

# 1 Acoustics of multiscale sorptive porous materials

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This paper investigates sound propagation in multiscale rigid-frame porous materials that support mass transfer processes, such as sorption and different types of diffusion, in addition to the usual visco-thermo-inertial interactions. The two-scale asymptotic expansion method of homogenization for periodic media is successively used to derive the macroscopic equations describing sound propagation through the material. This allowed us to conclude that the macroscopic mass balance is significantly modified by sorption, inter-scale (micro- to/from nanopore scales) mass diffusion, and inter-scale (pore to/from micro- and nanopore scales) pressure diffusion. This modification is accounted for by the dynamic compressibility of the effective saturating fluid that presents atypical properties that lead to slower speed of sound and higher sound attenuation, particularly at low frequencies. Contrarily, it is shown that the physical processes occurring at the micro-nano scale do not affect the macroscopic fluid flow through the material. The developed theory is exemplified by introducing an analytical model for multiscale sorptive granular materials that is experimentally validated by comparing its predictions with acoustic measurements on granular activated carbon. Furthermore, we provide empirical evidence supporting an alternative method for measuring sorption and mass diffusion properties of multiscale sorptive materials using sound waves.

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## 9 I. INTRODUCTION

10 Sound propagation in sorptive porous materials with multiple scales of heterogeneities,  
11 i.e. multiscale sorptive porous materials, is investigated in this paper. Sorption is a general  
12 term used to refer to adsorption, desorption, and absorption (penetration of the fluid into  
13 the solid phase). The former is a physical or chemical process in which the fluid molecules  
14 are adhered on to a surface. Adsorption can also be understood as an increase of fluid  
15 density in the vicinity of a fluid-solid interface. Desorption is the opposite phenomenon,  
16 i.e. the fluid molecules are released from a surface. The molecules adherence in physical  
17 adsorption is caused by weak van der Waals forces, while their release by either an increase  
18 of temperature or a decrease in pressure which leads to a break of the weak physical bond<sup>1</sup>.  
19 Adsorption/desorption is accompanied by mass diffusion that governs the flux of molecules  
20 from a region of higher concentration to one of lower concentration<sup>1</sup>.

21 The microstructure of multiscale sorptive materials features pores or inclusions of very  
22 dissimilar characteristic sizes ranging from nanometers to millimeters. An example of this  
23 type of materials is a packing of porous grains in which the grains themselves feature two  
24 scales of heterogeneities, i.e. a triple porosity sorptive material. It will be demonstrated  
25 in this paper that their distinctive characteristic is the simultaneous occurrence of sorption  
26 processes at the smallest scale, different inter-scale diffusion processes, and visco-thermal  
27 effects at different scales. Materials of interest that possess hierarchical microstructure and  
28 support the mentioned physical processes include activated carbons<sup>2</sup>, zeolites<sup>3</sup>, and metal-  
29 organic frameworks<sup>4</sup>, for example, in granular and pellet form.

30 The investigation of mass transport and sorption in multiscale porous materials is of  
31 interest, for example, in geophysics and gas engineering. For instance, in Ref. 5 the solute  
32 transport in fractured sorptive porous media was investigated. In Ref. 6, gas filtration in  
33 porous coal and the effect of gas constrained in nanometric pores was studied with the aim of  
34 understanding the physical mechanisms leading to coal-gas outbursts in coal mines. In Ref.  
35 7, a dynamic model for mass transfer in coal seams with application to CO<sub>2</sub> sequestration  
36 was investigated, while Ref. 8 dealt with the behavior of gas flow in multi-porosity shale  
37 gas reservoirs.

38 On the other hand, multiscale sorptive porous materials are widely used in chemical  
39 engineering applications such as filtration, gas storage, and catalysis, among others<sup>1-4,9</sup>. For

40 these applications, it is of interest to assess the sorption and diffusion properties of the  
41 materials. A measurement method, called frequency response<sup>10-12</sup>, has been used to this  
42 end. This method aims at measuring the mass diffusion and sorption parameters of porous  
43 materials and is based on periodically perturbing the equilibrium of a system. For example,  
44 in a batch system it is normally considered a slow periodic change in volume of a container  
45 in which the sorptive material is placed. This change in volume leads to a slow periodic  
46 change in pressure that is recorded and further used to obtain the material parameters  
47 by fitting a theoretical model to the data. Of these models, the ones described in Refs.  
48 13–15 for bidispersed structured sorbents are relevant to the present work. Their common  
49 features are: (i) the mass transport in both pore networks is modeled as a Fickian diffusion  
50 process, (ii) equilibrium between fluid and sorbed phases in the pores and linear isotherms are  
51 considered, and (iii) they are usually applied to describe diffusion and sorption in granular  
52 materials made by agglomerating porous microparticles (or crystals). In particular, the  
53 justification of (i) is the experimental condition normally used in the frequency response  
54 method: the measurements are taken at low pressures. At normal conditions, the mass  
55 transport in the pores formed in between millimeter-size inclusions is not of diffusive but  
56 advective type. This has been accounted for in Refs. 16 and 17 where sound propagation  
57 in a slit pore formed between two infinite nanoporous sorptive plates and in an array of  
58 cylindrical pores embedded in a nanoporous sorptive matrix were theoretically investigated,  
59 respectively. These works aimed at extending the working frequency range of the frequency  
60 response method and the structures studied can be considered as single-pore and double  
61 porosity sorptive materials.

62 In acoustics, the influence of sorption on sound propagation in a single tube has been  
63 investigated in Refs. 18 and 19. The main conclusion in Ref. 19 is that the contribution  
64 to sound energy dissipation due to viscosity, heat transfer, and mass diffusion are additive.  
65 Experimental work on the acoustical properties of granular activated carbon (GAC) has  
66 evidenced that partially filling a loudspeaker enclosure<sup>20</sup> or a Helmholtz resonator cavity<sup>21</sup>  
67 with GAC leads to an increase of their effective compliance. In addition, it has been shown  
68 that rigidly-backed layers of GAC display unusually large low frequency sound absorption  
69 coefficient<sup>21-23</sup>. The characteristic feature of activated carbons is that the low-frequency  
70 effective compressibility of the saturating gas attains values larger than the isothermal one  
71 predicted by the current theory of acoustics of porous media<sup>23</sup>. It was suggested in Ref. 23

72 that this behavior may be explained by considering an additional scale to the double porosity  
73 model introduced therein, as well as by accounting for sorption processes. This idea was  
74 developed further in Ref. 24 where rarefied gas flow in pores with size comparable to the  
75 molecular mean free path and sorption effects in nanopores were included into a model for  
76 sound propagation in granular activated carbon. One of the limitations of this model is that  
77 the inter-scale (inner-grain micro to/from nano pores) mass diffusion was assumed quasi-  
78 static. Hence, the model cannot be used to assess the influence of dynamic inter-scale mass  
79 diffusion on the acoustical properties of triple porosity sorptive materials. Furthermore,  
80 since this model was introduced in a phenomenological manner, its range of validity is not  
81 clearly identified. Recently, an upscaled model for sound propagation in double porosity  
82 sorptive materials has been developed in Ref. 25. This model cannot be used to describe  
83 the acoustical properties of triple porosity sorptive materials. The limitations of these two  
84 works are overcome in this paper. The aim is to describe the acoustical properties of a wide  
85 class of sorptive porous materials.

86 The first objective of this paper is to present a rigorous derivation of the macroscopic  
87 description of sound propagation in multiscale sorptive porous materials by making succes-  
88 sive use of the two-scale asymptotic method of homogenization for periodic media<sup>26,27</sup>. The  
89 application of this method leads to an upscaled model whose range of validity is clearly  
90 identified. The upscaled model accounts for viscosity and heat transfer effects at the pore  
91 scale, rarefied gas flow and heat transfer at the micropore scale, inter-scale (pore to/from  
92 micro-nanopore scales) pressure diffusion, inter-scale (micro- to/from nanopore scales) mass  
93 diffusion, and sorption at the nanopore scale. The developed theory applies to materials  
94 saturated with a pure gas. Typical examples may be the system nitrogen/granular acti-  
95 vated carbon, zeolites or metal-organic frameworks, which could approximate the acoustic  
96 behavior of this type of materials saturated with air.

97 The second objective of this paper is to determine the combined influence of sorption,  
98 rarefaction, and dynamic inter-scale diffusion processes on sound propagation through triple  
99 porosity sorptive materials. Crucially, it is demonstrated that sorption effects occurring in  
100 pores of nanometer size strongly modify the macroscopic mass balance. This modification is  
101 accounted for by the compressibility of the effective saturating fluid, which displays uncon-  
102 ventional properties that result in a slower speed of sound and higher attenuation of sound  
103 in the material. The strength of these macroscopic effects in the audible frequency range

104 largely depends on the dynamic pressure and mass diffusion properties of the material. This  
105 cannot be properly assessed using the models developed in Refs. 24 and 25 due to their  
106 limitations discussed above. Contrarily to sorption effects, it is shown that the effects of rar-  
107 efied gas flow in pores with size comparable to the molecular mean free path only intervene  
108 in the macroscopic acoustic behavior indirectly via a modification of the apparent pressure  
109 diffusivity. Furthermore, it is proven that, under the conditions established by homogeniza-  
110 tion and in coherence with our previous studies<sup>23,24</sup>, the constitutive fluid flow law and its  
111 associated effective parameter, i.e. the dynamic viscous permeability, are not modified by  
112 the physical processes occurring in the micro and nano pores.

113 The strong combined influence of sorption and mass diffusion on the acoustical properties  
114 of the materials is measurable. Hence the third objective of this paper corresponds to  
115 the use of acoustic measurements on macroscopic samples to deduce physical parameters  
116 characterizing sorption, that occurs at the nanoscale, and the effective diffusivity determining  
117 the inter-scale mass diffusion. This may provide a simple alternative procedure to measure  
118 the sorption and diffusion properties of multiscale sorptive materials. We provide empirical  
119 evidence supporting this claim and validate the developed theory experimentally.

120 The paper is organized as follows. The macroscopic description of sound propagation in  
121 multiscale sorptive porous materials is presented in Section II. The analysis of the effective  
122 parameters of the upscaled model follows. An analytical model for sound propagation in  
123 multiscale sorptive granular material is introduced in Section IV. This is further used in  
124 Section V to exemplify and experimentally validate the theory. The main findings are  
125 summarized in the conclusions.

## 126 **II. SOUND PROPAGATION IN MULTISCALE SORPTIVE POROUS** 127 **MATERIALS - THEORY**

128 This section deals with the derivation of the macroscopic equations that describe sound  
129 propagation in multiscale sorptive porous materials. The upscaling is done using the two-  
130 scale asymptotic method of homogenization for periodic media<sup>26,27</sup>. The material geometry  
131 and the main assumptions regarding its morphology are discussed first. We then focus on  
132 the case of a multiscale sorptive material with well separated macro, meso, micro and nano  
133 scales. This allows applying the upscaling method to the set of equations that describes

134 the acoustic behavior in the pore fluid network, where viscosity and heat transfer effects  
 135 take place, and in the micro-nano porous domain. The latter is modeled as an equivalent  
 136 continuum and accounts for viscosity and heat transfer effects at the micropore scale, inter-  
 137 scale mass diffusion characterized by the micropore geometry and two diffusion processes  
 138 (i.e. bulk and surface diffusion) occurring in the nanopores, and sorption on the walls of the  
 139 nanopores. The effective equations governing sound propagation in the micro-nano porous  
 140 domain have been derived in Ref. 25. We present in Appendix A the main steps of the  
 141 derivation as well as its extension to account for rarefied gas flow and heat transfer in pores  
 142 with size comparable to the molecular mean free path.

## 143 A. Geometry

144 Consider a periodic multiscale sorptive rigid-frame porous material saturated with a pure  
 145 Newtonian fluid. Figure 1 shows a diagram of the scales of the material and the relevant  
 146 geometrical descriptors. The macroscopic characteristic length  $L$  is related to the sound  
 147 wavelength  $\lambda$  through  $L = |\lambda|/2\pi$  and strongly exceeds all other characteristic lengths of  
 148 the material. The representative elementary volume (REV) of the material is denoted as  $\Omega_p$ .  
 149 This is constituted by the volume of the pores  $\Omega_{pf}$  and the volume of the micro-nano porous  
 150 domain  $\Omega_{mn}$ , i.e.  $\Omega_p = \Omega_{pf} \cup \Omega_{mn}$ . The solid part of  $\Omega_{mn}$  is assumed perfectly impervious  
 151 to gas transport. The surface of the pores is denoted as  $\Gamma_p$ . The micro-nano porous domain  
 152  $\Omega_{mn}$  has a REV  $\Omega_m$  that comprises the volume of the micropores  $\Omega_{mf}$  and that of the nano  
 153 porous domain  $\Omega_n$ , i.e.  $\Omega_m = \Omega_{mf} \cup \Omega_n$ . The surface of the micropores is denoted as  $\Gamma_m$ .  
 154 In turn, the nano porous domain is composed of the volume of the nanopores  $\Omega_{nf}$  and the  
 155 volume of its impervious solid part. The surface of the nanopores is represented by  $\Gamma_n$ .

157 The characteristic length associated with the pore (or the period of the material), mi-  
 158 cropore, and nanopore scales are denoted as  $l_p$ ,  $l_m$ , and  $l_n$ , respectively. These are well  
 159 separated, i.e.  $l_n \ll l_m \ll l_p$ , and for the materials of interest are usually millimetric,  
 160 micrometric, and nanometric in size (see e.g. Ref. 9).

161 Because of the separation between the pore and micropore scales, i.e.  $l_m/l_p \ll 1$ , the  
 162 micro-nano porous domain  $\Omega_{mn}$  can be modeled as an equivalent continuum<sup>28</sup>. Similarly,  
 163 since the characteristic size associated with the nanopore scale is much smaller than that  
 164 associated with the micropore scale, i.e.  $l_n/l_m \ll 1$ , the nano porous domain can be

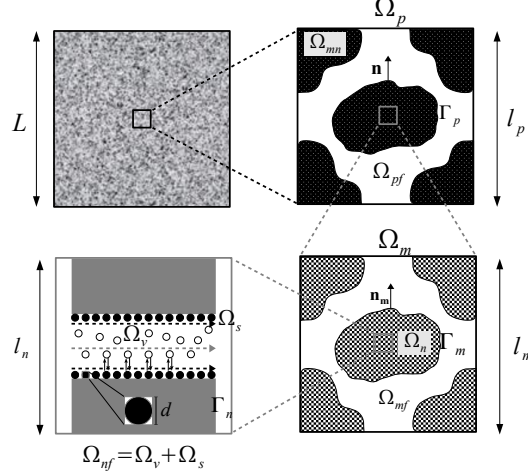


FIG. 1. Diagram of the scales of a multiscale sorptive porous material.

165 considered as an equivalent continuum governed by effective equations that are defined in  
 166 the whole domain  $\Omega_n$  and reflect the following local physical processes<sup>25</sup>: sorption occurring  
 167 on the walls of the nanopores, volumetric diffusion of free gas molecules in the bulk of the  
 168 nanopores, and surface diffusion of the adsorbed molecules. The diffusion processes are  
 169 respectively represented in Figure 1 by the horizontal dashed gray and black lines, while  
 170 the mass exchange between the gas (hollow circles) and adsorbate (black circles) phases is  
 171 depicted with vertical lines with arrows. The adsorbate volume is estimated by  $\Omega_s = |\Gamma_n|Nd$ ,  
 172 where  $N$  is the number of adsorbed layers and  $d$  is the diameter of the molecules. These are  
 173 respectively assumed equal to unity (i.e.  $N = 1$ , monolayer coverage) and smaller than the  
 174 nanopore characteristic size (i.e.  $d < l_n$ ). The void space available for the transport of free  
 175 molecules is represented by  $\Omega_v$ . Hence one has that  $\Omega_{nf} = \Omega_v \cup \Omega_s$ .

176 The porosity of the material is  $\phi_{pmn} = \phi_p + (1 - \phi_p)(\phi_m + (1 - \phi_m)\phi_n)$ , where  $\phi_p = \Omega_{pf}/\Omega_p$ ,  
 177  $\phi_m = \Omega_{mf}/\Omega_m$ , and  $\phi_n = \Omega_{nf}/\Omega_n$  are the porosities associated with the pores, micropores,  
 178 and nanopores, respectively. In turn, the porosity of the pores and micropores is  $\phi_{pm} =$   
 179  $\phi_p + (1 - \phi_p)\phi_m$ , while that of the micro-nano porous domain is  $\phi_{mn} = \phi_m + (1 - \phi_m)\phi_n$ .

180 The disparity in length scales between the pore size and the macroscopic characteristic  
 181 size associated with the acoustic phenomenon provides a small parameter  $\epsilon = l_p/L \ll 1$ .

## 182 B. Governing equations at the pore scale

183 The equations describing sound propagation in the pore fluid network are<sup>29–32</sup>: the lin-  
 184 earized equations of conservation of momentum (1), mass (2), and energy (3); and the  
 185 equation of state (4). The physical parameters involved are the dynamic viscosity  $\eta$ , spe-  
 186 cific heat capacity  $C_p$ , thermal conductivity  $\kappa$ , and equilibrium pressure  $P_0$ , density  $\rho_0$ , and  
 187 temperature  $\tau_0$ . The oscillating velocity, pressure, density, and temperature are denoted as  
 188  $\mathbf{u}_p$ ,  $p_p$ ,  $\rho_p$ , and  $\tau_p$ , respectively. Note that harmonic dependence of the type  $e^{j\omega t}$  is assumed  
 189 and, because of linearity, this term is omitted throughout the paper.

$$\eta \nabla^2 \mathbf{u}_p - \nabla p_p = j\omega \rho_0 \mathbf{u}_p \quad \text{in } \Omega_{pf}, \quad (1)$$

$$j\omega \rho_p + \rho_0 \nabla \cdot \mathbf{u}_p = 0 \quad \text{in } \Omega_{pf}, \quad (2)$$

$$\kappa \nabla \cdot \nabla \tau_p = j\omega C_p \rho_0 \tau_p - j\omega p_p \quad \text{in } \Omega_{pf}, \quad (3)$$

$$\frac{p_p}{P_0} = \frac{\rho_p}{\rho_0} + \frac{\tau_p}{\tau_0} \quad \text{in } \Omega_{pf}. \quad (4)$$

193 The equations governing sound propagation in the micro-nano porous domain, which is  
 194 modeled as an equivalent continuum, have been derived in Ref. 25 for the case of negligible  
 195 rarefaction effects in the micropores. We extend this model in Appendix A to account for  
 196 these effects. The derived effective equation of conservation of mass (5) and dynamic Darcy's  
 197 law (6) are given by:

$$\nabla \cdot \mathbf{U}_m + j\omega p_m \mathbf{C}_{mn}(\omega) = 0 \quad \text{in } \Omega_{mn}, \quad (5)$$

$$\mathbf{U}_m = -\frac{\mathbf{k}_m(\omega)}{\eta} \cdot \nabla p_m \quad \text{in } \Omega_{mn}, \quad (6)$$

199 where  $p_m$  is pressure in the micropores,  $\mathbf{U}_m$  is the Darcy's velocity,  $\mathbf{k}_m(\omega)$  is the dynamic  
 200 viscous permeability associated with the micropores and  $\mathbf{C}_{mn}(\omega)$  is the compressibility of the  
 201 effective fluid saturating the micro-nano porous domain (hence the subscript  $mn$ ). Explicit  
 202 expressions for the latter two parameters will be given further below.

203 Eqs. (1)-(6) are coupled via the following conditions on the pore boundary  $\Gamma_p$  expressing  
 204 the continuity of normal mass flux (7) and pressure (8), and of negligible temperature  
 205 variations (9). Note that  $\mathbf{n}$  is the outward-pointing normal vector (see Figure 1).

$$\rho_0 \mathbf{u}_p \cdot \mathbf{n} = \rho_0 \mathbf{U}_m \cdot \mathbf{n} \quad \text{on } \Gamma_p, \quad (7)$$

$$p_p = p_m \quad \text{on } \Gamma_p, \quad (8)$$



$$\tau_p = 0 \quad \text{on} \quad \Gamma_p. \quad (9)$$

208 Together with the boundary condition (7), it is set that the tangential mass flux is zero,  
 209 i.e.  $\rho_0(\mathbf{u}_p - (\mathbf{u}_p \cdot \mathbf{n})\mathbf{n}) = \mathbf{0}$ . In addition, by considering the continuity of heat flux and its  
 210 estimation<sup>33</sup>  $\kappa\tau_p/l_p = \kappa\tau_m/l_m$ , one obtains  $\tau_m = l_m\tau_p/l_p$ . Then, the variation of temperature  
 211 in the pores is much larger than that in the micropores, i.e.  $\tau_p \gg \tau_m$ , and the condition  
 212 Eq. (9) is thereby justified.

### 213 C. Homogenization procedure

214 The scale separation between the pore and macroscopic characteristic sizes, i.e.  $l_p/L =$   
 215  $\epsilon \ll 1$ , allows us to use the two-scale asymptotic expansion method of homogenization for  
 216 periodic media to derive an equivalent macroscopic model. To represent the evolution at  
 217 the two spatial scales, the following dimensionless space variables  $\mathbf{x}/L = \mathbf{x}^*$  and  $\mathbf{x}/l_p = \mathbf{y}^*$   
 218 are introduced, where  $\mathbf{x}$  stands for the usual space variable. These dimensionless space  
 219 variables are associated with the variations at the local and macroscopic scales, respectively.  
 220 Equivalently, taking  $L$  as reference length, we will use the following two dimensional space  
 221 variables  $\mathbf{x} = L\mathbf{x}^*$  and  $\mathbf{y} = L\mathbf{y}^* = \mathbf{x}L/l_p = \epsilon^{-1}\mathbf{x}$ . Then, the usual gradient operator  $\nabla$  is  
 222 given by  $\nabla_{(xy)} = \nabla_x + \epsilon^{-1}\nabla_y$  (and  $\nabla_{(xy)}^2 = \nabla_x^2 + 2\epsilon^{-1}\nabla_{xy} + \epsilon^{-2}\nabla_y^2$ ). Note that for simplicity  
 223 in the notation, we have used non-bold letters for the spatial variables.

224 The use of two space variables should be combined with a rescaling of the usual equations  
 225 based upon a single space variable. The reason for the rescaling lies in the fact that when  
 226 expressed with the two space variables  $(x, y)$ , the actual physical gradient of a quantity  $Q$   
 227 that varies at the macroscopic scale, i.e.  $\nabla_x Q$ , becomes  $\nabla_{(xy)} Q$ . Similarly, if the quantity  
 228 varies at the local scale, the actual physical gradient  $\nabla_y Q$  reads  $\epsilon\nabla_{(xy)} Q$ . Therefore, the  
 229 gradient of variables oscillating at the local scale should be rescaled. In other words, to  
 230 formulate the set of rescaled equations governing sound propagation at the pore scale one  
 231 should analyze at which scale the physical quantities fluctuate as well as the relative order  
 232 of magnitude of the terms in the governing equations. This analysis for the variables and  
 233 terms in Eqs. (1)-(4) is well established. The arguments and procedure can be found, for  
 234 example, in Refs. 27 and 30 and are now recalled.

235 In the long-wavelength regime the pore pressure fluctuates at the macroscopic scale, i.e.

236  $|\nabla p_p| = O(p_p/L)$  and, while the fluid velocity and its rate of deviatoric deformation vary at  
 237 the pore scale (i.e.  $|\eta\nabla^2\mathbf{u}_p| = O(\eta u_p/l_p^2)$ ), the microscopic divergence itself is of the order  
 238 of the macroscopic divergence, i.e.  $|\nabla \cdot \mathbf{u}_p| = O(u_p/L)$ . Note that, for example,  $u_p$  is an  
 239 estimation of  $|\mathbf{u}_p|$ . On the other hand, the temperature varies at the pore scale, which leads  
 240 to the following estimation  $|\kappa\nabla^2\tau_p| = O(\kappa\tau_p/l_p^2)$ . Regarding the relative order of magnitude,  
 241 we are interested in the case when the viscous and inertial terms in the oscillatory Stokes  
 242 equation (1) balance the pressure gradient. Hence the estimations of the three terms in Eq.  
 243 (1) are of the same order of magnitude, i.e.  $O(\eta u_p/l_p^2) = O(\omega\rho_0 u_p) = O(p_p/L)$ . Similarly, the  
 244 estimations of the three terms in the equation of conservation of energy (3) are of the same  
 245 order of magnitude, i.e.  $O(\kappa\tau_p/l_p^2) = O(\omega\rho_0 C_p \tau_p) = O(\omega p_p)$ . Furthermore, the estimation of  
 246 the terms in the equations of conservation of mass and of state satisfy  $O(u_p/L) = O(\omega\rho_p/\rho_0)$   
 247 and  $O(p_p/P_0) = O(\rho_p/\rho_0) = O(\tau_p/\tau_0)$ , respectively.

248 In the micro-nano porous domain, both the Darcy's velocity  $\mathbf{U}_m$  and the micropore  
 249 pressure  $p_m$  fluctuate at the pore scale. In addition, the estimations of the terms in the  
 250 mass balance equation (5), as well as those in the dynamic Darcy's law Eq. (6)), are of the  
 251 same order of magnitude:  $O(U_m/l_p) = O(\omega p_m C_{mn})$  and  $O(U_m) = O(\mathcal{K}_m p_m/\eta l_p)$ .

252 Regarding the boundary conditions, the continuity of pressure on the pore boundaries  
 253  $\Gamma_p$  sets  $O(p_p) = O(p_m)$  while the long-wavelength condition imposes that the advective  
 254 mass flux pulsed from the micro-nano porous domain on  $\Gamma_p$  is of one order smaller than the  
 255 advective mass flux generated by the incident wave in the pores, i.e.

$$\mathcal{U} = \frac{|\rho_0 \mathbf{U}_m \cdot \mathbf{n}|}{|\rho_0 \mathbf{u}_p \cdot \mathbf{n}|} = O(\epsilon). \quad (10)$$

256 This estimate is justified by the following argument. Consider a cell  $\Omega$  and denote the  
 257 ingoing mass flux on one face (of surface  $S$ ) as  $S\rho_0 u_{p1}$ , the outgoing mass flux on the  
 258 opposite face as  $S\rho_0 u_{p2}$ , and the mass flux pulsed from the micro-nano porous domain  $\Omega_{mn}$   
 259 as  $\rho_0 U_m \Gamma_p$ . By hypothesis, a regime of long wavelength  $L \gg l$  is considered. Thus,  
 260  $(S\rho_0 u_{p2} - S\rho_0 u_{p1})/S\rho_0 u_{p1} \approx l_p/L$ . Since from conservation of mass  $S\rho_0 u_{p2} \approx S\rho_0 u_{p1} +$   
 261  $\Gamma_p \rho_0 U_m$ , it follows that  $\Gamma_p \rho_0 U_m / S\rho_0 u_{p1} \approx l_p/L = \epsilon$ .

262 In terms of physical parameters, the ratio between the mass fluxes on the pore boundary  
 263  $\Gamma_p$  can be written as:

$$\mathcal{U} = \frac{|\rho_0 \mathbf{U}_m \cdot \mathbf{n}|}{|\rho_0 \mathbf{u}_p \cdot \mathbf{n}|} = \frac{\mathcal{K}_m p_m}{\eta l_p} \frac{\eta L}{l_p^2 p_p} = \frac{\mathcal{K}_m L}{l_p^2} \frac{L}{l_p} = \frac{l_m^2}{l_p^2} \epsilon^{-1} = O(\epsilon). \quad (11)$$

264 Note that i) the continuity of pressure on  $\Gamma_p$  (i.e. Eq. (8)) has been used and ii) the  
 265 estimation (11) holds as long as characteristic sizes associated with the pore and micropore  
 266 scales are well separated, i.e.  $l_m/l_p = O(\epsilon)$ . Analogously, this condition can be expressed in  
 267 terms of the viscous permeabilities associated with the pore and micropore fluid networks  
 268 as  $\mathcal{K}_m/\mathcal{K}_p = O(\epsilon^2)$ , i.e. the permeabilities are highly contrasted. Furthermore, the mass  
 269 flux ratio estimation is consistent with the modeling of the micro-nano porous domain as an  
 270 equivalent continuum.

271 The rescaled set of equations describing sound propagation at the pore scale is then given  
 272 by Eqs. (12)-(19). Note that we adopt the usual homogenization convention that consists in  
 273 keeping the same notation as for the single-space-variable formulation for both the variables  
 274 and the gradient operator. For example,  $\nabla$  and  $\mathbf{u}_p$  stand for  $\nabla_{(xy)}$  and  $\mathbf{u}_p(x, y)$ , respectively.

$$\epsilon^2 \eta \nabla^2 \mathbf{u}_p - \nabla p_p = j\omega \rho_0 \mathbf{u}_p \quad \text{in } \Omega_{pf}, \quad (12)$$

$$j\omega \left( \frac{p_p}{P_0} - \frac{\tau_p}{\tau_0} \right) + \nabla \cdot \mathbf{u}_p = 0 \quad \text{in } \Omega_{pf}, \quad (13)$$

$$\epsilon^2 \kappa \nabla \cdot \nabla \tau_p = j\omega C_p \rho_0 \tau_p - j\omega p_p \quad \text{in } \Omega_{pf}, \quad (14)$$

$$\epsilon \nabla \cdot \mathbf{U}_m + j\omega p_m C_{mn} = 0 \quad \text{in } \Omega_{mn}, \quad (15)$$

$$\mathbf{U}_m = -\frac{\mathbf{k}_m}{\eta} \cdot \epsilon \nabla p_m \quad \text{in } \Omega_{mn}. \quad (16)$$

$$\rho_0 \mathbf{u}_p \cdot \mathbf{n} = \epsilon \rho_0 \mathbf{U}_m \cdot \mathbf{n} \quad \text{on } \Gamma_p, \quad (17)$$

$$p_m = p_p \quad \text{on } \Gamma_p, \quad (18)$$

$$\tau_p = 0 \quad \text{on } \Gamma_p. \quad (19)$$

282 The physical variables are then looked for in the form of asymptotic expansions in powers of  
 283 the small parameter  $\epsilon = l_p/L$  as  $Q(x, y) = \sum_{i=0}^{\infty} \epsilon^i Q^{(i)}(x, y)$  where  $Q = p_p, \mathbf{u}_p, \tau_p, \rho_p, p_m, \mathbf{U}_m$ .  
 284 These are then replaced in Eqs. (12)-(19) and the terms of the same order are identified.  
 285 This leads to cell problems whose solutions are used in conjunction with the leading-order  
 286 mass balance equation spatially averaged over the pore volume to obtain the macroscopic  
 287 equations that describe sound propagation in multiscale sorptive porous materials introduced  
 288 in the next section. The mathematical details of the derivation are presented in Appendix  
 289 B.

290 **D. Macroscopic description of sound propagation in multiscale sorptive**  
 291 **materials**

292 The macroscopic mass balance equation and fluid flow constitutive law determining sound  
 293 propagation in multiscale sorptive porous materials are a key result of this paper and are  
 294 respectively given by Eqs. (20) and (21) (see Appendix B for their derivation).

$$\nabla_x \cdot \langle \mathbf{u}_p^{(0)} \rangle + j\omega p_p^{(0)} \mathbf{C}(\omega) = 0, \quad (20)$$

295

$$\langle \mathbf{u}_p^{(0)} \rangle = -\frac{\mathbf{k}(\omega)}{\eta} \cdot \nabla_x p_p^{(0)}. \quad (21)$$

296 Here the dynamic viscous permeability tensor is given by  $\mathbf{k}(\omega) = \mathbf{k}_p(\omega)$ , with  $\mathbf{k}_p(\omega)$  being the  
 297 dynamic viscous permeability tensor associated with the pore fluid network. The effective  
 298 dynamic compressibility  $\mathbf{C}(\omega)$  is given by Eq. (22) and corresponds to the sum of the  
 299 classical effective dynamic compressibility  $\mathbf{C}_p(\omega)$  (Eq. (23)) that accounts for heat transfer  
 300 in the pores and an additional effective dynamic compressibility  $\mathbf{C}_{mn}(\omega)$  that results from  
 301 the mechanisms of heat transfer in the micropores, inter-scale (micro-nano) mass diffusion,  
 302 and sorption in the nanopores. The contribution of  $\mathbf{C}_{mn}(\omega)$  is weighted by the function  
 303  $\mathcal{F}_{pmn}(\omega)$ , which accounts for inter-scale pressure diffusion effects.

$$\mathbf{C}(\omega) = \mathbf{C}_p(\omega) + (1 - \phi_p) \mathbf{C}_{mn}(\omega) \mathcal{F}_{pmn}(\omega). \quad (22)$$

304 The dynamic compressibility  $\mathbf{C}_p$  of the effective fluid saturating the pore network is given  
 305 by:

$$\mathbf{C}_p(\omega) = \frac{\phi_p}{P_0} \left( 1 - \frac{j\omega}{\omega_{tp}} \frac{\gamma - 1}{\gamma} \frac{\Theta_p(\omega)}{\Theta_{p0}} \right), \quad (23)$$

306 where  $\Theta_p(\omega)$  is the dynamic thermal permeability associated with the pore fluid network and  
 307 the thermal characteristic frequency determining the transition from isothermal to adiabatic  
 308 sound propagation in the pore fluid network is defined as  $\omega_{tp} = \kappa \phi_p / \rho_0 C_p \Theta_{p0}$ , where  $\Theta_{p0} =$   
 309  $\Theta_p(\omega = 0)$  is the static thermal permeability<sup>32</sup>.

310 The function  $\mathcal{F}_{pmn}$  is given by:

$$\mathcal{F}_{pmn}(\omega) = 1 - \frac{j\omega}{\omega_b} \frac{\mathcal{B}_{app0}}{\mathcal{B}_{app}} \frac{B(\omega)}{B_0}, \quad (24)$$

311 where  $B(\omega)$  is the inter-scale pressure diffusion function,  $\mathcal{B}_{app} = \mathcal{K}_m / \eta \mathbf{C}_{mn}$  is the apparent  
 312 pressure diffusivity, and the pressure diffusion characteristic frequency is estimated as  $\omega_b =$

313  $(1 - \phi_p)\mathcal{B}_{app0}/B_0$ . Here  $B_0 = B(\omega = 0)$  is the static value of the inter-scale pressure diffusion  
 314 function,  $\mathcal{B}_{app0} = \mathcal{K}_{m0}/\eta\mathcal{C}_{mn0}$  is the static apparent pressure diffusivity,  $\mathcal{K}_{m0} = \mathcal{K}_m(\omega = 0)$   
 315 is the static viscous permeability associated with the micropore fluid network, and  $\mathcal{C}_{mn0} =$   
 316  $\mathcal{C}_{mn}(\omega = 0)$ .

317 The dynamic compressibility of the effective fluid saturating the micro-nano porous do-  
 318 main  $\mathcal{C}_{mn}(\omega)$  is given by (see Appendix A for its derivation):

$$\mathcal{C}_{mn}(\omega) = \mathcal{C}_m(\omega) + (1 - \phi_m)\mathcal{C}_n\mathcal{F}_{mn}(\omega), \quad (25)$$

319 where the dynamic compressibility of the effective fluid saturating the micropores  $\mathcal{C}_m$  is  
 320 calculated using Eq. (23) with the subscript  $p$  being replaced by  $m$ . The effective compress-  
 321 ibility of the nano porous domain  $\mathcal{C}_n$  and the function  $\mathcal{F}_{mn}$  that accounts for inter-scale  
 322 (micro-nano) mass diffusion are given by:

$$\mathcal{C}_n = \frac{H_e}{P_0}, \quad (26)$$

323

$$\mathcal{F}_{mn}(\omega) = 1 - \frac{j\omega G(\omega)}{\omega_d G_0}, \quad (27)$$

324 where the mass diffusion characteristic frequency is defined as  $\omega_d = (1 - \phi_m)\mathcal{D}_{app}/G_0$ . Here  
 325  $G_0$  is the static value of the dynamic inter-scale mass diffusion  $G(\omega)$  (see Eq. (A.35)),  
 326  $\mathcal{D}_{app} = D_e/H_e$  is the apparent mass diffusivity,  $D_e$  is the effective diffusion coefficient, and  
 327  $H_e$  is the effective linearized sorption equilibrium constant<sup>25</sup>. Note that  $D_e$  can be expressed  
 328 in terms of the diffusion coefficients associated with the volumetric diffusion of free gas  
 329 molecules in the bulk of the nanopores and surface diffusion of the adsorbed molecules on  
 330 the walls of the nanopores (see Eq. (A.5)), while  $H_e$  in terms of the linearized sorption  
 331 equilibrium constant  $H$  (see Eq. (A.6)) and can be related to the parameters of the classical  
 332 Langmuir isotherm model<sup>34</sup> via Eq. (A.37).

333 Further assuming macroscopic isotropy, the dynamic viscous permeability becomes  $\mathbf{k} =$   
 334  $\mathcal{K}\mathbf{I}$ , where  $\mathbf{I}$  is the unitary second-rank tensor. Then, the characteristic impedance  $Z_c$ , wave  
 335 number  $k_c$ , and speed of sound  $\mathcal{C}$  in the material are given by<sup>35</sup> (with  $\mathcal{K} = \mathcal{K}_p$ ):

$$Z_c(\omega) = \sqrt{\frac{\eta}{j\omega\mathcal{K}\mathcal{C}}} \quad , \quad k_c(\omega) = \omega\sqrt{\frac{\eta\mathcal{C}}{j\omega\mathcal{K}}} \quad , \quad \mathcal{C}(\omega) = \frac{\omega}{k_c(\omega)}. \quad (28)$$

336 Quantities that will be used to experimentally validate the theory are the surface  
 337 impedance  $Z_w$  and sound absorption coefficient  $\alpha$  of a rigidly-backed layer of material

338 of thickness  $d_l$ . These are given by:

$$Z_w(\omega) = -jZ_c \cot(k_c d_l), \quad \alpha(\omega) = 1 - \left| \frac{Z_w - Z_0}{Z_w + Z_0} \right|^2 = \frac{4X^*}{(1 + X^*)^2 + (Y^*)^2}. \quad (29)$$

339 where  $Z_0 = \rho_0 c_0$  is the characteristic impedance of the saturating fluid,  $c_0$  its speed of  
 340 sound, and  $X^* = \text{Re}(Z_w(\omega)/Z_0)$  and  $Y^* = \text{Im}(Z_w(\omega)/Z_0)$  are the normalized resistance  
 341 and reactance, respectively.

342 In summary, the derived macroscopic equations (20) and (21) that describe sound prop-  
 343 agation in multiscale sorptive rigid-frame porous materials allows us to conclude that the  
 344 dynamic Darcy's law and the dynamic viscous permeability are not modified by the physical  
 345 processes occurring in the micro-nano porous domain. Only the fluid flow in the pore fluid  
 346 network influences the dynamic permeability values. Conversely, the effective dynamic com-  
 347 pressibility becomes significantly modified by i) inter-scale pressure diffusion (pores to/from  
 348 the micro-nano porous domain), ii) inter-scale mass diffusion (micropores to/from the nano  
 349 porous domain), and iii) sorption occurring on the walls of the nanopores. This modification  
 350 comes from the appearance of a source term in the macroscopic mass balance equation, i.e.  
 351 the third term in Eq. (B.8), that accounts for the contribution of these physical processes.  
 352 Since the quantities in Eqs. (28) and (29) depend on  $C(\omega)$ , these are all modified by the  
 353 physical processes that influence the effective dynamic compressibility.

### 354 III. ANALYSIS OF THE EFFECTIVE PARAMETERS

355 The limiting values of the effective dynamic compressibility  $C(\omega \ll \omega_{\min}) = C_{lf}$  and  
 356  $C(\omega \gg \omega_{\max}) = C_{hf}$ , where  $\omega_{\min} = \min(\omega_{tp}, \omega_d, \omega_b, \omega_{tm})$  and  $\omega_{\max} = \max(\omega_{tp}, \omega_d, \omega_b)$ , are  
 357 an important result of this paper. These are given by (see Appendix C for their derivation):

$$C_{hf}P_0 = \frac{\phi_p}{\gamma}, \quad (30)$$

358

$$C_{lf}P_0 = \phi_p + (1 - \phi_p)(\phi_m + (1 - \phi_m)H_e) - \frac{j\omega}{\omega_c}, \quad (31)$$

359 where  $\omega_c$  is a global characteristic frequency that accounts for the thermal and diffusive  
 360 processes in the material and is defined by:

$$\frac{1}{\omega_c} = \phi_p \frac{\gamma - 1}{\gamma} \frac{1}{\omega_{tp}} + (1 - \phi_p)\phi_m \left( \frac{\gamma - 1}{\gamma} \frac{1}{\omega_{tm}} + \frac{1}{\omega_b} \right) + (1 - \phi_p)(1 - \phi_m)H_e \left( \frac{1}{\omega_b} + \frac{1}{\omega_d} \right). \quad (32)$$

Eq. (30) shows that for frequencies much higher than those characterizing the diffusion mechanisms, the influence of the physical processes occurring in the micro-nano porous domain on the macroscopic effective dynamic compressibility is negligible.

On the right-hand side of Eq. (32), the first term is associated with the effects of heat transfer between the solid frame and the fluid in the pores. The second term accounts for heat transfer between the solid frame and the fluid in the micropores and the influence of inter-scale pressure diffusion. The third term is associated with inter-scale pressure and mass diffusion. Note that sorption modifies both the mass diffusion- and pressure diffusion-related effects via the dependence of the respective characteristic frequencies on the effective linearized sorption equilibrium constant  $H_e$ . By construction one has that  $\omega_{tp} < \omega_{tm}$ ,  $\omega_d < \omega_{tm}$ , and  $\omega_b < \omega_{tm}$ . Depending on the morphologies of the pore fluid network and micro-nano porous domain as well as their associated thermal and apparent mass diffusivities, the mass diffusion characteristic frequency could be either smaller, in the order of, or larger than the thermal characteristic frequency associated with the pore fluid network, i.e. either  $\omega_d < \omega_{tp}$ ,  $\omega_d = O(\omega_{tp})$ , or  $\omega_d > \omega_{tp}$ . Similarly, the same type of sorting relationship can be observed for  $\omega_b$  and  $\omega_{tp}$ . On the other hand, one may observe  $\omega_d = O(\omega_b)$  or  $\omega_d > \omega_b$  for fast diffusing system, while for strongly sorptive gas-solid system presenting slow mass diffusion the inequality  $\omega_d < \omega_b$  is likely to be observed.

Defining the ratio between the effective adsorbate concentration in the nanopores and the effective gas concentration in the pores and micropores as  $M_H = (1 - \phi_p)(1 - \phi_m)H_e/\phi_{pm}$ , it follows from Eq. (31) that the static compressibility is given by:

$$C_0 = C(\omega = 0) = \frac{1}{P_0}(\phi_p + (1 - \phi_p)(\phi_m + (1 - \phi_m)H_e)) = \frac{1}{P_0}\phi_{pm}(1 + M_H). \quad (33)$$

This equation shows that, as a consequence of sorption, the low-frequency effective dynamic compressibility can attain a value substantially larger than that of conventional porous materials.

It is worth highlighting that the direct relation between the effective linearized sorption equilibrium parameter  $H_e$  and the real part of the low-frequency effective dynamic compressibility, given by Eq. (33), allows deducing  $H_e$  from measurements of  $\text{Re}(C(\omega \rightarrow 0))$ . On the other hand, the effective diffusion coefficient  $D_e$  can be measured from measurements of the imaginary part of the effective dynamic compressibility and the use of Eq. (31), provided that the other macroscopic parameters involved are known.

391 In general, the effective dynamic compressibility reduces to that of conventional single  
 392 porosity non-sorptive materials when  $\phi_n = \phi_m = 0$ , i.e.  $\mathbf{C}(\omega) = \mathbf{C}_p(\omega)$ . Hence its limiting  
 393 values are  $P_0\mathbf{C}(\omega \gg \omega_{tp}) = \phi_p/\gamma$  and  $P_0\mathbf{C}(\omega \ll \omega_{tp}) = \phi_p - j\omega/\omega_c$ , with  $1/\omega_c$  being equal  
 394 to the first term on the right-hand side of Eq. (32).

395 In absence of nanopores, i.e.  $\phi_n = 0$  (or  $H_e = 0$ ), the effective dynamic compress-  
 396 ibility of double porosity non-sorptive materials with highly contrasted permeabilities<sup>33,36</sup>  
 397 is retrieved, i.e.  $\mathbf{C}(\omega) = \mathbf{C}_p + (1 - \phi_p)\mathbf{C}_m(\omega)\mathcal{F}_{pm}(\omega)$ . Here  $\mathcal{F}_{pm}$  is calculated using Eq.  
 398 (24) and keeping in mind that the apparent pressure diffusivity becomes  $\mathcal{B}_{app} = \mathcal{B}_{app0} =$   
 399  $\mathcal{K}_{m0}P_0/\phi_m\eta$ . Consequently, the pressure diffusion characteristic frequency is given by  $\omega_b =$   
 400  $(1 - \phi_p)\mathcal{K}_{m0}P_0/\phi_m\eta B_0$  and the limiting values of the effective dynamic compressibility are  
 401  $P_0\mathbf{C}(\omega \ll \omega_{tp}) = \phi_{pm} - j\omega/\omega_c$  and  $P_0\mathbf{C}(\omega \gg \omega_b) = \phi_p/\gamma$ . In this case,  $1/\omega_c$  is given by  
 402 Eq. (32) for  $H_e = 0$ .

403 The case of a material without micropores, i.e.  $\phi_m = 0$ , given the extremely large  
 404 separation between the characteristic sizes associated with the nanopore and pore scales  
 405 considered in this work, is of less interest since the mass flux pulsed from the nano porous  
 406 domain is extremely small in comparison with the mass flux in the pores. Hence the mass  
 407 flux ratio is estimated by  $\mathcal{U} = O(\epsilon^2)$  and the material effectively behaves as a single porosity  
 408 non-sorptive material in the audible frequency range.

409 For a material without mesoscopic pores, i.e.  $\phi_p = 0$ , the effective dynamic compressibility  
 410 reduces to that of double porosity sorptive materials introduced in Ref. 25 (see also Appendix  
 411 A), while the dynamic viscous permeability is given by  $\mathcal{K} = \mathcal{K}_m$ .

412 The case of pressure and mass diffusion occurring without sorption is observed when i) the  
 413 number of adsorbed layers is equal to zero (i.e.  $N = 0$ ), ii) the concentrations of the adsorbed  
 414 and gaseous phases are identical (i.e.  $H = 1$ ), or iii) the characteristic size of the nanopores  
 415 is much larger than the size of the molecules, i.e.  $l_n \gg d$ . In all these situations, the  
 416 effective dynamic compressibility reduces to that of a triple-porosity non-sorptive material  
 417 and is calculated using Eq. (22). The parameters involved take the values:  $D_e = \phi_n D_n$ ,  
 418  $H_e = \phi_n$ , and  $\mathcal{B}_{app} = \mathcal{K}_{m0}P_0/\phi_{mn}\eta$ . Hence, one has that  $P_0\mathbf{C}(\omega \ll \omega_{\min}) = \phi_{pmn} - j\omega/\omega_c$   
 419 and  $P_0\mathbf{C}(\omega \gg \omega_{\max}) = \phi_p/\gamma$ , where  $1/\omega_c$  is given by Eq. (32) with  $H_e = \phi_n$ .

420 As discussed previously, the fluid flow at the leading order remains unaffected by the  
 421 physical processes occurring in the micro-nano porous domain under the conditions estab-  
 422 lished through homogenization (i.e.  $\mathcal{U} = O(\epsilon)$  and  $\mathcal{J} = O(\epsilon)$ , see Eqs. (10) and (A.11)).



423 Hence the properties of the dynamic viscous permeability are the same as those of this  
 424 parameter for single porosity non-sorptive materials. Considering leading-order terms only,  
 425 the dynamic viscous permeability takes the following limiting values<sup>31</sup>:  $\mathcal{K}(\omega \ll \omega_{vp}) = \mathcal{K}_0$   
 426 and  $\mathcal{K}(\omega \gg \omega_{vp}) = -j\phi_p\delta_v^2/\alpha_\infty$ , where  $\omega_{vp} = \phi_p\eta/\rho_0\mathcal{K}_0\alpha_\infty$  is the viscous characteristic  
 427 frequency,  $\delta_v = \sqrt{\eta/\rho_0\omega}$  is the viscous boundary layer thickness, and  $\alpha_\infty$  is the tortuosity.

428 Using the asymptotic values of  $\mathcal{C}(\omega)$  and  $\mathcal{K}(\omega)$ , the following limiting values for the speed  
 429 of sound and wave number are obtained:

$$\begin{aligned} \mathcal{C}(\omega \rightarrow 0) &= \sqrt{\frac{j\omega\mathcal{K}_0}{\eta} \frac{P_0}{\phi_{pm}(1+M_H)}} = \frac{\mathcal{C}_{0\phi_p}}{\sqrt{\frac{\phi_{pm}}{\phi_p}(1+M_H)}} = \frac{\mathcal{C}_{0\phi_{pm}}}{\sqrt{1+M_H}} \quad (34) \\ &= \frac{\mathcal{C}_{0\phi_{pmn}}}{\sqrt{\frac{\phi_{pm}}{\phi_{pmn}}(1+M_H)}}, \\ \mathcal{C}(\omega \rightarrow \infty) &= \mathcal{C}_{\infty\phi_p} = \mathcal{C}_{\infty\phi_{pm}} = \mathcal{C}_{\infty\phi_{pmn}} = \frac{c_0}{\sqrt{\alpha_\infty}}. \end{aligned}$$

430

$$\begin{aligned} k_c(\omega \rightarrow 0) &= \omega \sqrt{\frac{\eta}{j\omega\mathcal{K}_0} \frac{\phi_{pm}}{P_0}(1+M_H)} = k_{c0\phi_p} \sqrt{\frac{\phi_{pm}}{\phi_p}(1+M_H)} = k_{c0\phi_{pm}} \sqrt{1+M_H} \quad (35) \\ &= k_{c0\phi_{pmn}} \sqrt{\frac{\phi_{pm}}{\phi_{pmn}}(1+M_H)}, \\ k_c(\omega \rightarrow \infty) &= \frac{\omega}{\mathcal{C}_{\infty\phi_p}} = \frac{\omega}{\mathcal{C}_{\infty\phi_{pm}}} = \frac{\omega}{\mathcal{C}_{\infty\phi_{pmn}}}. \end{aligned}$$

431 The subscripts  $\phi_p$ ,  $\phi_{pm}$ , and  $\phi_{pmn}$  denote the limiting values for single, double, and triple  
 432 porosity non-sorptive materials respectively. These expressions show that, at low fre-  
 433 quencies, the sound waves are both slowed down and more attenuated by a factor of  
 434  $\sqrt{(\phi_{pm}/\phi_p)(1+M_H)}$ ,  $\sqrt{1+M_H}$ , and  $\sqrt{(\phi_{pm}/\phi_{pmn})(1+M_H)}$  in comparison with single,  
 435 double, and triple porosity non-sorptive materials, respectively. At high frequencies, the  
 436 influence of sorption and the inter-scale diffusion processes vanishes.

437 Although not an effective parameter, it is of interest to investigate the low-frequency  
 438 asymptotic behavior of the surface impedance  $Z_w(\omega \rightarrow 0) = Z_{w0}$  of a rigidly-backed layer of  
 439 multiscale sorptive material of thickness  $d_l$ . Provided that  $|k_c(\omega)d_l| \ll 1$  and  $\omega \ll \omega_{\min}$ ,  
 440 the real and imaginary parts of the low-frequency surface impedance can be approximated by  
 441 (with  $\sigma_{p0} = \eta/\mathcal{K}_{p0}$  being the static flow resistivity and  $\Phi = \phi_p + (1-\phi_p)(\phi_m + (1-\phi_m)H_e) =$   
 442  $\phi_{pm}(1+M_H)$  the apparent porosity of the material):

$$\text{Im}(Z_{w0}) \approx -\frac{P_0}{\omega d_l \Phi} \quad ; \quad \text{Re}(Z_{w0}) \approx d_l \left( \frac{\sigma_{p0}}{3} + \frac{P_0}{\Phi^2 d_l^2 \omega_c} \right) \quad (36)$$

443 These equations are an important result of this paper and can serve as a basis for developing  
 444 novel measurement methods of sorption and diffusion parameters of multiscale sorptive ma-  
 445 terials using sound waves. They show that the effective linearized sorption equilibrium con-  
 446 stant  $H_e$  can be extracted from measurements of the imaginary part of the surface impedance  
 447 at low frequencies, while its real part can be related to the effective diffusion coefficient  $D_e$ ,  
 448 provided that the other macroscopic parameters involved are known. A theoretical study  
 449 assessing the feasibility of such a method has been presented in Ref. 17 for the particular  
 450 case of an array of cylindrical pores embedded in a nanoporous sorptive matrix. In the  
 451 present work, Eqs. (36) are developed for triple porosity sorptive materials with complex  
 452 material morphology and depend on macroscopic independently measurable parameters.

453 The following relationships (i.e. Eq. (37)) show that the magnitude of the imaginary part  
 454 of the surface impedance of a rigidly-backed layer of multiscale sorptive material is smaller  
 455 than that of single, double, and triple porosity non-sorptive materials.

$$\text{Im}(Z_{w0}) = \frac{\phi_p}{\phi_{pm}} \frac{\text{Im}(Z_{w0\phi_p})}{1 + M_H} = \frac{\text{Im}(Z_{w0\phi_{pm}})}{1 + M_H} = \frac{\phi_{pmn}}{\phi_{pm}} \frac{\text{Im}(Z_{w0\phi_{pmn}})}{1 + M_H} \quad (37)$$

456 The relationships between the real part of the surface impedance for sorptive and non-  
 457 sorptive materials can be written as:

$$\text{Re}(Z_{w0}) = \text{Re}(Z_{w0\xi}) \left( 1 - \frac{1 - \left(\frac{\Phi}{\Phi_\xi}\right)^2 \frac{\omega_c}{\omega_{c\xi}}}{1 + \frac{\sigma_{p0}\omega_c d_l^2 \Phi^2}{3P_0}} \right)^{-1}. \quad (38)$$

458 Here the subscript  $\xi$  takes values  $\phi_p$ ,  $\phi_{pm}$ , and  $\phi_{pmn}$  for single, double, and triple porosity  
 459 non-sorptive materials. In turn,  $\Phi_{\phi_p} = \Phi(\phi_m = \phi_n = 0) = \phi_p$ ,  $\Phi_{\phi_{pm}} = \phi_{pm}$ , and  $\Phi_{\phi_{pmn}} =$   
 460  $\phi_{pmn}$ . The characteristic frequency takes the following values  $\omega_{c\phi_p} = \omega_c(\phi_m = \phi_n = 0)$ ,  
 461  $\omega_{c\phi_{pm}} = \omega_c(\phi_n = 0)$ , and  $\omega_{c\phi_{pmn}} = \omega_c(H_e = \phi_n)$ . Depending on the parameters of the  
 462 material, this ratio can be smaller or larger than one. This implies that the real part of  
 463 the surface impedance of a sorptive material can be either smaller or larger than that of  
 464 non-sorptive materials.

465 Finally, using Eq. (29), the sound absorption coefficient  $\alpha$  can be written in terms of  
 466 the low-frequency normalized resistance  $X_0^* = (d_l/Z_0)(\sigma_{p0}/3 + P_0/(\Phi^2 d_l^2 \omega_c))$  and reactance  
 467  $Y_0^* = -P_0/Z_0 \omega d_l \Phi$  as:

$$\alpha(\omega \rightarrow 0) = \frac{1}{\frac{(1+X_0^*)^2}{4X_0^*} + \frac{(Y_0^*)^2}{4X_0^*}} \quad (39)$$

468 The denominator of this expression is minimized when its first term tends to one (i.e.  $X_0^* \rightarrow$   
469 1) and its second term is as small as possible. As  $\Phi$  increases, the magnitude of the low-  
470 frequency reactance decreases. Taking into account that  $\Phi$  can take values larger than one  
471 when  $H_e$  is large, the decrease in magnitude of the low-frequency reactance is much more  
472 pronounced for sorptive materials in comparison with non-sorptive ones. It then follows that  
473 sorptive materials tend to provide larger sound absorption coefficient at low frequencies.  
474 This appears as a direct consequence of the decrease of the magnitude of the low-frequency  
475 reactance, which is determined by the larger effective low-frequency dynamic compressibility  
476 caused by sorption.

#### 477 **IV. ANALYTICAL MODEL FOR MULTISCALE SORPTIVE GRANULAR** 478 **MATERIALS**

479 An analytical model for multiscale sorptive granular materials is introduced in this section  
480 to exemplify the theory developed in this work. A model for a packing of porous grains  
481 whose inner structure comprises two different scales of heterogeneities is considered. The  
482 hierarchical structure of the material is the same as that in Ref. 24. However, the calculation  
483 of the effective parameters associated with the nano porous domain differ. Specifically, the  
484 compressibility of the nano porous domain is calculated using a single parameter  $H_e$ , instead  
485 of three parameters; and, since the inter-scale mass diffusion is not assumed quasi-static, an  
486 analytical expression for the frequency-dependent function representing this phenomena is  
487 introduced.

488 The smaller inner-grain scale corresponds to the nano porous domain and is modeled as  
489 an effective medium with parameters (see Eqs. (A.6) and (A.5)):  $H_e = \phi_n(\varphi + (1 - \varphi)H)$  and  
490  $D_e = \phi_n(\varphi D_n + (1 - \varphi)D_s H)$ , where  $\phi_n$  is the nano porosity. Taking into account that the  
491 ratio between the nanopore surface area and volume is inversely proportional to the nanopore  
492 characteristic size, the transport void fraction is approximated as  $\varphi = 1 - d/r_n$ , where  $r_n$   
493 is the radius of the cylindrical nanopores. It is additionally considered that the diffusion  
494 mechanism in the bulk of the nanopores is Knudsen diffusion<sup>1</sup>, i.e.  $D_n = D_k = 2r_n v_T/3$   
495 with  $v_T$  being the mean thermal speed; while the surface diffusion coefficient is calculated  
496 as<sup>1</sup>:  $D_s = (1/4)\zeta v_T \exp(-E_a/R_g \tau_0)$ . Here  $R_g$  is the gas constant,  $\zeta$  is the distance between  
497 adjacent sites (which is approximated by the molecule size, i.e.  $\zeta \approx d$ ), and  $E_a$  is the energy

498 of activation needed for a jump, which is in the order of a third of the heat of adsorption<sup>1</sup>.  
 499 The linearized sorption equilibrium constant  $H$  can be modeled using a Langmuir isotherm  
 500 model<sup>25</sup> (see Eq. (A.37)). Alternatively, one can directly use values of  $H_e$  and  $D_e$  as inputs to  
 501 the model. It is worth noting that the former can be obtained from isotherm measurements<sup>1</sup>,  
 502 while the latter using chromatographic methods, among others<sup>1,9,14</sup>.

503 The larger inner-grain scale is the micropore domain  $\Omega_m$ . This is modeled as an array  
 504 of cylindrical micropores with radius  $r_m$  and micro porosity  $\phi_m$ . The dynamic viscous and  
 505 thermal permeabilities associated with the micropore fluid network accounting for rarefaction  
 506 effects, i.e.  $\mathcal{K}_m(\omega)$  and  $\Theta_m(\omega)$ , are calculated from the solution of Eqs. (A.18)-(A.19)  
 507 and (A.21) with boundary conditions (A.42) and (A.43), respectively. These permeabilities  
 508 depend on the Knudsen number  $\text{Kn} = \ell/r_m$  with  $\ell$  being the molecular mean free path, and  
 509 their expressions, which involve Bessel functions  $J_{0,1}$  of the first kind of order 0 and 1, have  
 510 been obtained in Ref. 37 and are shown in Table I.

511 The pore-scale geometry is modeled as an array of spherical grains with grain radius  $r_p$  and  
 512 inter-granular void porosity  $\phi_p$ . The dynamic thermal and viscous permeabilities associated  
 513 with the pore fluid network, i.e.  $\Theta_p(\omega)$  and  $\mathcal{K}_p(\omega)$ , are calculated from the solution of  
 514 Eqs. (A.18)-(A.20) and (A.21)-(A.22) (with the subscript  $m$  being replaced by  $p$ ). A self-  
 515 consistent approach, as detailed in Refs. 38 and 39, has been used. The expressions for  
 516 these permeabilities are shown in Tables I and II, respectively.

517 The inter-scale (inner-grain micropores to/from nano porous domain) mass diffusion func-  
 518 tion  $G(\omega)$  is obtained from the solution of Eqs. (A.23)-(A.24). A self-consistent approach,  
 519 as detailed in Refs. 25 and 40, has been used. The final expression involves modified Bessel  
 520 functions  $I_i$  and  $K_i$  of the first and second kind of order  $i$  and is shown in Table I.

521 The inter-scale (inter-granular voids to/from grains) pressure diffusion function  $B(\omega)$  is  
 522 calculated from the solution of Eqs. (B.4)-(B.5) using a self-consistent approach<sup>23</sup> and its  
 523 expression is shown in table I.

TABLE I. Analytical model for the dynamic compressibility of the effective fluid saturating a multiscale sorptive granular material.

<p>Effective dynamic compressibility</p> $\mathbf{C}(\omega) = \mathbf{C}_p(\omega) + (1 - \phi_p)\mathbf{C}_{mn}(\omega)\mathcal{F}_{pmn}(\omega),$ $\mathcal{F}_{pmn}(\omega) = 1 - \frac{j\omega B(\omega)}{(1-\phi_p)\mathcal{B}_{app}(\omega)}.$
<p>Effective dynamic compressibility and thermal permeability of the inter-granular voids</p> $\mathbf{C}_p(\omega) = \frac{\phi_p}{P_0} \left( 1 - j\omega\rho_0 C_p \frac{\gamma-1}{\gamma} \frac{\Theta_p(\omega)}{\phi_p\kappa} \right),$ $\Theta_p(\omega) = -j(1 - \beta^3)\delta_t^2 \left( 1 - \frac{\beta}{1-\beta^3} \frac{3}{z_t^2} \left( 1 - \beta z_t \frac{1+z_t \tanh(z_t(\beta-1))}{z_t + \tanh(z_t(\beta-1))} \right) \right),$ <p>where <math>\beta = \sqrt[3]{1 - \phi_p}</math>, <math>z_t = j^{1/2} \frac{r_p}{\beta\delta_t}</math>, and <math>\delta_t = \sqrt{\frac{\kappa}{\rho_0 C_p \omega}}</math>.</p>
<p>Effective dynamic compressibility of the micro-nano porous domain (i.e. the grains)</p> $\mathbf{C}_{mn}(\omega) = \mathbf{C}_m(\omega) + (1 - \phi_m)\mathbf{C}_n\mathcal{F}_{mn}(\omega),$ $\mathcal{F}_{mn}(\omega) = 1 - \frac{j\omega G(\omega)}{(1-\phi_m)\mathcal{D}_{app}}.$
<p>Effective dynamic compressibility and thermal permeability of the inner-grain micropores</p> $\mathbf{C}_m(\omega) = \frac{\phi_m}{P_0} \left( 1 - j\omega\rho_0 C_p \frac{\gamma-1}{\gamma} \frac{\Theta_m(\omega)}{\phi_m\kappa} \right),$ $\Theta_m(\omega) = -j\phi_m\delta_t^2 \left( 1 - \frac{2}{X_t} \frac{J_1(X_t)}{J_0(X_t) - k_t X_t J_1(X_t)} \right),$ <p>with <math>X_t = j^{3/2} \frac{r_m}{\delta_t}</math>, <math>k_t = 2 \frac{\gamma \text{Kn}}{(\gamma+1)\text{Pr}}</math>, <math>\text{Kn} = \frac{\ell}{r_m}</math>, and <math>\text{Pr} = \frac{C_p \eta}{\kappa}</math>.</p>
<p>Effective compressibility of the inner-grain nano porous domain</p> $\mathbf{C}_n = \frac{H_e}{P_0}.$
<p>Inter-scale (inner-grain micropores to/from nano porous domain) mass diffusion function</p> $G(\omega) = -j(1 - \phi_m)\delta_d^2 \left( 1 - \frac{2\phi_m}{1-\phi_m} \frac{R_1(\xi_d)}{R_0(\xi_d)} \right),$ <p>with <math>R_i(\xi_d) = \left( \frac{1}{\xi_d} \right)^i \left( K_i(\xi_d) + (-1)^i \frac{K_1(\xi_d/\sqrt{\phi_m})}{I_1(\xi_d/\sqrt{\phi_m})} I_i(\xi_d) \right)</math>, <math>i = 0, 1</math>;</p> $\xi_d = \sqrt{j} \frac{r_m}{\delta_d}, \quad \delta_d = \sqrt{\frac{\mathcal{D}_{app}}{\omega}}, \quad \text{and} \quad \mathcal{D}_{app} = \frac{D_e}{H_e}.$
<p>Inter-scale (inter-granular voids to/from grains) pressure diffusion function</p> $B(\omega) = -j(1 - \phi_p)\delta_b^2 \left( 1 - 3\xi_b^{-2}(1 - \xi_b \cot(\xi_b)) \right),$ <p>where <math>\xi_b = j^{3/2} \frac{r_p}{\delta_b}</math>, <math>\delta_b = \sqrt{\frac{\mathcal{B}_{app}}{\omega}}</math>, and <math>\mathcal{B}_{app}(\omega) = \frac{\mathcal{K}_m(\omega)}{\eta \mathbf{C}_{nm}(\omega)}</math>.</p>
<p>Viscous permeability of the inner-grain micropores</p> $\mathcal{K}_m(\omega) = -j\phi_m\delta_v^2 \left( 1 - \frac{2}{X_v} \frac{J_1(X_v)}{J_0(X_v) - k_v X_v J_1(X_v)} \right),$ <p>with <math>X_v = j^{3/2} \frac{r_m}{\delta_v}</math>, <math>k_v = \text{Kn} = \frac{\ell}{r_m}</math>, and <math>\delta_v = \sqrt{\frac{\eta}{\rho_0 \omega}}</math>.</p>

TABLE II. Analytical model for the dynamic viscous permeability of a multiscale sorptive granular material.

<p>Dynamic viscous permeability</p> $\mathcal{K}(\omega) = \mathcal{K}_p(\omega) = -j(1 - \beta^3)\delta_v^2 \left( 1 - \frac{\beta^3}{1-\beta^3} \frac{\varsigma^{\frac{1-\beta^3}{\beta^3}+1}}{\varsigma-1} \right),$ $\varsigma = \frac{3}{z^2} \frac{A_p z + B_p \tanh(z(\beta-1))}{a_p z + b_p \tanh(z(\beta-1))},$ $A_p = (3 + (\beta z)^2) \left( 1 + \frac{z^2}{6} \right) - 3\beta \left( 1 + \frac{z^2}{2} \right),$ $a_p = \frac{1}{3}(3 + (\beta z)^2) - 3\beta - \frac{2}{\beta} \left( 1 + \frac{z^2}{6} \right) + \frac{4}{\cosh(z(\beta-1))},$ $B_p = (3 + (\beta z)^2) \left( 1 + \frac{z^2}{2} \right) - 3\beta z^2 \left( 1 + \frac{z^2}{6} \right),$ $b_p = 3 + \beta(\beta - 1)z^2 - \frac{2}{\beta} \left( 1 + \frac{z^2}{2} \right),$ $\beta = \sqrt[3]{1 - \phi_p}, \quad \text{and} \quad z = j^{1/2} \frac{r_p}{\beta \delta_v}.$
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TABLE III. Static values of the effective dynamic permeabilities and inter-scale diffusion functions of a multiscale sorptive granular material.

$\Theta_{p0} = \frac{r_p^2}{15} \left( \frac{5-9\beta+5\beta^3-\beta^6}{\beta^3} \right)$
$\Theta_{m0} = \phi_m \frac{r_m^2}{8} \left( 1 + \frac{8\gamma}{(\gamma+1)\text{Pr}} \text{Kn} \right)$
$G_0 = \frac{r_m^2}{8\phi_m} (-2\ln(\phi_m) + 4\phi_m - \phi_m^2 - 3)$
$B_0 = (1 - \phi_p) \frac{r_p^2}{15}$
$\mathcal{K}_{m0} = \phi_m \frac{r_m^2}{8} (1 + 4\text{Kn})$
$\mathcal{K}_{p0} = \frac{r_p^2}{3\beta^2} \left( \frac{2+3\beta^5}{\beta(3+2\beta^5)} - 1 \right)$

524 Using the static values of the thermal permeabilities and inter-scale diffusion functions  
525 shown in table III, one can explicitly write the characteristic frequencies determining the  
526 behavior of the effective dynamic compressibility as:

$$\omega_{tp} = \frac{\phi_p \kappa}{\rho_0 C_p \Theta_{p0}} = \frac{\kappa}{\rho_0 C_p} \frac{15}{r_p^2} \frac{\beta^3(1 - \beta^3)}{(5 - 9\beta + 5\beta^3 - \beta^6)}, \quad (40)$$

$$\omega_{tm} = \frac{\phi_m \kappa}{\rho_0 C_p \Theta_{m0}} = \frac{\kappa}{\rho_0 C_p} \frac{8}{r_m^2} \left( 1 + \frac{8\gamma \text{Kn}}{(\gamma + 1)\text{Pr}} \right)^{-1}, \quad (41)$$

$$\omega_d = \frac{(1 - \phi_m) \mathcal{D}_{app}}{G_0} = \frac{D_e}{H_e} \frac{8}{r_m^2} \frac{\phi_m (1 - \phi_m)}{(-2 \ln \phi_m + 4\phi_m - \phi_m^2 - 3)}, \quad (42)$$

$$\omega_b = \frac{(1 - \phi_p) \mathcal{B}_{app0}}{B_0} = \frac{P_0}{\eta} \frac{15}{8} \frac{r_m^2}{r_p^2} \phi_m \frac{1 + 4\text{Kn}}{\phi_m + (1 - \phi_m) H_e}. \quad (43)$$

Eq. (41) shows that rarefaction effects reduce the thermal characteristic frequency  $\omega_{tm}$ , while Eq. (42) indicates that the effective mass diffusion is slowed down by sorption. This effect on the pressure diffusion characteristics of the material can, however, be compensated by rarefaction effects, as shown by Eq. (43).

The introduced model depends effectively on six parameters, namely the grain radius  $r_p$ , inter-granular porosity  $\phi_p$ , micropore radius  $r_m$ , micro porosity  $\phi_m$ , and the effective parameters of the nano porous domain, i.e.  $D_e$  and  $H_e$ . This model can be simplified since for the materials of interest, the micropore size is in the order of the molecular mean free path and sound propagation in the micropores can therefore be considered as viscosity-dominated and isothermal. Hence, the dynamic viscous and thermal permeabilities associated with the micropore fluid network can be approximated, in the audible frequency range, by  $\mathcal{K}_m(\omega) = \mathcal{K}_{m0}$  and  $\Theta_m(\omega) = \Theta_{m0}$ . This leads to  $\mathcal{C}_m = \phi_m/P_0$ ,  $\mathcal{C}_{mn} = \phi_m/P_0 + (1 - \phi_m)\mathcal{C}_n \mathcal{F}_{mn}$ , and  $\mathcal{B}_{app} = \mathcal{K}_{m0}/\eta \mathcal{C}_{mn}$ . This shows that, in the audible frequency range, sound propagation in multiscale sorptive materials is mainly affected by viscosity and heat transfer effects at the pore scale and the inter-scale mass and pressure diffusion processes. The former diffusion process is influenced by sorption while the latter by both sorption and rarefaction.

## V. ILLUSTRATING EXAMPLES AND EXPERIMENTAL VALIDATION

### A. Illustrating examples

The acoustical properties of multiscale sorptive porous materials are illustrated in this section. First, we present results for the effective dynamic compressibility due to its significant influence on the acoustical properties of this type of materials. We consider the following parameters of a multiscale granular sorptive material:  $r_p = 1$  mm,  $\phi_p = 0.4$ ,  $r_m = 1$   $\mu$ m,  $\phi_m = 0.5$ ,  $r_n = 1$  nm,  $\phi_n = 0.2$ ,  $H = 75$ , and  $E_a = 10$  kJ/mole. The value of the energy of activation needed for a jump  $E_a$  has been set equal to a third of the heat of

553 adsorption, which is in the order of 20 to 40 kJ/mole for activated carbons<sup>1</sup>. In addition,  
 554 it should be reminded that the developed theory applies to materials saturated with a pure  
 555 fluid. However, for simplicity, the saturating fluid parameters are set equal to those of air  
 556 (with molecular size  $d \approx 0.38$  nm), which are close to those of nitrogen at the considered  
 557 normal pressure ( $P_0 = 101325$  Pa) and temperature ( $\tau_0 = 293.15$  K) conditions.

558 Figure 2 shows the real part of the effective dynamic compressibility, [calculated using the](#)  
 559 [model shown in table I](#), of a triple-porosity sorptive material normalized to the isothermal  
 560 value of this quantity for a triple-porosity non-sorptive material, i.e.  $\text{Re}(P_0\mathbf{C}(\omega)/\phi_{pmn})$ .  
 561 This is compared with that of single, double, and triple porosity non-sorptive materials.  
 562 Respectively, these correspond to a packing of solid grains (i.e.  $\phi_n = \phi_m = 0$ ), and of  
 563 porous grains without nanopores (i.e.  $\phi_n = 0$ ) and with  $H_e = \phi_n$  and effective diffusion  
 564 coefficient  $D_e = \phi_n D_k$ . Sorption induces a significant increase of the real part of the dynamic  
 565 compressibility at low frequencies. For the four cases, the low-frequency limiting values are  
 566 well described by Eq. (33), while at high frequencies the real part of the effective dynamic  
 567 compressibilities tend to the same limiting value, as predicted by Eq. (30).

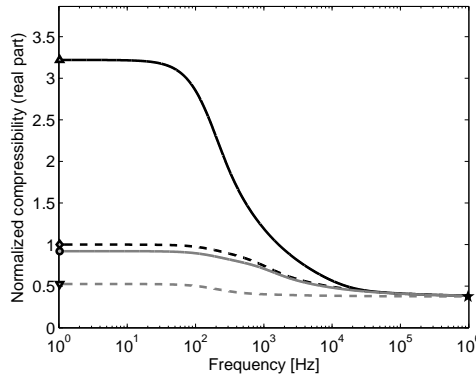


FIG. 2. Real part of the normalized effective dynamic compressibility  $\text{Re}(P_0\mathbf{C}(\omega)/\phi_{pmn})$  as a function of frequency. Continuous black line: multiscale sorptive granular material. Dashed black line : triple porosity non-sorptive granular material (i.e.  $H_e = \phi_n$  and  $D_e = \phi_n D_k$ ). Continuous gray line: double porosity non-sorptive granular material (i.e.  $\phi_n = 0$ ). Dashed gray line: single porosity non-sorptive granular material (i.e.  $\phi_n = \phi_m = 0$ ). The markers show the asymptotic values predicted by Eqs. (33) and (30).

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570 The negative of the imaginary part of the normalized effective dynamic compressibility,  
 571 which is associated with the acoustic losses in the material, and the characteristic frequencies



572 determining its behavior are shown in Figure 3. These have been calculated using the  
 573 model shown in table I and Eqs. (32) and (40)-(43), respectively. Multiscale sorptive  
 574 materials provide much larger sound attenuation than non-sorptive ones in the low frequency  
 575 range. This is a direct consequence of the combined effect of heat transfer in the pores and  
 576 inter-scale mass and pressure diffusion with the latter two being influenced by sorption.  
 577 Such attenuation is maximized around the characteristic frequencies associated with these  
 578 phenomena, i.e. around  $\omega_{tp}$ ,  $\omega_d$ , and  $\omega_b$ .

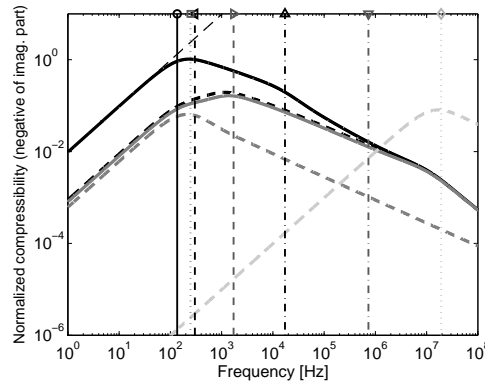


FIG. 3. Negative of the imaginary part of the normalized effective dynamic compressibility  $-\text{Im}(P_0C(\omega)/\phi_{pmn})$  as a function of frequency. Continuous black line: multiscale sorptive granular material. Dashed black line : triple porosity non-sorptive granular material (i.e.  $H_e = \phi_n$  and  $D_e = \phi_n D_k$ ). Continuous gray line: double porosity non-sorptive granular material (i.e.  $\phi_n = 0$ ). Dashed dark gray line: single porosity non-sorptive granular material (i.e.  $\phi_n = \phi_m = 0$ , non-porous grains). Dashed light gray line: single porosity non-sorptive monolithic material with micropores only (i.e.  $\phi_n = \phi_p = 0$ ). Thin dashed black line : low-frequency asymptotic value  $-\text{Im}(P_0C_{lf}(\omega)/\phi_{pmn})$  (see Eq. (31)). The vertical lines with markers represent characteristic frequencies. Circle :  $f_c = \omega_c/2\pi$ . Square:  $f_{tp}$ . Left-pointing triangle :  $f_b$ . Right-pointing triangle :  $f_b$  (no sorption). Upward-pointing triangle :  $f_d$ . Downward-pointing triangle :  $f_d$  (no sorption). Diamond :  $f_{tm}$ . The material parameters are as in Figure 2.

579 In comparison with a triple porosity non-sorptive material, the characteristic frequencies  
 580  $\omega_d$  and  $\omega_b$  decrease as a direct consequence of the local slowing down of mass diffusion by  
 581 sorption (see Eq. (42)), and the increase in effective dynamic compressibility also caused  
 582 by this phenomena (see Eqs. (C.3) and (43)). This can be seen by comparing the vertical  
 583 lines with left-pointing and upward-pointing triangles with those with right-pointing and

584 downward-pointing triangles. Moreover, in the absence of sorption, the expected peak as-  
585 sociated with inter-scale mass diffusion becomes negligible. This is because the influence  
586 of the inner-grain physics on the macroscopic behavior becomes smaller when  $\omega \gg \omega_b$ .  
587 This indicates that in order to be able to observe (and/or take advantage of) the effects  
588 associated with the inner-grain physics, one should ideally have that  $O(\omega_b) = O(\omega_d)$ , i.e.  
589  $O((1 - \phi_p)\mathcal{B}_{app0}B_0^{-1}) = O((1 - \phi_m)\mathcal{D}_{app}G_0^{-1})$ . This condition, which guarantees the rich  
590 interplay between the different physical phenomena and geometric features of multiscale  
591 sorptive materials, for the geometry considered here is given by Eq. (44) with  $\mathcal{A}$  being a  
592 constant that is ideally  $\mathcal{A} \geq 1$  but not extremely larger than unity.

$$\frac{64}{15} \frac{\eta}{P_0} \frac{\mathcal{D}_{app}}{r_m^2} \frac{r_p^2}{r_m^2} \frac{\phi_m + (1 - \phi_m)H_e}{1 + 4\text{Kn}} \frac{1 - \phi_m}{-2 \ln(\phi_m) + 4\phi_m - \phi_m^2 - 3} = \mathcal{A}. \quad (44)$$

593 For the material parameters considered in the example shown in Figure 3, the double  
594 porosity non-sorptive material behaves similarly as the triple-porosity non-sorptive one. The  
595 two single porosity non-sorptive materials (i.e. a packing of non-porous grain  $\phi_n = \phi_m = 0$   
596 and a monolithic material with micropores only  $\phi_n = \phi_p = 0$ ) also displayed in Figure  
597 3, clearly show the positions of the peaks associated with heat transfer in the pores and  
598 micropores, as well as their influence on the behavior of the multiscale material. It should  
599 be noted that, in this example, rarefaction effects are negligible since  $\text{Kn} = 0.06$ . Hence these  
600 effects do not substantially compensate the decrease in  $\omega_b$  caused by sorption. For materials  
601 with smaller micropores, rarefaction effects can become significant, as will be shown later  
602 in the paper. On the other hand, the asymptotic value of the compressibility, given by  
603 Eq. (31) and shown with thin dashed lines in Figure 3, correctly predicts the behavior for  
604 frequencies  $\omega \ll \omega_{\min}$ , while the global characteristic frequency  $\omega_c$  appears as a parameter  
605 that allows identifying, in a simple manner, the frequency range where the sound attenuation  
606 is maximized. Furthermore, it should be emphasized that the results presented in Figures  
607 2 and 3 are valid in the absence of scattering. This is estimated to occur at a frequency in  
608 the order of  $f_{sca} = c_0/2\pi l_p \approx c_0/4\pi r_p$ .

609 The effects previously discussed, i.e. inter-scale mass and pressure diffusion and sorption,  
610 lead to a decrease in sound speed and an increase of the overall sound attenuation, as  
611 predicted by Eqs. (34)-(35). This is shown in Figures 4 and 5 where the normalized speed  
612 of sound and attenuation coefficient are respectively presented. [Note that i\) these have been](#)  
613 [calculated using the model shown in tables I and II and Eq. \(28\), and ii\) the normalization of](#)

614 the former has been made to  $c_0/\sqrt{\alpha_\infty}$  with<sup>38</sup>  $\alpha_\infty = (3 - \phi_p)/2$ ; while that of the attenuation  
 615 coefficient to  $\omega/c_0$ . In addition, the ratio of these quantities for sorptive and non-sorptive  
 616 materials is well predicted by Eqs. (34) and (35), as can be seen in the inset plots.

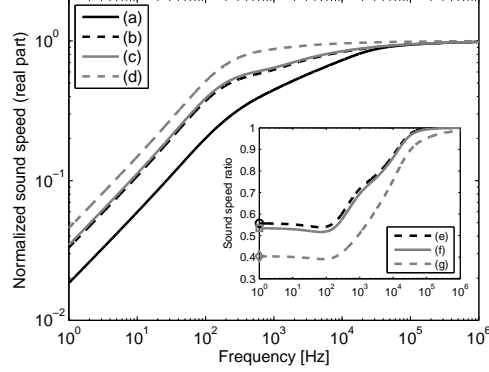


FIG. 4. Real part of the normalized sound speed  $\sqrt{\alpha_\infty}\text{Re}(\mathcal{C}(\omega))/c_0$  as a function of frequency for multiscale sorptive (a) and non-sorptive triple (b), double (c), and single (d) porosity granular materials. The inset plot shows the real part of the sound speed ratios: (e)  $\text{Re}(\mathcal{C}(\omega))/\text{Re}(\mathcal{C}_{\phi_{pmn}}(\omega))$ , (f)  $\text{Re}(\mathcal{C}(\omega))/\text{Re}(\mathcal{C}_{\phi_{pm}}(\omega))$ , and (g):  $\text{Re}(\mathcal{C}(\omega))/\text{Re}(\mathcal{C}_{\phi_p}(\omega))$ . The markers correspond to the low-frequency asymptotic values of these ratios (see Eq. (34)).

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620 As discussed in Section III, the behavior in frequency of  $\mathcal{C}(\omega)$  is determined by the  
 621 characteristic frequencies  $\omega_{tp}$ ,  $\omega_d$ ,  $\omega_b$ , and  $\omega_{tm}$ , which depend on physical and geometrical  
 622 parameters of the material. In particular, it has been shown in Figure 3 that the global  
 623 characteristic frequency  $\omega_c$  provides a good indication of the frequency range where the  
 624 acoustic losses are maximized. The global characteristic frequency, [calculated using Eq.](#)  
 625 [\(32\)](#), is shown in Figure 6 as a function of the effective diffusion coefficient  $D_e$  for different  
 626 values of the effective linearized sorption equilibrium constant ( $H_e = 2, 4, 8$ ) and micropore  
 627 radii ( $r_m = 0.2 \mu\text{m}$  and  $r_m = 2 \mu\text{m}$ ). The inset plot shows the characteristic frequencies  
 628  $f_d$ ,  $f_b$ , and  $f_c$  as a function of  $D_e$  for  $r_m = 0.2 \mu\text{m}$  and  $H_e = 4$ . For a given micropore  
 629 radius value, the global characteristic frequency  $f_c$  increases as  $D_e$  does until it reaches a  
 630 plateau for frequencies  $f_d \gg f_b$ . The transition is characterized by  $f_d = f_b$ , as shown with  
 631 a vertical dotted line in the inset plot. The plateau region is reached at larger values of  $D_e$   
 632 when the micropore radius becomes larger. On the other hand,  $f_c$  decreases as  $H_e$  increases  
 633 and is dominated by either  $f_b$  or  $f_d$ , whichever is much lower.

634 The influence of the grain radius  $r_p$  on  $f_c$  is shown in Figure 7. As previously observed,

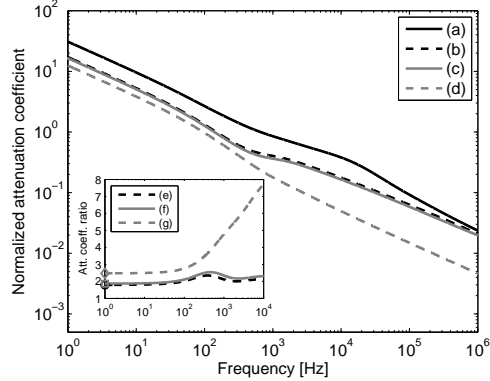


FIG. 5. Normalized attenuation coefficient  $-c_0 \text{Im}(k_c(\omega))/\omega$  as a function of frequency for multi-scale sorptive (a) and non-sorptive triple (b), double (c), and single (d) porosity granular materials. The inset plot shows the attenuation coefficient ratios : (e)  $\text{Im}(k_c(\omega))/\text{Im}(k_{c\phi_{pm}}(\omega))$ , (f)  $\text{Im}(k_c(\omega))/\text{Im}(k_{c\phi_p}(\omega))$ , and (g):  $\text{Im}(k_c(\omega))/\text{Im}(k_{c\phi_p}(\omega))$ . The markers correspond to the low-frequency asymptotic values of these ratios (see Eq. (35)).

635 the global characteristic frequency, [calculated using Eq. \(32\)](#), is dominated by  $f_d$  when  
 636  $f_d \ll f_b$ , while by  $f_b$  when  $f_d \gg f_b$ . In addition,  $f_c$  decreases when  $H_e$  becomes larger.  
 637 For small values of  $D_e$ , the influence of the grain radius on the global characteristic frequency  
 638 is negligible. This is because  $f_c$  is determined by  $f_d$ , which does not depend on  $r_p$ . As  $D_e$   
 639 increases,  $f_c$  becomes larger when the grain radius is decreased. This indicates that in order  
 640 to observe and/or take advantage of the sound attenuating properties of multiscale sorptive  
 641 materials in the audible frequency one may prefer materials with small instead of large  
 642 grains.

643 The influence of rarefaction effects in the modeling is now analyzed. Figure 8 shows the  
 644 global characteristic frequency  $f_c$ , [calculated using Eq. \(32\)](#), as a function of the Knudsen  
 645 number  $\text{Kn}$  for several values of micro porosity  $\phi_m$ . Note that the results are plotted up  
 646 to  $\text{Kn} = 1$ . Although the modified continuum description is not theoretically valid for  
 647  $\text{Kn} > 0.1$ , satisfactory agreement between measured data and theoretical predictions of  
 648 rarefied gas flow through straight cylindrical tubes has been demonstrated in Ref. 37. As  
 649 previously, the global characteristic frequency is determined by either  $f_d$  or  $f_b$ , whichever  
 650 is much lower; as shown in the inset plot. When  $f_c$  is determined by  $f_b$ , the prediction of  
 651 the global characteristic frequency is underestimated if one does not account for rarefaction  
 652 effects in the modeling. As an example,  $f_c$  could be predicted to be 5 times smaller when

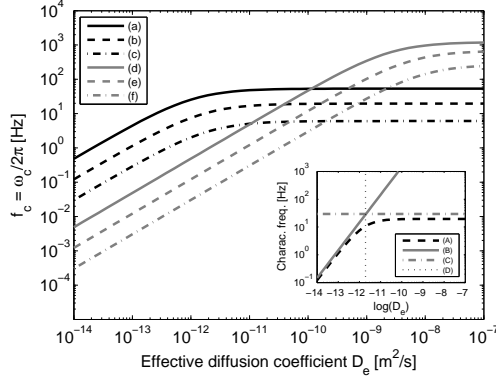


FIG. 6. Global characteristic frequency  $f_c$  as a function of the effective diffusion coefficient  $D_e$  for different values of effective linearized sorption equilibrium constant  $H_e$  and micropore radius  $r_m$ . The other parameters are as in Figure 2. Main plot : (a)  $H_e = 2$  and  $r_m = 0.2 \mu\text{m}$ . (b)  $H_e = 4$  and  $r_m = 0.2 \mu\text{m}$ . (c)  $H_e = 8$  and  $r_m = 0.2 \mu\text{m}$ . (d)  $H_e = 2$  and  $r_m = 2 \mu\text{m}$ . (e)  $H_e = 4$  and  $r_m = 2 \mu\text{m}$ . (f)  $H_e = 8$  and  $r_m = 2 \mu\text{m}$ . The inset plot shows the characteristic frequencies (A)  $f_c$ , (B)  $f_d$ , and (C)  $f_b$  as a function of  $D_e$  for  $H_e = 4$  and  $r_m = 0.2 \mu\text{m}$ . The vertical dotted line (D) shows  $D_e$  for  $f_d = f_b$ .

653  $r_m = \ell$ . Evidently, for small values of Knudsen number the influence of rarefaction effects is  
 654 negligible. On the other hand, the global characteristic frequency presents higher values as  
 655 the micro porosity  $\phi_m$  increases and is maximized when the micropore radius approximately  
 656 takes a value for which  $O(f_d) = O(f_b)$ .

657 Figure 9 shows the normal incidence sound absorption coefficient of a rigidly-backed 3-cm  
 658 thick layer of multiscale sorptive granular material in comparison with that of non-sorptive  
 659 granular ones, i.e. packing of solid grains (single porosity,  $\phi_n = \phi_m = 0$ ), of porous grains  
 660 without nanopores (double porosity,  $\phi_n = 0$ ), and of porous grains with two inner-grain  
 661 scales (triple porosity material,  $H_e = \phi_n$ ). [These have been calculated using Eqs. \(29\) and](#)  
 662 [\(28\) and the model shown in tables I and II.](#) The plot clearly shows that multiscale sorptive  
 663 granular materials provide larger sound absorption coefficient than non-sorptive materials.  
 664 This increase is enhanced further for larger values of  $H_e$ . The inset plot shows that  $\alpha$  is well  
 665 approximated at low frequencies by its asymptotic value Eq. (39). Hence, this provides a  
 666 simple expression to evaluate the low-frequency sound absorption coefficient of multiscale  
 669 sorptive materials.

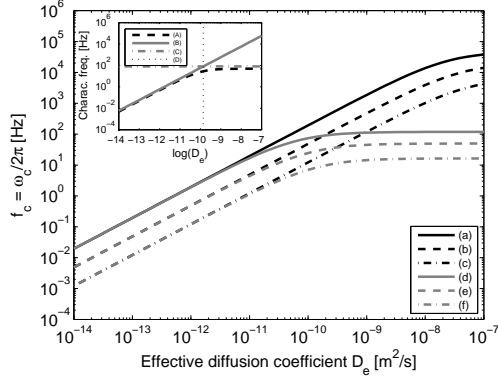


FIG. 7. Global characteristic frequency  $f_c$  as a function of the effective diffusion coefficient  $D_e$  for different values of linearized effective sorption equilibrium constant  $H_e$  and grain radius  $r_p$ . The other parameters are as in Figure 2. Main plot : (a)  $H_e = 2$  and  $r_p = 0.1$  mm. (b)  $H_e = 4$  and  $r_p = 0.1$  mm. (c)  $H_e = 8$  and  $r_p = 0.1$  mm. (d)  $H_e = 2$  and  $r_p = 2$  mm. (e)  $H_e = 4$  and  $r_p = 2$  mm. (f)  $H_e = 8$  and  $r_p = 2$  mm. The inset plot shows the characteristic frequencies (A)  $f_c$ , (B)  $f_d$ , and (C)  $f_b$  as a function of  $D_e$  for  $H_e = 4$  and  $r_p = 2$  mm. The vertical dotted line (D) shows  $D_e$  for  $f_d = f_b$ .

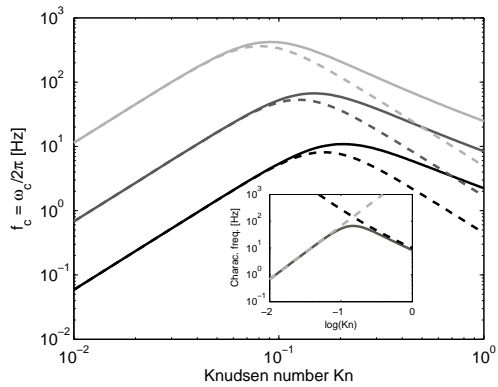


FIG. 8. Global characteristic frequency  $f_c$  as a function of the Knudsen number  $\text{Kn}$  for micro porosity values  $\phi_m = 0.1$  (black lines),  $\phi_m = 0.3$  (dark gray lines), and  $\phi_m = 0.6$  (light gray lines). Continuous lines : model accounting for rarefaction effects. Dashed lines : model without accounting for rarefaction effects. The other parameters are  $r_p = 0.75$  mm,  $\phi_p = 0.3$ ,  $D_e = 10^{-10}$  m<sup>2</sup>/s, and  $H_e = 2$ . The inset plot shows  $f_c$  (continuous dark gray line),  $f_b$  (dashed black line), and  $f_d$  (dashed light gray line) as a function of  $\text{Kn}$  for  $\phi_m = 0.3$ .

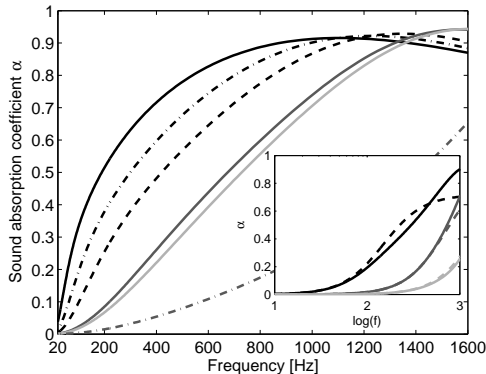


FIG. 9. Sound absorption coefficient  $\alpha$  of a 3-cm rigidly-backed layer of material. Multiscale sorptive granular materials: black continuous ( $H_e = 8$ ), dashed ( $H_e = 4$ ), and dashed-dotted ( $H_e = 2$ ) lines. Packing of non-sorptive porous grains with two inner-grain scales of heterogeneities (triple porosity, continuous dark gray line), without nano pores (double porosity, continuous light gray line), and of solid grains (single porosity, dashed-dotted gray line). Inset plot: low-frequency  $\alpha$  for a multiscale sorptive granular material ( $H_e = 4$ , continuous black line), and double (continuous dark gray line) and single (continuous light gray line) porosity non-sorptive granular materials; and its asymptotic values (Eq. (39), dashed lines). The other parameters are  $r_p = 0.75$  mm,  $\phi_p = 0.3$ ,  $r_m = 0.5$   $\mu\text{m}$ ,  $\phi_m = 0.5$ ,  $D_e = 10^{-9}$  m<sup>2</sup>/s, and  $\phi_n = 0.1$ .

## 670 B. Experimental validation

### 671 1. Material characterization

672 This section describes the characterization of a granular activated carbon (GAC) sample.  
 673 The characterization procedure for the parameters describing the inter-granular physical  
 674 process and those in the micropores is similar to that in Ref. 24. The characterization  
 675 of the parameters describing sorption and mass diffusion differs. Measurements of surface  
 676 impedance are used to deduce these parameters, as detailed below.

677 The highly activated GAC sample is made out of coal, its N<sub>2</sub> surface area is 1274 m<sup>2</sup>/g,  
 678 and its model parameters are shown in Table IV. It is detailed throughout this section how  
 679 these six parameters have been measured or deduced.

680 An equivalent grain radius  $r_p$  of the GAC sample has been measured using optical gran-  
 681 ulometry following the procedure detailed in Refs. 22 and 23. The equivalent grain radius

TABLE IV. Parameters of the granular activated carbon sample.

$\phi_p$	$r_p$ [mm]	$\phi_m$	$r_m$ [ $\mu\text{m}$ ]	$H_e$	$D_e \times 10^9 [\text{m}^2/\text{s}]$
0.2997	0.7363	0.7064	0.3695	7.1189	2.9156

682 distribution follows a log-normal distribution  $f(t|\mu, \theta) = (1/t\theta\sqrt{2\pi}) \exp -(\ln t - \mu)^2/2\theta^2$   
 683 with parameters  $\mu = -7.2513$  and  $\theta = 0.2741$ . The equivalent grain radius  $r_p$  is set to the  
 684 mean value of the equivalent grain radius distribution, i.e.  $r_p = 0.7363$  (0.2056) mm, where  
 685 the value in round brackets corresponds to the standard deviation.

686 The GAC sample exhibits well separated scales. This implies that the overall permeability  
 687 can be approximated by that of the pore-scale fluid network. Measurements of flow resistivity  
 688  $\sigma_{p0} = \eta/\mathcal{K}_{p0}$ , taken by following the procedure described in Ref. 41, are used to estimate  
 689 the inter-granular void porosity  $\phi_p$ . This is made by using the measured value of  $r_p$  and  
 690 inverting the expression for  $\mathcal{K}_{p0}$  shown in table III, as detailed in Ref. 23. The average flow  
 691 resistivity value is  $\sigma_{p0} = 24.5923$  (1.5104) kPa.s/m<sup>2</sup>, which yields  $\phi_p = 0.2997$  (0.005).

692 The overall porosity  $\phi_{pmn} = 1 - \rho_b/\rho_C$  is calculated from the measurement of the bulk  
 693 density  $\rho_b$  and the density of the material solid frame, which is assumed to be that of carbon  
 694 black  $\rho_C = 2.2$  g/cm<sup>3</sup>. The measured bulk density of the GAC sample is  $\rho_b = 0.335$  g/cm<sup>3</sup>.  
 695 Hence, the overall porosity is  $\phi_{pmn} = 0.8477$ .

696 The micro porosity was calculated as  $\phi_m = (\phi_{pmn} - \phi_p - (1 - \phi_p)\phi_n)/(1 - \phi_p)(1 - \phi_n)$ .  
 697 The nano porosity has been supplied by the manufacturer and is  $\phi_n = 0.2593$ . Hence, the  
 698 micro porosity is  $\phi_m = 0.7064$ .

699 The effective linearized sorption equilibrium constant  $H_e$  is deduced from measurements  
 700 of the imaginary part of the surface impedance  $Z_w(\omega)$  of rigidly-backed layers of the GAC  
 701 sample. These measurements were taken by following the procedure described in the ISO  
 702 standard<sup>42</sup>. A vertically-installed Brüel & Kjær 4206 impedance tube was used. GAC  
 703 samples with layer thickness values ranging from 2 cm to 8 cm in steps of 1 cm were  
 704 measured<sup>22,24</sup>. The previously derived asymptotic value of the imaginary part of the surface  
 705 impedance, i.e. Eq. (36), is used to determine the apparent porosity  $\Phi$ . We remind that this  
 706 asymptotic is valid for  $|k_c d_l| \ll 1$  and  $\omega \ll \omega_{min} \approx \omega_c$ . A function  $z_1(\omega) = \frac{P_0}{-\text{Im}(Z_w)d_l} = \Phi\omega$   
 707 that is linear in frequency can be defined from Eq. (36). The apparent porosity  $\Phi$  is the  
 708 proportionality constant and can be obtained by linear fitting of  $z_1(\omega)$ . An example of this



709 is shown in the right-hand-side inset plot of Figure 10 for a 4-cm thick GAC sample. Then,  
 710 the effective linearized sorption equilibrium constant is calculated as  $H_e = ((\Phi - \phi_p)/(1 -$   
 711  $\phi_p) - \phi_m)/(1 - \phi_m)$ . Its value is  $H_e = 7.1189 \pm 0.1508$ .

712 Measurements of the real part of the surface impedance and the use of its asymptotic  
 713 value allow obtaining the global characteristic frequency  $f_c$ . Using Eq. (36), a function  
 714 linear in frequency  $z_2(\omega) = \Phi^2 d_l^2 (\text{Re}(Z_w)/d_l - \sigma_{p0}/3)/P_0 = \omega/\omega_c$  can be defined. In this  
 715 case, its slope is the inverse of the global characteristic frequency. Linear fitting of  $z_2(\omega)$   
 716 leads to  $f_c = 88.5362 \pm 5.4570$  Hz. An example of this is shown in the left-hand-side inset  
 717 plot of Figure 10 for a 4-cm thick GAC sample.

718 The remaining parameters to be determined are  $r_m$  and  $D_e$ . The micropore radius  $r_m$  is  
 719 calculated via a best-fitting routine in which the square of the absolute difference between  
 720 the predicted and measured surface impedance is minimized. As part of this routine, the  
 721 measured global characteristic frequency and the other parameters are used to calculate  
 722 the effective diffusion coefficient  $D_e$  by inverting Eq. (32), as shown by Eq. (45). The  
 723 values of the micropore radius and effective diffusion coefficient are  $r_m = 0.3695 \mu\text{m}$  and  
 724  $D_e = 2.9156 \cdot 10^{-9} \text{ m}^2/\text{s}$ . The values of  $r_m$  and  $r_p$  are consistent with the hypothesis of  
 725 large separation of scales, i.e.  $\epsilon \approx r_m/r_p = 5.0183 \times 10^{-4}$ , and the estimation Eq. (A.12)  
 726 holds, i.e.  $\eta D_e/r_m^2 P_0 = 3.8715 \times 10^{-6} = O(\epsilon^2)$ . The measured and predicted surface  
 727 impedance, calculated using Eqs. (29) and (28) and the model shown in tables I and II, of  
 728 a rigidly-backed 4-cm thick GAC layer is shown in the main plot of Figure 10. Note that  
 729 the magnitude of the imaginary part of the surface impedance of GAC is much smaller than  
 730 that of a triple porosity non-sorptive material.

$$D_e = \frac{H_e G_0}{(1 - \phi_m)} \left( \frac{\frac{1}{\omega_c} - \frac{\gamma-1}{\gamma} \frac{\phi_p}{\omega_{tp}} - (1 - \phi_p) \left( \frac{\gamma-1}{\gamma} \frac{\phi_m}{\omega_{tm}} + \frac{\phi_m}{\omega_b} \right)}{(1 - \phi_p)(1 - \phi_m)H_e} - \frac{1}{\omega_b} \right)^{-1}. \quad (45)$$

731 The values of  $r_m$ ,  $D_e$ , and  $H_e$  compare well with those commonly found in activated  
 732 carbons. For example, the size of the larger inner-grain pores, i.e.  $2r_m$ , is typically<sup>1,2</sup> in the  
 733 order of  $0.8 \mu\text{m}$ . Using Eq. (A.37) and considering that the nanopore size is comparable to  
 734 that of the fluid molecules, the linearized sorption equilibrium constant expressed in units  
 735 of adsorbed moles per mass of adsorbent is  $H = 187.9 \mu\text{mol/g}$  at normal conditions (i.e.  
 736  $P_0 = 101325 \text{ Pa}$  and  $\tau_0 = 293.15 \text{ K}$ ). This is comparable to the data found in literature  
 737 for materials saturated with nitrogen at normal pressure condition. For example, from the

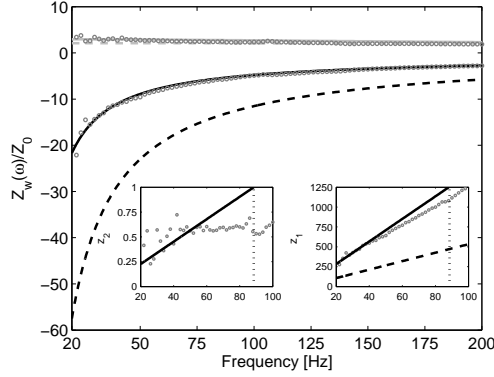


FIG. 10. Comparison between measured<sup>22,24</sup> (circles) and predicted real (gray lines) and imaginary (black lines) parts of the normalized surface impedance of a rigidly-backed 4-cm thick layer of granular activated carbon. Continuous lines – Triple porosity sorptive material. Dashed lines – Triple porosity non-sorptive material. The left-hand side inset plot shows  $z_2(\omega)$  as a function of frequency. Circles – Measurements. Continuous line – Fitted linear function with slope  $1/f_c$ . The right-hand side inset plot shows  $z_1(\omega)$  as a function of frequency. Circles – Measurements. Continuous line – Fitted linear function with slope  $\Phi$ . Dashed line – Fitted linear function with  $\Phi = \phi_{pmn}$ , i.e. triple porosity non-sorptive material. In both inset plots, the dotted lines correspond to  $f_c$

738 measured data in Ref. 43 for a granular activated carbon sample having a surface area  
 739 of  $1220 \text{ m}^2/\text{g}$  (i.e. comparable to the GAC sample used in this paper), it is deduced that  
 740  $H = 185.7 \text{ } \mu\text{mol}/\text{g}$  at  $293.15 \text{ K}$ . A value of  $H = 319.9 \text{ } \mu\text{mol}/\text{g}$  is calculated from the  
 741 measured data in Ref. 44 for a carbon molecular sieve 5A (i.e. a type of activated carbon)  
 742 at  $303.15 \text{ K}$ . For a PCB-type activated carbon, the data in Ref. 45 leads to  $H = 288.3$   
 743  $\mu\text{mol}/\text{g}$  (at  $293.15 \text{ K}$ ), while for a carbon molecular sieve in pellet form a value of  $H = 175.8$   
 744  $\mu\text{mol}/\text{g}$  (at  $300.15 \text{ K}$ ) is calculated from the data in Ref. 46.

745 The values of  $D_e$  and  $H_e$  lead to an apparent mass diffusivity of  $\mathcal{D}_{app} = 4.1 \cdot 10^{-10} \text{ m}^2/\text{s}$ .  
 746 For the case of nanopore pore size comparable to that of the fluid molecules,  $\mathcal{D}_{app}$  can be  
 747 interpreted as an activated diffusivity. Chromatographic measurements of this parameter  
 748 have been reported in Ref. 47 for carbon molecular sieve 5A saturated with nitrogen.  
 749 Its value at  $293.15 \text{ K}$  is  $\mathcal{D}_{app} = 1.85 \cdot 10^{-11} \text{ m}^2/\text{s}$ . For the same type of material and  
 750 saturating fluid, measurements reported in Ref. 44 using a gravimetric technique provides  
 751  $\mathcal{D}_{app} = 1.19 \cdot 10^{-10} \text{ m}^2/\text{s}$  at  $303 \text{ K}$ . In Ref. 48 measurements on an activated carbon

752 monolith saturated with nitrogen provided a value of  $\mathcal{D}_{app} = 1.35 \cdot 10^{-10} \text{ m}^2/\text{s}$  at 293.15  
 753 K. Furthermore, values of  $\mathcal{D}_{app} = 1.93 \cdot 10^{-9} \text{ m}^2/\text{s}$  and  $\mathcal{D}_{app} = 7.4 \cdot 10^{-10} \text{ m}^2/\text{s}$  at 296 K  
 754 have been reported in Ref. 49 for two different activated carbon monoliths saturated with  
 755 nitrogen.

## 756 2. *Measurements and predictions of sound absorption coefficient*

757 Figure 11 compares measured<sup>22,24</sup> and predicted normal incidence sound absorption coef-  
 758 ficient of a rigidly-backed 3-cm thick layer of granular activated carbon. The model, given by  
 759 Eqs. (29), (28), and those in tables I and II, accurately predicts the measured data. Its pre-  
 760 dictions for non-sorptive materials are also plotted to highlight the absorptive properties of  
 761 granular activated carbon. It is clearly observed that a sorptive material shows remarkably  
 762 higher sound absorption coefficient values at low frequencies in comparison with non-sorptive  
 763 ones, regardless of whether these feature multiple scales of heterogeneities. The absorption  
 764 of sound is primarily caused by the combined influence of viscosity and heat transfer effects  
 765 at the pore scale and the inter-scale mass and pressure diffusion processes, being the former  
 766 one influenced by sorption while the latter by both sorption and rarefaction.

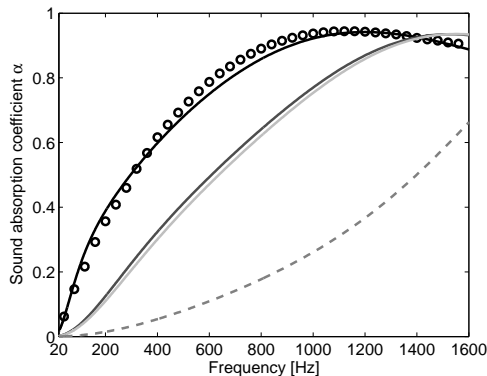


FIG. 11. Normal incidence sound absorption coefficient of a rigidly-backed 3-cm thick layer of granular activated carbon. Measurements<sup>22,24</sup> (circles) versus predictions for triple porosity sorptive material (black line) and triple (dark gray line), double (light gray line), and single (dashed gray line) porosity non-sorptive materials.

## 767 VI. CONCLUSIONS

768 This paper investigated sound propagation in multiscale rigid-frame porous materials  
769 accounting for the effects of viscosity and heat transfer at the pore scale, viscosity and  
770 heat transfer including rarefaction effects at the micropore scale, inter-scale (pore to/from  
771 micro-nanopore scales) pressure diffusion, inter-scale (micropore to/from nanopore scales)  
772 mass diffusion, and sorption in the nanopores. The two-scale asymptotic method of homog-  
773 enization for periodic media has been successively used to derive the macroscopic equations  
774 describing sound propagation through the material. These show that, at the leading order,  
775 the physical processes in the micro-nano porous domain do not modify the macroscopic fluid  
776 flow, provided that the advective mass flux pulsed from the micropores on the pore bound-  
777 aries is of one order smaller than the advective mass flux generated by the incident wave in  
778 the pores. As a consequence, the dynamic Darcy's law and the dynamic viscous permeability  
779 correspond to those of single porosity non-sorptive materials. Contrarily, the dynamic com-  
780 pressibility of the effective saturating fluid is significantly altered by the physical processes  
781 occurring at the micro- and nano scales.

782 We have demonstrated that sorption effects occurring in pores of nanometer size still  
783 significantly modify the macroscopic mass balance. This modification is accounted for by the  
784 dynamic compressibility of the effective saturating fluid which presents atypical properties  
785 that lead to a slower speed of sound and higher sound attenuation in the material. The  
786 strength of these macroscopic effects in the audible frequency range largely depends on  
787 the geometry and pressure and mass diffusion properties of the material. For example,  
788 relatively fast diffusing gas-solid systems whose pressure diffusion behavior is determined  
789 by a characteristic frequency that is in the order of that characterizing mass diffusion may  
790 be preferred over either slow mass diffusing systems or materials featuring slow pressure  
791 diffusion.

792 Contrarily to sorption effects, we have shown that rarefied gas flow in pores with size  
793 comparable to the molecular mean free path only intervene in the macroscopic acoustic  
794 behavior indirectly via a modification of the apparent pressure diffusivity.

795 The derived low-frequency asymptotics of the surface impedance, which is a quantity  
796 commonly measured in the field of acoustics of porous media, were used in conjunction with  
797 a characterization procedure to deduce the effective linearized sorption equilibrium constant

798 and effective diffusion coefficient of granular activated carbon. This provides empirical evi-  
799 dence supporting an alternative acoustic method for measuring sorption and mass diffusion  
800 properties of multiscale sorptive materials.

801 The developed theory was then validated experimentally by comparing its predictions  
802 with sound absorption measurements on a granular activated carbon sample showing good  
803 agreement.

804 In addition to the direct applications of the results presented in this paper to acoustics,  
805 one can consider applications to chemical engineering metrology and geophysics. Extensions  
806 of this work may include the study of sound propagation in multiscale sorptive porous  
807 materials saturated with fluid mixtures as well as the inclusion of the elasticity of the solid  
808 frame into the modeling.

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815 Reducing Materials and Structures (DENORMS)” which facilitated discussion on this topic.

## 816 **Appendix A**

817 This appendix provides the derivation of the upscaled model for sound propagation in  
818 the micro-nano porous domain. First we recall the model for double porosity sorptive ma-  
819 terials developed in Ref. 25. This is then extended to account for rarefaction effects at the  
820 micropore scale.

### 821 **1. Governing equations**

822 The governing equations for diffusion and sorption of a pure fluid in the nano porous  
823 domain are formulated first. In doing so, it is assumed that<sup>1,25</sup>: i) sorption occurs on the  
824 walls of the nanopores, ii) the adsorbed molecules (adsorbate) and the gas phase molecules

825 saturating the nanopores are in "dynamic instantaneous" equilibrium, and iii) the fluid  
 826 molecules are diffused through the nano porous domain via two diffusion mechanisms, i.e.  
 827 diffusion in the bulk of the nanopores and surface diffusion on the nanopore walls.

828 Since the characteristic sizes associated with the micro and nano scales are well separated,  
 829 i.e.  $l_n/l_m \ll 1$ , the nano porous domain is modeled as an equivalent continuum governed  
 830 by effective equations defined in  $\Omega_n$  and reflect the local physical processes. For simplicity,  
 831 it is assumed that the nano porous domain is isotropic. The constitutive flux equation  
 832 accounting for the two mentioned diffusion mechanisms is given by<sup>1,25</sup>:

$$J = -\phi_n(\varphi D_n \nabla c_n + (1 - \varphi) D_s \nabla c_s), \quad (\text{A.1})$$

833 where  $J$  is the molar mass flux,  $c_n$  and  $c_s$  are, respectively, the concentration of the gas (in  
 834 mole/volume of fluid) and adsorbed (in mole/volume of adsorbed phase) phases,  $D_n$  is the  
 835 nanopore diffusion coefficient, and  $D_s$  is the surface diffusion coefficient. The transport void  
 836 fraction  $\varphi = \Omega_v/\Omega_{nf}$  represents the fraction of the nano porous void space available for the  
 837 transport of free molecules. Similarly,  $(1 - \varphi) = \Omega_s/\Omega_{nf}$  represents the fraction of space  
 838 available for the transport of adsorbed molecules (see Figure 1).

839 Performing a mass balance in a volume element of the nano porous domain leads to the  
 840 following equation<sup>1,25</sup> :

$$j\omega\phi_n(\varphi c_n + (1 - \varphi)c_s) = \phi_n \nabla \cdot (\varphi D_n \nabla c_n + (1 - \varphi) D_s \nabla c_s) \quad \text{in } \Omega_n. \quad (\text{A.2})$$

841 As mentioned above, the adsorbed phase is assumed to be in "dynamic instantaneous"  
 842 equilibrium with the gas phase. This is valid when the local adsorption kinetics is much  
 843 faster than the diffusion processes. Such a situation is commonly found in nano porous media  
 844 and is justified by the fact that the average residence time of adsorption ranges from  $10^{-13}$   
 845 to  $10^{-9}$  s for physical adsorption<sup>1</sup>. The equilibrium relationship between the two phases is  
 846 then given by:

$$c_s = H c_n \quad \text{in } \Omega_n, \quad (\text{A.3})$$

847 where  $H$  is the linearized sorption equilibrium constant and is associated with the slope of  
 848 the local isotherm at a given equilibrium point, as shown for a Langmuir isotherm model  
 849 further below.

850 Replacing Eq. (A.3) into Eq. (A.2) and writing the concentration in terms of density,

851 i.e.  $\rho_n = c_n M$ , where  $M$  is the molar mass of the gas, lead to the Fickian equation<sup>1,5,25</sup>:

$$j\omega\rho_n H_e = D_e \nabla \cdot \nabla \rho_n \quad \text{in } \Omega_n, \quad (\text{A.4})$$

852 where the effective diffusion coefficient  $D_e$  and the effective linearized sorption equilibrium  
853 constant  $H_e$  are given by:

$$D_e = \phi_n(\varphi D_n + (1 - \varphi)D_s H), \quad (\text{A.5})$$

854

$$H_e = \phi_n(\varphi + (1 - \varphi)H). \quad (\text{A.6})$$

855 The equations of fluid motion in the micropores comprise the linearized equations of  
856 conservation of momentum, mass, and energy, and equation of state. These are respectively  
857 given by Eqs. (1), (2), (3), and (4) with the subscript  $p$  being replaced by  $m$  and are coupled  
858 with Eq. (A.4) via the following boundary conditions expressing the continuity of normal  
859 mass flux and pressure, and of negligible temperature variations and tangential mass flux  
860 on the micropore boundary  $\Gamma_m$ .

$$\rho_0 \mathbf{u}_m \cdot \mathbf{n}_m = -D_e \nabla \rho_n \cdot \mathbf{n}_m \quad \text{on } \Gamma_m, \quad (\text{A.7})$$

861

$$\rho_0(\mathbf{u}_m - (\mathbf{u}_m \cdot \mathbf{n}_m)\mathbf{n}_m) = \mathbf{0} \quad \text{on } \Gamma_m, \quad (\text{A.8})$$

862

$$p_n = p_m \quad \text{i.e.} \quad \rho_n = \frac{\rho_0}{F_0} p_m \quad \text{on } \Gamma_m, \quad (\text{A.9})$$

863

$$\tau_m = 0 \quad \text{on } \Gamma_m. \quad (\text{A.10})$$

864 Here the density and pressure in the effective nano porous domain are represented by  $\rho_n$   
865 and  $p_n$ . The oscillatory fluid velocity, pressure, and temperature in the micropores are  
866 respectively denoted as  $\mathbf{u}_m$ ,  $p_m$ , and  $\tau_m$ . The outward-pointing normal vector is  $\mathbf{n}_m$  (see  
867 Figure 1).

## 868 2. Homogenization procedure

869 The homogenization procedure described in Section II C is then applied. Now we consider  
870 the small parameter  $\varepsilon = l_m/l_p \ll 1$  and that the fast spatial variable  $y$  is associated with  
871 fluctuations at the micropore (local) scale while the slow spatial variable  $x$  with variations  
872 at the pore (larger) scale. The rescaled set of governing equations is formulated by assessing  
873 whether the variables fluctuate at the local or larger scale, as well as by considering the  
874 relative order of magnitude of the different terms in the governing equations. The oscillatory

875 fluid velocity  $\mathbf{u}_m$ , temperature  $\tau_m$ , and density  $\rho_n$  vary at the local scale, while the pressure  
876  $p_m$  at the larger scale<sup>25</sup>. The relative order of magnitude of the different terms in the  
877 equations of conservation of momentum, mass, and energy, and of state associated with  
878 the micropore fluid network are the same as those discussed in Section II C. On the other  
879 hand, the most general regime of mass diffusion/sorption occurs when both terms in Eq.  
880 (A.4) are of the same order of magnitude, i.e.  $O(D_e \rho_n / l_m^2) = O(\omega \rho_n H_e)$ . In addition, the  
881 long-wavelength condition imposes that, on the micropore boundary  $\Gamma_m$ , the diffusive mass  
882 flux is of one order smaller than the advective mass flux in the micropores<sup>25</sup>, i.e.

$$\mathcal{J} = \frac{|D_e \nabla \rho_n|}{|\rho_0 \mathbf{u}_m|} = O(\varepsilon). \quad (\text{A.11})$$

883 This can be written in terms of physical parameters as<sup>25</sup>:

$$\mathcal{J} = O\left(\frac{\rho_n}{u_m} \frac{D_e}{l_m \rho_0}\right) = O\left(\frac{\eta D_e l_p}{P_0 l_m^3}\right) = O(\varepsilon), \quad \text{i.e.} \quad \frac{\eta D_e}{l_m^2 P_0} = O(\varepsilon^2), \quad (\text{A.12})$$

884 and indicates that the effective description of sound propagation in the micro-nano porous  
885 domain to be derived is valid when the combination of physical parameters  $P_0$ ,  $\eta$ ,  $D_e$ , and  
886  $l_m$  satisfies the estimation Eq. (A.12).

887 The rescaled set of equations is then given by Eqs. (12), (13), and (14) with the subscript  
888  $p$  being replaced by  $m$ , Eq. (A.13), and boundary conditions (A.14)-(A.17).

$$\varepsilon^2 D_e \nabla \cdot \nabla \rho_n = j \omega \rho_n H_e \quad \text{in} \quad \Omega_n. \quad (\text{A.13})$$

889

$$\rho_0 \mathbf{u}_m \cdot \mathbf{n}_m = -\varepsilon^2 D_e \nabla \rho_n \cdot \mathbf{n}_m \quad \text{on} \quad \Gamma_m, \quad (\text{A.14})$$

890

$$\rho_0 (\mathbf{u}_m - (\mathbf{u}_m \cdot \mathbf{n}_m) \mathbf{n}_m) = \mathbf{0} \quad \text{on} \quad \Gamma_m, \quad (\text{A.15})$$

891

$$\rho_n = \frac{\rho_0}{P_0} p_m \quad \text{on} \quad \Gamma_m, \quad (\text{A.16})$$

892

$$\tau_m = 0 \quad \text{on} \quad \Gamma_m. \quad (\text{A.17})$$

893 Note that the  $\varepsilon^2$ -rescaling in the boundary condition (A.14) results from (i) the physical  
894 estimate (A.11) stating that the diffusive flux is of one order smaller than the advective one,  
895 and (ii) the fact that  $\rho_n$  varies at the local scale.

896 The physical variables are then looked for in the form of asymptotic expansions in powers  
897 of the small parameter  $\varepsilon = l_m / l_p$  as  $Q(x, y) = \sum_{i=0}^{\infty} \varepsilon^i Q^{(i)}(x, y)$  where  $Q = p_m, \mathbf{u}_m, \tau_m, \rho_n$ .  
898 These are then substituted into the rescaled set of equations and the terms of the same  
899 order are identified. At  $\varepsilon^{-1}$  it follows from the equation of conservation of momentum that



900  $\nabla_y p_m^{(0)} = 0$ , which means that the pressure varies at the larger scale, i.e.  $p_m^{(0)} = p_m^{(0)}(x)$ .

901 Further identification provides the following leading-order cell problems.

902 Fluid flow:

$$903 \quad \eta \nabla_y^2 \mathbf{u}_m^{(0)} - \nabla_y p_m^{(1)} = j\omega \rho_0 \mathbf{u}_m^{(0)} + \nabla_x p_m^{(0)} \quad \text{in } \Omega_{mf}, \quad (\text{A.18})$$

$$904 \quad \nabla_y \cdot \mathbf{u}_m^{(0)} = 0 \quad \text{in } \Omega_{mf}, \quad (\text{A.19})$$

$$\mathbf{u}_m^{(0)} = \mathbf{0} \quad \text{on } \Gamma_m. \quad (\text{A.20})$$

905 Heat conduction:

$$906 \quad \kappa \nabla_y \cdot \nabla_y \tau_m^{(0)} = j\omega C_p \rho_0 \tau_m^{(0)} - j\omega p_m^{(0)} \quad \text{in } \Omega_{mf}, \quad (\text{A.21})$$

$$\tau_m^{(0)} = 0 \quad \text{on } \Gamma_m. \quad (\text{A.22})$$

907 Mass diffusion:

$$908 \quad D_e \nabla_y \cdot \nabla_y \rho_n^{(0)} = j\omega \rho_n^{(0)} H_e \quad \text{in } \Omega_n, \quad (\text{A.23})$$

$$\rho_n^{(0)} = \frac{\rho_0}{P_0} p_m^{(0)} \quad \text{on } \Gamma_m. \quad (\text{A.24})$$

909 The solutions of the fluid flow (i.e. Eqs. (A.18)-(A.20)) and heat conduction (i.e. Eqs.  
910 (A.21)-(A.22)) problems are given by<sup>29,30,32</sup>:

$$911 \quad \mathbf{u}_m^{(0)} = -\frac{\bar{\mathbf{k}}_m(y, \omega)}{\eta} \cdot \nabla_x p_m^{(0)}, \quad (\text{A.25})$$

$$912 \quad p_m^{(1)} = -\bar{\pi}_m(y, \omega) \cdot \nabla_x p_m^{(0)} + \bar{p}_m^{(1)}(x), \quad (\text{A.26})$$

$$\tau_m^{(0)} = \frac{\bar{\theta}_m(y, \omega)}{\kappa} j\omega p_m^{(0)}, \quad (\text{A.27})$$

913 where  $\bar{\mathbf{k}}_m(y, \omega)$  and  $\bar{\theta}_m(y, \omega)$  represent the  $\Omega_m$  - periodic local fields of velocity and tem-  
914 perature respectively. The pressure field has been expressed in terms of its zero mean part  
915  $\bar{\pi}_m(y, \omega)$  and an integration constant  $\bar{p}_m^{(1)}(x)$ .

916 The solution of the mass diffusion problem (i.e. Eqs.(A.23) and (A.24)) is given by<sup>6,25</sup>:

$$\frac{\rho_n^{(0)}}{\rho_0} = \frac{p_m^{(0)}}{P_0} \left( 1 - \frac{j\omega \bar{g}(y, \omega)}{\mathcal{D}_{app}} \right), \quad (\text{A.28})$$

917 where  $\bar{g}(y, \omega)$  represents the  $\Omega_m$  - periodic local diffusive density field and the apparent  
918 diffusivity  $\mathcal{D}_{app}$  is defined as:

$$\mathcal{D}_{app} = \frac{D_e}{H_e} = \frac{\varphi D_n + (1 - \varphi) D_s H}{\varphi + (1 - \varphi) H}. \quad (\text{A.29})$$

### 919 3. Effective model for sound propagation in the micro-nano porous domain

920 The leading-order mass balance equation, i.e. Eq. (13) with the subscript  $p$  being replaced  
 921 by  $m$ , integrated over the micropore space  $\Omega_{mf}$  and divided by the volume  $\Omega_m$  is given by:

$$j\omega \left\langle \frac{p_m^{(0)}}{P_0} - \frac{\tau_m^{(0)}}{\tau_0} \right\rangle_m + \langle \nabla_x \cdot \mathbf{u}_m^{(0)} \rangle_m + \langle \nabla_y \cdot \mathbf{u}_m^{(1)} \rangle_m = 0, \quad (\text{A.30})$$

922 where the averaging operator is defined as:

$$\langle \cdot \rangle_m = \frac{1}{\Omega_m} \int_{\Omega_{mf}} \cdot d\Omega. \quad (\text{A.31})$$

923 The term  $\langle \nabla_y \cdot \mathbf{u}_m^{(1)} \rangle_m$  in Eq. (A.30) is calculated by making successive use of the di-  
 924 vergence theorem, noting that the surface integrals on opposite boundaries of the cell can-  
 925 cel out due to periodicity, and using the boundary condition (A.14) identified at  $\varepsilon^1$  (i.e.  
 926  $\rho_0 \mathbf{u}_m^{(1)} \cdot \mathbf{n}_m = -D_e \nabla_y \rho_n^{(0)} \cdot \mathbf{n}_m$  on  $\Gamma_m$ ) and Eqs. (A.23) and (A.28). Its expression is:

$$\langle \nabla_y \cdot \mathbf{u}_m^{(1)} \rangle_m = j\omega p_m^{(0)} \frac{H_e}{P_0} \frac{1}{\Omega_m} \int_{\Omega_n} \left( 1 - \frac{j\omega \bar{g}(y, \omega)}{\mathcal{D}_{app}} \right) d\Omega. \quad (\text{A.32})$$

927 The effective model for sound propagation in the micro-nano porous domain is then  
 928 obtained by i) replacing Eq. (A.32) into Eq. (A.30), ii) substituting  $\tau_m^{(0)}$  (i.e. Eq. (A.27))  
 929 and  $\rho_n^{(0)}$  (i.e. Eq. A.28) in Eq. (A.30), iii) using the thermodynamic identity  $P_0/\tau_0 =$   
 930  $\rho_0 C_p (\gamma - 1)/\gamma$  (where  $\gamma$  is the specific heat ratio), and iv) considering that the averaged  
 931 velocity is determined from Eq. (A.25). The mass balance equation and constitutive fluid  
 932 flow law are given by:

$$\nabla \cdot \mathbf{U}_m + j\omega p_m C_{mn} = 0, \quad (\text{A.33})$$

933

$$\mathbf{U}_m = -\frac{\mathbf{k}_m(\omega)}{\eta} \cdot \nabla p_m. \quad (\text{A.34})$$

934 Here the Darcy's velocity is given by  $\mathbf{U}_m = \langle \mathbf{u}_m^{(0)} \rangle_m$  and since the derivatives are taken  
 935 with respect to the larger-scale spatial variable and the pressure and Darcy's velocity are  
 936 leading order terms, we have dropped the superscript  $(0)$  and the index  $x$  here and in the  
 937 main text. The dynamic viscous permeability associated with the micropore fluid network  
 938 is calculated as  $\mathbf{k}_m(\omega) = \langle \bar{\mathbf{k}}_m(y, \omega) \rangle_m$ . The effective dynamic compressibility  $C_{mn}(\omega)$  is  
 939 given by Eq. (25), i.e.  $C_{mn} = C_m + (1 - \phi_m) C_n \mathcal{F}_{mn}$ , and corresponds to the sum of the  
 940 classical effective dynamic compressibility accounting for heat transfer in the micropores  
 941  $C_m(\omega)$ , and an additional effective dynamic compressibility  $C_n$  that accounts for sorption in

942 the nanopores and is modified by the inter-scale mass diffusion. The latter is also affected  
 943 by sorption and is accounted for by  $\mathcal{F}_{mn}(\omega)$ . The compressibility  $C_m$  is calculated using  
 944 Eq. (23) with the subscript  $p$  being replaced by  $m$  and the associated dynamic thermal  
 945 permeability is calculated as  $\Theta_m(\omega) = \langle \bar{\theta}_m(y, \omega) \rangle_m$ . The compressibility  $C_n$  is given by Eq.  
 946 (26) and  $\mathcal{F}_{mn}$  is related to the inter-scale (micro-nano) mass diffusion function  $G(\omega)$  via Eq.  
 947 (27). The latter is calculated as<sup>25</sup>:

$$G(\omega) = \frac{1}{\Omega_m} \int_{\Omega_n} \bar{g}(y, \omega) d\Omega. \quad (\text{A.35})$$

948 In summary, the effective model for sound propagation in the micro-nano porous domain,  
 949 given by Eqs. (A.33) and (A.34), allows concluding that the constitutive fluid flow law of  
 950 the micropore fluid network and the dynamic viscous permeability are not modified by the  
 951 inter-scale mass diffusion and sorption in the nanopores. Conversely, the effective dynamic  
 952 compressibility becomes significantly modified by inter-scale mass diffusion and sorption.  
 953 This modification comes from the appearance of a source term in the mass balance equation  
 954 (i.e. the third term in Eq. (A.30)) that accounts for the contribution of both processes.

955 As shown in Section III, the low-frequency behavior of the effective dynamic compress-  
 956 ibility is determined by  $H_e$ , which depends on  $H$ . A linearized dynamic sorption model  
 957 derived in Ref. 25 allows linking  $H$  with the parameters of the classical Langmuir kinetics  
 958 model<sup>34</sup>. Its use leads to replace Eq. (A.3) by  $c_s = \mathcal{H}(\omega)c_n$ , where the linearized sorption  
 959 “dynamic equilibrium” constant is given by:

$$\mathcal{H}(\omega) = \frac{\rho_N}{\rho_0} \frac{bP_0}{(1 + bP_0)^2} \frac{1}{(1 + \frac{j\omega}{\omega_a})}. \quad (\text{A.36})$$

960 Here  $\omega_a = k_a P_0 + k_d$  is the sorption characteristic frequency,  $k_a$  is the adsorption constant  
 961 (in 1/Pa/s),  $k_d$  is the desorption constant (in 1/s), and  $b = k_a/k_d$  is the Langmuir constant  
 962 (in 1/Pa), and  $\rho_N$  is the maximum density increment due to sorption. Since the average  
 963 residence time of adsorption, i.e.  $\tau_a = 1/\omega_a$ , ranges from  $10^{-13}$  to  $10^{-9}$  s for physical  
 964 adsorption<sup>1</sup>, the sorption characteristic frequency takes very high values and  $\mathcal{H}(\omega)$  can be  
 965 approximated by<sup>25</sup>:

$$H = \mathcal{H}(\omega \ll \omega_a) = \frac{\rho_N}{\rho_0} \frac{bP_0}{(1 + bP_0)^2}. \quad (\text{A.37})$$

#### 966 4. Rarefaction effects on sound propagation in the micro-nano porous domain

967 For micropores with sizes comparable to the molecular mean free path  $\ell = \frac{\eta}{P_0} \sqrt{\pi R_g \tau_0 / 2M}$   
968 with  $R_g$  being the gas constant, effects related to the molecular nature of the gas start  
969 becoming considerable<sup>37,40,50–53</sup>. The ratio between  $\ell$  and the micropore characteristic size,  
970 known as the Knudsen number  $\text{Kn} = \ell/l_m$ , measures the degree of rarefaction. Its value is  
971 used for assessing the validity of the continuum hypothesis<sup>50</sup>. For  $\text{Kn} < 0.01$ , this hypothesis  
972 remains valid and the set of equations discussed in Section A 1 holds. For  $0.01 < \text{Kn} < 0.01$   
973 (commonly referred to as the slip-flow regime), the continuum description is valid everywhere  
974 in the micropore fluid network except in a thin Knudsen layer close to the micropore walls.  
975 In order to account for this effect, the continuum description is modified by allowing a degree  
976 of tangential-velocity slip<sup>51</sup>. Hence the boundary conditions (A.7) and (A.8) are replaced  
977 by Eq. (A.38). Similarly, molecular effects influence the thermal behavior of materials with  
978  $l_m = O(\ell)$ . For  $0.01 < \text{Kn} < 0.1$ , the continuum description is modified to account for the  
979 temperature-jump on the micropore boundaries. This is achieved by replacing Eq.(A.10)  
980 by the so-called temperature-jump boundary condition Eq. (A.39), which states that the  
981 temperature is proportional to the normal component of the temperature gradient<sup>37,40,52</sup>. In  
982 these equations,  $\mathbf{t}_m$  is the tangential vector collinear with the velocity slip and the velocity  
983 slip and temperature jump coefficients are respectively denoted by  $c_v$  and  $c_t$  and are assumed  
984 equal to one, i.e. the molecules are reflected diffusively<sup>50</sup>.

$$\mathbf{u}_m = \left( -\frac{D_e}{\rho_0} \nabla \rho_n \cdot \mathbf{n}_m \right) \mathbf{n}_m - c_v \ell (\mathbf{t}_m \cdot \nabla \mathbf{u}_m \cdot \mathbf{n}_m) \mathbf{t}_m \quad \text{on } \Gamma_m, \quad (\text{A.38})$$

985

$$\tau_m = c_t \frac{2\gamma}{\gamma + 1} \text{Pr} \ell \nabla \tau_m \cdot \mathbf{n}_m \quad \text{on } \Gamma_m. \quad (\text{A.39})$$

986 Reminding that the velocity  $\mathbf{u}_m$ , temperature  $\tau_m$ , and density  $\rho_n$  vary at the local scale, and  
987 using Eq. (A.11); these boundary conditions rewritten in rescaled form are given by:

$$\mathbf{u}_m = \left( -\varepsilon^2 \frac{D_e}{\rho_0} \nabla \rho_n \cdot \mathbf{n}_m \right) \mathbf{n}_m - \varepsilon \ell (\mathbf{t}_m \cdot \nabla \mathbf{u}_m \cdot \mathbf{n}_m) \mathbf{t}_m \quad \text{on } \Gamma_m, \quad (\text{A.40})$$

988

$$\tau_m = \varepsilon \frac{2\gamma}{\gamma + 1} \text{Pr} \ell \nabla \tau_m \cdot \mathbf{n}_m \quad \text{on } \Gamma_m. \quad (\text{A.41})$$

989 The application of the homogenization procedure leads to the fluid flow and heat conduction  
990 leading-order cell problems Eqs. (A.18)-(A.19) and (A.21) with the boundary conditions  
991 (A.20) and (A.22) being respectively replaced by :

$$\mathbf{u}_m^0 = -\ell (\mathbf{t}_m^{(0)} \cdot \nabla_y \mathbf{u}_m^{(0)} \cdot \mathbf{n}_m) \mathbf{t}_m^{(0)} \quad \text{on } \Gamma_m, \quad (\text{A.42})$$

$$\tau_m^{(0)} = \frac{2\gamma}{\gamma + 1} \text{Pr} \ell \nabla_y \tau_m^{(0)} \cdot \mathbf{n}_m \quad \text{on} \quad \Gamma_m. \quad (\text{A.43})$$

992 These resulting leading-order cell problems have been solved in Refs. 22, 24, and 53. Their  
 993 solutions, which are given by Eqs. (A.44) and (A.45), replace Eqs. (A.25) and (A.27).

$$\mathbf{u}_m^{(0)} = -\frac{\bar{\mathbf{k}}_m(y, \omega, \text{Kn})}{\eta} \cdot \nabla_x p_m^{(0)}, \quad (\text{A.44})$$

994

$$\tau_m^{(0)} = \frac{\bar{\theta}_m(y, \omega, \text{Kn})}{\kappa} j\omega p_m^{(0)}. \quad (\text{A.45})$$

995 Further algebra leads to the effective model for sound propagation in the micro-nano porous  
 996 domain given by Eqs. (20)-(21). However, its effective parameters are modified by rarefac-  
 997 tion effects. Specifically, the dynamic viscous and thermal permeabilities associated with the  
 998 micropore fluid network are calculated as  $\mathbf{k}_m(\omega, \text{Kn}) = \langle \bar{\mathbf{k}}_m(y, \omega, \text{Kn}) \rangle_m$  and  $\Theta_m(\omega, \text{Kn}) =$   
 999  $\langle \bar{\theta}_m(y, \omega, \text{Kn}) \rangle_m$ . Consequently, the effective dynamic compressibilities  $\mathcal{C}_m$  and  $\mathcal{C}_{mn}$  are also  
 1000 affected by rarefaction effects.

1001 It then follows that sound propagation in multiscale sorptive materials is affected by  
 1002 rarefaction effects via the dependence of the effective dynamic compressibility  $\mathbf{C}$  on both  
 1003  $\mathcal{C}_{mn}$  and  $\mathcal{F}_{pmn}$ . In particular, the modification by rarefaction effects of the latter comes  
 1004 from their influence on the pressure field  $p_m^{(0)}$  in Eq. (B.6), which is determined by the  
 1005 apparent pressure diffusivity (i.e. Eq. (B.7)) that becomes Knudsen number-dependent, i.e.  
 1006  $\mathcal{B}_{app}(\omega, \text{Kn}) = \mathcal{K}_m(\omega, \text{Kn})/\eta\mathcal{C}_{mn}(\omega, \text{Kn})$ .

## 1007 Appendix B

1008 This appendix provides the mathematical details of the derivation of the macroscopic  
 1009 equations (20) and (21) that describe sound propagation in multiscale sorptive materials.

1010 Replacing the variables written as asymptotic expansions in powers of the small parameter  
 1011  $\epsilon$ , i.e.  $Q(x, y) = \sum_{i=0}^{\infty} \epsilon^i Q^{(i)}(x, y)$  where  $Q = p_p, \mathbf{u}_p, \tau_p, \rho_p, p_m, \mathbf{U}_m$ , into Eqs. (12)-(19) and  
 1012 identifying the terms of the same order lead to the following results. At  $\epsilon^{-1}$ , it follows  
 1013 from the equation of conservation of momentum that  $\nabla_y p_p^{(0)} = 0$ , which implies that the  
 1014 pore pressure is a macroscopic variable, i.e.  $p_p^{(0)} = p_p^{(0)}(x)$ . Further identification leads to  
 1015 the oscillatory Stokes and heat conduction problems in the pore fluid network. These are  
 1016 respectively given by Eqs. (A.18)-(A.20) and (A.21)-(A.22) with the subscript  $m$  being

1017 replaced by  $p$ . Their solutions are given by<sup>29,30,32</sup>:

$$\mathbf{u}_p^{(0)} = -\frac{\bar{\mathbf{k}}_p(y, \omega)}{\eta} \cdot \nabla_x p_p^{(0)}, \quad (\text{B.1})$$

1018

$$p_p^{(1)} = -\bar{\pi}_p(y, \omega) \cdot \nabla_x p_p^{(0)} + \bar{p}_p^{(1)}(x), \quad (\text{B.2})$$

1019

$$\tau_p^{(0)} = \frac{\bar{\theta}_p(y, \omega)}{\kappa} j\omega p_p^{(0)}. \quad (\text{B.3})$$

1020 where  $\bar{\mathbf{k}}_p(y, \omega)$  and  $\bar{\theta}_p(y, \omega)$  represent the  $\Omega_p$ -periodic local fields of velocity and temperature  
 1021 respectively. The pressure field has been expressed in terms of its zero mean part  $\bar{\pi}_p(y, \omega)$   
 1022 and an integration constant  $\bar{p}_p^{(1)}(x)$ .

1023 In the micro-nano porous domain, which is assumed isotropic for simplicity, the local  
 1024 pressure field is imposed by the pore pressure on  $\Gamma_p$  and is governed by the following set of  
 1025 equations:

$$\nabla_y \cdot \left( -\frac{\mathcal{K}_m}{\eta} \nabla_y p_m^{(0)} \right) + j\omega p_m^{(0)} \mathcal{C}_{mn} = 0 \quad \text{in } \Omega_{mn}, \quad (\text{B.4})$$

1026

$$p_m^{(0)} = p_p^{(0)} \quad \text{on } \Gamma_p. \quad (\text{B.5})$$

1027 This problem is formally identical to that of pressure diffusion in double porosity materials  
 1028 with highly-contrasted permeabilities<sup>33,36</sup>. Therefore, its solution is given by:

$$p_m^{(0)} = p_p^{(0)} \left( 1 - \frac{j\omega \bar{b}(y, \omega)}{\mathcal{B}_{app}} \right), \quad (\text{B.6})$$

1029 where  $\bar{b}(y, \omega)$  represents the  $\Omega_p$  - periodic local diffusive pressure field and the apparent  
 1030 pressure diffusivity  $\mathcal{B}_{app}$  is defined by:

$$\mathcal{B}_{app} = \frac{\mathcal{K}_m}{\eta \mathcal{C}_{mn}}. \quad (\text{B.7})$$

1031 The integration of the leading-order mass balance equation over the pore volume leads  
 1032 to:

$$j\omega \left\langle \frac{p_p^{(0)}}{P_0} - \frac{\tau_p^{(0)}}{\tau_0} \right\rangle + \nabla_x \cdot \langle \mathbf{u}_p^{(0)} \rangle + \langle \nabla_y \cdot \mathbf{u}_p^{(1)} \rangle = 0, \quad (\text{B.8})$$

1033 where the averaging operator is defined as:

$$\langle \cdot \rangle = \frac{1}{\Omega_p} \int_{\Omega_{pf}} \cdot d\Omega. \quad (\text{B.9})$$

1034 The last term on the left-hand side of Eq. (B.8) is calculated by i) using the divergence  
 1035 theorem, ii) taking into account that the surface integrals on opposite boundaries of the cell

1036 cancel out due to periodicity and using Eq. (17) identified at order  $\epsilon$ , iii) transforming the  
 1037 resulting surface integral into a volume integral using the divergence theorem, iv) integrating  
 1038 over the micro-nano porous domain the Eq. (15) identified at the dominant order, and v)  
 1039 replacing Eq. (B.6). The final result is:

$$\langle \nabla_y \cdot \mathbf{u}_p^{(1)} \rangle = j\omega p_p^{(0)}(1 - \phi_p) \mathbf{C}_{mn}(\omega) \left( 1 - \frac{j\omega B(\omega)}{(1 - \phi_p) \mathcal{B}_{app}} \right), \quad (\text{B.10})$$

1040 where the inter-scale pressure diffusion function  $B(\omega)$  is calculated as:

$$B(\omega) = \frac{1}{\Omega_p} \int_{\Omega_{mn}} \bar{b}(y, \omega) d\Omega. \quad (\text{B.11})$$

1041 The macroscopic description of sound propagation in multiscale sorptive materials, i.e.  
 1042 Eqs. (20) and (21), is then obtained by substituting Eqs. (B.10) and (B.3) into Eq. (B.8),  
 1043 using the thermodynamic identity  $P_0/\tau_0 = \rho_0 C_p(\gamma - 1)/\gamma$ , and considering that the averaged  
 1044 fluid velocity is determined from Eq. (B.1), with the dynamic viscous permeability tensor  
 1045 being calculated as  $\mathbf{k}_p(\omega) = \langle \bar{\mathbf{k}}_p(y, \omega) \rangle$ . The dynamic thermal permeability associated to  
 1046 the pore scale is calculated as  $\Theta_p(\omega) = \langle \bar{\theta}_p(y, \omega) \rangle$  (see Eq. (B.3)).

## 1047 Appendix C

1048 This appendix presents the derivation of the asymptotic values of effective dynamic com-  
 1049 pressibility  $\mathbf{C}(\omega)$  given by Eqs. (30) and (31).

1050 The behavior of  $\mathbf{C}_i(\omega)$  (with  $i = p, m$ ) is characterized by that of  $\Theta_i(\omega)$ , which is in  
 1051 turn determined by  $\omega_{ti}$ . The thermal permeability varies<sup>32</sup> from  $\Theta_i(\omega \ll \omega_{ti}) = \Theta_{i0}$  to  
 1052  $\Theta_i(\omega \gg \omega_{ti}) = -j\phi_i \delta_t^2$ , where  $\delta_t = \sqrt{\kappa/\rho_0 C_p \omega}$  is the thermal boundary layer thickness.  
 1053 Therefore, the effective dynamic compressibility  $\mathbf{C}_i(\omega)$  varies as<sup>32</sup>:

$$\mathbf{C}_i = \begin{cases} \frac{\phi_i}{P_0} \left( 1 - \frac{\gamma-1}{\gamma} \frac{j\omega}{\omega_{ti}} \right) & \text{for } \omega \ll \omega_{ti} \\ \frac{\phi_i}{\gamma P_0} & \text{for } \omega \gg \omega_{ti}. \end{cases} \quad (\text{C.1})$$

1054 The behavior of  $\mathcal{F}_{mn}(\omega)$  is determined by that of  $G(\omega)$ . The latter is characterized by  
 1055 the mass diffusion characteristic frequency  $\omega_d$  and varies<sup>25</sup> from  $G(\omega \ll \omega_d) = G_0$  to  
 1056  $G(\omega \gg \omega_d) = -j(1 - \phi_m) \delta_d^2$ , where  $\delta_d = \sqrt{\mathcal{D}_{app}/\omega}$  is the mass diffusion boundary layer  
 1057 thickness. It then follows that  $\mathcal{F}_{mn}(\omega \ll \omega_d) = 1 - j\omega/\omega_d$  and  $\mathcal{F}_{mn}(\omega \gg \omega_d) = 0$ .  
 1058 Therefore, the effective dynamic compressibility  $\mathbf{C}_{mn}$  takes the following asymptotic values:

$$\mathbf{C}_{mn} = \begin{cases} \frac{1}{P_0} \left( \phi_m + (1 - \phi_m) H_e - j\omega \left( \frac{\gamma-1}{\gamma} \frac{\phi_m}{\omega_{tm}} + H_e \frac{(1-\phi_m)}{\omega_d} \right) \right) & \text{for } \omega \ll \omega_d \\ \frac{\phi_m}{\gamma P_0} & \text{for } \omega \gg \omega_{tm}. \end{cases} \quad (\text{C.2})$$

1059 From Eq. (C.2), it is deduced that the static compressibility of the effective fluid satu-  
 1060 rating the micro-nano porous domain is given by<sup>25</sup> :

$$C_{mn0} = C_{mn}(\omega = 0) = \frac{\phi_m + (1 - \phi_m)H_e}{P_0}. \quad (\text{C.3})$$

1061 The behavior of the function  $\mathcal{F}_{pmn}(\omega)$  is determined by  $B(\omega)$  and  $\mathcal{B}_{app}(\omega)$ . Focusing on  
 1062 isothermal sound propagation in the micropores (i.e.  $\omega \ll \omega_{tm}$ ) and quasi-static inter-scale  
 1063 mass diffusion (i.e.  $\omega \ll \omega_d$ ), the apparent pressure diffusivity is estimated by  $\mathcal{B}_{app} \approx$   
 1064  $\mathcal{B}_{app0} = \mathcal{K}_{m0}/\eta C_{mn0}$ . Then, the pressure diffusion function varies from  $B(\omega \ll \omega_b) = B_0$   
 1065 to  $B(\omega \gg \omega_b) = -j(1 - \phi_p)\delta_{b0}^2$ , where  $\delta_{b0} = \sqrt{\mathcal{B}_{app0}/\omega}$  is an estimation of the pressure  
 1066 diffusion boundary layer thickness. It then follows that  $\mathcal{F}_{pmn}(\omega \ll \omega_b) = 1 - j\omega/\omega_b$  and  
 1067  $\mathcal{F}_{pmn}(\omega \gg \omega_b) = 0$ . Combining this result with Eqs. (22), (C.1), and (C.2), and only  
 1068 retaining linear terms in frequency, one obtains Eqs. (30) and (31).

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