes under external 1352 hydrostatic pressure is mathematically intractable. Many experimental observations have 1353 been interpreted as corresponding to collapse, at a very wide range of pressures, and fitted to 1354 a variety of theoretical equations. A complete solution for the simple (ideal) elastic ring was 1355 reported only in 2011.¹⁸¹ Good agreement with the Levy-Carrier formula for a thin-walled 1356 tube, $P_C = \frac{3D}{B^3}$ where D is the bending stiffness was confirmed, and the collapse to a peanut 1357 shape above P_C was found to be quite slow, complete only at about $1.5P_C$.¹⁸² Torres-Dias 1358 et al. reported that this fitted experimental data for the quenching of the RBM for a range 1359 of SWCNT diameters, giving an estimate of D = 1.7 eV.⁴⁵ 1360

Some caveats must be mentioned. First, the effects of the thickness h need to be known 1361 and taken into account, as for the pressure coefficients. Also, it is clear that vdW inter-1362 actions will reduce the collapse pressure, so that SWCNTs tubes above about 4 nm will 1363 spontaneously collapse (the same physics as the folding of Sec. III D 6). This scarcely affects 1364 smaller tubes where the bending energies involved are very much greater. Then, for diam-1365 eters below about 1 nm, Torres-Dias *et al.*⁴⁵ reported that the collapse pressure is reduced 1366 below the Levy-Carrier formula. This is quite a strong effect, observed experimentally and 1367 in theoretical modelling,⁴⁵ and also previously noticed by Elliott $et \ al.^{183}$ It extrapolates to 1368

 $P_C = 0$ for a diameter of about 0.4 nm, not much smaller than the smallest nanotubes ever reported.¹⁸⁴ One source of this behaviour could be softening of the bending potential with angle. Another source is the reduction in the Euler buckling load even of straight pillars when the compliance is discretised rather than continuous. This effect is reported by Carter et al,⁷⁷ and discussed in Sec. III D 5. Both of these explanations remain to be analysed in detail.

When the diameter of a SWCNTs is large enough, *i.e.* above about 5 nm, a spontaneous 1375 collapse occurs. This leads to a cross-section in the form of a dogbone or peanut with a 1376 twist along the axis¹⁸⁵ if the SWCNT is free (in liquid or gas for example). The cavities of 1377 the edges have a diameter of the order of C_{60} fullerene, like the cavities of folded graphene 1378 (Section IIID6) while the stacking depends on the chirality.¹⁸⁶ Indeed, the phenomenonis 1379 very closely related to folding, with the same balance between adhesion energy and bending 1380 stiffness. Like folding, it has not been fully exploited to refine our knowledge of these two 1381 important mechanical parameters of graphene. 1382

Del Grande *et al.*¹⁸⁷ note that the energy barrier for a circular tube to collapse is many eV, 1383 so it should not be possible for it to happen through thermal activation – but once initiated 1384 at one point in the tube, the collapse will readily propagate along the tube. This is not an 1385 uncommon situation in condensed matter physics. It may be compared with the initiation 1386 of plastic deformation in a perfect crystal, where the activation energy for the creation of 1387 a dislocation is very high, or with boiling in a pure liquid, where the activation barrier for 1388 the formation of a bubble is very high. In these examples as in doubtless many others, it is 1389 a local defect, impurity, or perturbation that breaks the impasse. Del Grande *et al.* suggest 1390 that the collapse of large SWCNTs is likely to be produced by small mechanical stresses 1391 that naturally occur during synthesis. The nanoscale force required to initiate collapse, to 1392 bypass the energy barrier, is about 5 nN^{187} which is easily achievable in AFM compression 1393 experiments.¹⁸⁸ 1394

1395 E. DWCNTs coefficients under pressure

While the mechanical behaviour of SWCNTs under pressure is reasonably well understood as described in the previous section, double-walled nanotubes provide further opportunities - and challenges - to better understand the mechanical properties of graphene. Again, these

opportunities come from the pressure coefficients of the phonon modes while the tubes 1399 remain circular, and then from the collapse pressures and modes of collapse. Many papers 1400 report pressure coefficients of the Raman G-modes of the outer and inner walls of DWCNTs 1401 which are not dissimilar. Yet it is hard to understand how the external pressure may be 1402 transmitted to the inner tube, given the enormous anisotropy of the graphite elastic stiffness 1403 tensor. Moreover, the sum of the reported pressure coefficients of the inner and outer tube 1404 is usually considerably in excess of the coefficient of an empty outer tube, *i.e.* an SWCNT of 1405 the same diameter. Yet the load on the walls of the inner and out tubes should sum to the 1406 load of an SNCNT of the same diameter, and so therefore should the pressure coefficients. 1407

Experimentally, pressure transmitted to an inner nanotube can be monitored by the 1408 upshift of the GM or RBM. The former should be less dependent on the PTM as the upshift 1409 is from the C-C bond stiffening under pressure, whereas the latter is from the increasing 1410 interaction of a tube with its surroundings. Consequently, it is easier to describe the GM 141 pressure coefficient. The experimental challenge to monitor pressure by the GM frequency 1412 is the assignment of the GM to tubes of a specific diameter and chirality, and, further, to 1413 distinguish inner and outer tubes in DWCNTs. While the RBM is diameter-dependent, 1414 tubes of different diameters have very similar GM frequency, if not the same, at ambient 1415 condition. An ideal situation would be to have only one RBM and its corresponding G-mode 1416 dominating the spectrum (either due to a special sample containing only one chirality, or 1417 having only one chirality in resonance at a specific laser excitation). Many factors can add to 1418 the complication of the situation: 1) common CNT samples contain tubes of many different 1419 chiralities; 2) many more than one chirality can be in resonance or close to the resonance 1420 condition; 3) while the outgoing laser energy is only shifted by 10 - 20 meV for the RBM, 1421 the difference is 200 meV for the GM, making it possible that the GM is in resonance with 1422 the outgoing laser while its corresponding RBM is far from the resonance condition; 4) 1423 for DWCNTs, one has to further assign Raman peaks to the outer or the inner tube, and 1424 the interaction between inner and outer tubes modifies the Kataura plot,¹⁷². In particular, 1425 Hirschmann et al.¹⁸⁹ showed that the wall-to-wall distance between inner and outer tubes in 1426 DWCNTs increases with increasing tube diameters, which makes the RBM upshifts of the 1427 inner tubes from intertube interaction no longer a constant, as most earlier work supposed. 1428 This requires further caution on the assignment of RBMs to inner tubes, but can be used 1429 to refine our calculations if needed. 1430

Early studies on DWCNTs under pressure observed at least two components in a GM 1431 profile, shifting with pressure at different rates. It is tempting to assign these two components 1432 to outer and inner tubes for two reasons, one is that stress transmitted to the inner tube 1433 should be lower than hydrostatic pressure, resulting in two different responses to pressure, 1434 and the other is that outer (or inner) tubes in resonance at the same condition can have very 1435 close diameters and they should response similarly to pressure. Among various work, the 1436 GM pressure coefficients can be different, as different tubes are in resonance; they can be 1437 either PTM dependent, as PTM modifies the transition energy of CNTs, making different 1438 tubes in resonance, or PTM independent as no other charility in those samples is available 1439 near the resonance condition. The results of these high pressure study on DWCNTs are 1440 summarised in Table V below. 1441

Outer tube	Inner tube	PTM	Laser Excitation	Reference and notes
$(cm^{-1}GPa^{-1}) (cm^{-1}GPa^{-1})$			(nm)	
9.6	6.4	paraffin oil	514	Ref. 190
8.4	5.5	NaCl	514	Ref. 190
5.8	3.3	methanol-ethanol	633	Ref. 191
6.9	4.1	Oxygen	633	Ref. 191
8.6	5.1	Argon	633	Ref. 191
5.5	4.3	methanol-ethanol	514	Ref. 192

TABLE V. Experimental shift rates of DWCNTs GM with pressure.

1442 1443

In this section we have seen that much could be learned about the mechanical proper-1444 ties of graphene from further experimental and theoretical work on nanotubes, particularly 1445 under pressure. The major obstacle is that the RBM modes are highly resonant and the 1446 resonances shift with pressure. For the G-mode, this means that as pressure increases, dif-1447 ferent diameter tubes (which are not well-resolved) may dominate the Raman spectrum at 1448 different pressures.¹⁹³ For the RBM the tubes of different diameters and chiralities are well-1449 resolved. However, both for identifying or for choosing which tubes are observed, and for 1450 tracking given tubes over a substantial pressure range, tunable Raman excitation is needed 1451 (e.q. a dye laser or Ti-sapphire laser), together with a tunable Raman spectrometer. The 1452 number of laboratories worldwide with such equipment, and also a high-pressure capability, 1453

1455 VIII. MODELS FOR THE MECHANICS OF GRAPHENE

Mechanical properties, more perhaps than any other properties of matter, invite the 1456 construction of models, for purposes ranging from visualisation, through understanding, to 1457 prediction. We comment briefly here on what can be suitably expressed by or learnt from 1458 different models, starting from the simplest. Continuum models for graphene have been 1459 considered. The flat plate of isotropic material and a thickness chosen to give the right 1460 bending stiffness was mentioned in Section II A. While useful for considering the behavious 1461 of graphene as a beam, plate or shell, the model does not attempt to replicate the thickness 1462 a_{33} or the out-of-plane compressibility c_{33} of graphite. For that reason a continuum model 1463 was considered in which the nuclei and the sp^2 bonding orbitals were treated as an infinitely 1464 thin sheet with the 2D c_{11} and c_{12} values of graphene, sandwiched between two layers of soft 1465 material modelling the pi-orbitals.³⁹ This model could replicate the thickness, the bending 1466 stiffness and the out-of-plane compressibility by a suitable choice of an anisotropic 3D elastic 1467 tensor for this soft material. The model is suitable for considering the behaviour of, for 1468 example, nanotubes under pressure (see above, Sec. VIIC). For example, it can be used to 1469 consider questions such as the radius at which a nanotube is loaded by external pressure. It 1470 is also a model that can be readily discretised as a ball-and-spring model. In this case, the 1471 nuclei are the balls, the sp^2 bonds are springs (2D stretching potentials) between them and 1472 the pi-orbitals are springs that terminate on a point that is not an atom. These points could 1473 also be joined by springs that give the bending stiffness.³⁹ That introduces the question, 1474 whether this is the physically-realistic representation of the origin of the bending stiffness of 1475 graphene, or whether it actually arises from the torsional stiffness of the sp^2 bonds through 1476 4-atom potentials in-plane. A comparison with the torsional vibrational modes of ethane 1477 (sp^3) and ethylene (sp^2) could be useful here, 1478

At this point we are approaching the valence force field (VFF, Keating) models¹⁹⁴ and those used in MD and Monte Carlo simulation. Keating, however, used only two-atom (stretch) and three-atom (angular) springs, or interactions, with nearest neighbours only to model the elastic constants c_{ij} and internal strains of, *e.g.* silicon. Keating did not consider phonon frequencies. It is in considering the phonons as well as the elastic constants that ¹⁴⁸⁴ Keating models tend to break down.

In MD simulation, Ref. 26 used two-atom interactions with atoms out to the fifth-nearest 1485 neighbours to model the phonon frequencies and the phonon dispersions. The necessity 1486 to include the fifth-nearest neighbours is demonstrated by investigating the origin of the 1487 GM frequency. Consideration of the 2D bulk modulus, or area modulus A, gives the force 1488 constant of the C-C bonds as 748 N/m, or 46.7 eV/A², from the experimental values of elastic 1480 constants of graphite.²⁰. We assume that the GM frequency comes only from the nearest C-C 1490 stretching, and we can obtain the GM frequency as 1450 cm^{-1} . The gap to the experimental 1491 value of about 1580 cm^{-1} can be filled by other contributions beyond the nearest neighbour. 1492 Quantifying these contributions requires more information than the frequency of LO at Γ 1493 point (G-Mode)²⁷ in the phonon dispersion relation of graphite. We have already given an 1494 example that one has to include the second nearest neighbour out-of-plane interaction to 1495 describe the separation of E_{1u} and E_{2q} . It was found that up to the fourth nearest neighbour 1496 interaction has to be included to well fit the dispersion from Γ point to M (especially the 1497 initial increase of E_{1u} frequency from Γ point),⁴¹ obtained by inelastic neutron scattering.⁴². 1498 A further fifth nearest neighbour interaction was included to fit more recent in-elastic x-ray 1499 data, which gave a finer description of the local minimum of TO at K point.²⁶ Surprisingly, 1500 the empirical force constant model including up to the fifth nearest neighbour that fits well 1501 the experimental data of the full phonon dispersion of graphite (as shown in Fig. 9), gives a 1502 force constant for the nearest neighbour C-C stretching, as small as $25.88 \text{ eV}/\text{A}^2$, indicating 1503 that about half of the contributions to the C-C vibrational frequency in graphite come from 1504 other interactions than the nearest C-C stretching. This is truly unexpected, yet explaining 1505 the existing data best. 1506

¹⁵⁰⁷ Despite contributing only half to the G-mode frequency, the nearest C-C stretching is ¹⁵⁰⁸ expected to contribute dominantly to the upshift of G-mode with pressure from Pauli exclu-¹⁵⁰⁹ sion, which is a measure of C-C bond anharmonicity. The evolution of the phonon dispersion ¹⁵¹⁰ relation in graphite would not only verify or dispute the small value of the nearest C-C force ¹⁵¹¹ constant, but also quantify the anharmonicity, when it becomes available. We should point ¹⁵¹² out that in addition to in-plane contributions, the GM frequency can also be modified by ¹⁵¹³ deformed π -orbitals from out-of-plane, as discussed in section VIB.

It is clear that Keating potentials are not generally capable of representing both the elastic constants and the phonon frequencies, and certainly fail in this regard for graphene.



FIG. 9. Phonon dispersion of graphite from inelastic x-ray scattering (symbols). Triangles are from Ref. 26. Circles are from Ref. 195. Squares are INS data from Ref. 42. Solid lines are the fit of experimental data by a force constant model including up to the fifth-nearest neighbor. The dashed line is a quadratic extrapolation of the data. Reproduced with permission from Phys. Rev. B 76, 035439 (2007).²⁶ Copyright (2007) The American Physical Society.

It is not clear from the literature what the simplest models are that can do this, nor if such a model is physically interpretable (two-atom interactions between fifth-nearest neighbours do not have any obvious physical interpretation). On the other hand, VFF models continue to be developed and exploited for various purposes, *e.g.* thermomechanical properties¹⁹⁶ and phonon frequencies.¹⁹⁷

Usually, molecular dynamics modelling uses optimized reactive potentials (multi-body potentials such as the second generation reactive empirical bond order (REBO) potential of Brenner¹⁹⁸ or the recent implementation of the ReaxFF potential¹⁹⁹) which allow the study of mechanical properties at the nanoscale such as fracture. An example is the finding that fracture requires a force of about 8 nN per C=C bond.²⁰⁰ This is equivalent to about 90 GPa uniaxial stress and compares with the 130 GPa strength reported by Lee *et al.*²⁵ in AFM nanoindentation (see also Fig. 3 (a)).

¹⁵²⁸ Turning to quantum-mechanical models, the simple tight-binding description is used for

¹⁵²⁹ obtaining the electronic structure²⁰¹ but is not suitable for mechanical properties.

¹⁵³⁰ DFT provides models that can be made to replicate experimental data excellently.²² ¹⁵³¹ However, apart from the maps of electron density, there is little in DFT output that can ¹⁵³² assist a physical understanding of the predicted properties. It could be said that in graphene ¹⁵³³ research as in high-pressure research, the greatest value of DFT is that it can tell us what ¹⁵³⁴ happens under experimental conditions that are not (yet) accessible to experimenters. For ¹⁵³⁵ that reason we discuss it under different headings above. An example in graphite is the ¹⁵³⁶ determination of c_{13} (Sec. III A, see also Sec. IV F).

1537 IX. CONCLUSIONS

This paper reviews the mechanical properties of graphene, both those that are expected 1538 to be similar to graphite and those expected to be different from graphite – and anomalies. 1539 Graphene is commonly called a 2D material, which implies a thickness tending to zero. 1540 However, the π -electrons above and below the 2D plane of carbon nuclei extend the electron 1541 density of monolayer graphene into the third dimension, perpendicular to the 2D plane. For 1542 example, we can define a vdW thickness of graphene, 3.35 Å, which is the experimentally 1543 measured spacing of graphene layers in graphite. One key conclusion is that, far from being 1544 a 2D material, graphene has a well-defined 3D structure, which may be modelled in various 1545 ways to help understand its mechanical properties. That is not to say that it cannot display 1546 2D physics, much as can a 100 Å quantum well – which has a 3D physical structure of e.q.1547 GaAs sandwiched between GaAlAs. Following from that, those of its mechanical properties 1548 which are related to those of graphite are indeed very similar, if not identical. 1549

Without neighboring layers, unsupported graphene is not mechanically stable and has intrinsic ripples. The low bending stiffness further promotes the formation of ripples, making them common in graphene samples. It also contributes to the softening of the ZA phonon dispersion, resulting in a negative thermal expansion (although, again, it is not clear whether this is significantly different from that of graphite). Properties such as the out-of-plane stiffness, though expected to be similar to graphite, require indirect approaches to define and to quantify.

¹⁵⁵⁷ Due to the small sample size of exfoliated graphene – at least out-of-plane – experiments ¹⁵⁵⁸ to measure many of its mechanical properties requires special design. In addition, the environment surrounding graphene adds further complexity to the interpretation of these experimental data, from determining factors as substrates transferring strain to graphene, to subtle modification by influencing the π -orbital distribution.

There are many derivative structures from graphene, in a way making the extraordinary properties of graphene tunable. They can also be used to help understand the properties of graphene. Among those, measurements on carbon nanotubes in some circumstances give the most accurate values for mechanical properties of graphene, perhaps even better than measurements on graphene itself, as nanotubes can be self-supporting, free-standing, and stable, thus excluding many of those complexities.

Finally, composite materials in which matrix material is reinforced mechanically by the inclusion of graphene flakes are perhaps one of the most exciting applications in which the mechanical properties of graphene are central. Many other so-called 2D materials are also used, and their mechanical properties are often less well characterised than those of graphite. One may expect them to be related to the corresponding bulk materials much as graphene is related to graphite.

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1578 DATA AVAILABILITY

Data sharing is not applicable to this article as no new data were created or analyzed in this review.

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